

陈宜峰教授在有机不对称催化方面的研究

汇报人：张丹 2025年1月10日

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- 4. Catalytic Intermolecular Cross-Coupling of Olefins**

1. Author introduction

- **Education and research experience**



Yi-Feng Chen, Ph. D.

2003-2007 **B.S.** Soochow University. (Advisor: Jianping Zhou)

2007-2012 **Ph.D.** SIOC. (Advisor: Yuanhong Liu)

2012-2013 **Research Assistant.** SIOC. (Advisor: Yuanhong Liu)

2013-2014 **Postdoctoral Fellow.** MIT. (Advisor: Stephen L. Buchwald)

2014 – 2017 **Postdoctoral Fellow.** Yale University. (Advisor: Timothy R. Newhouse)

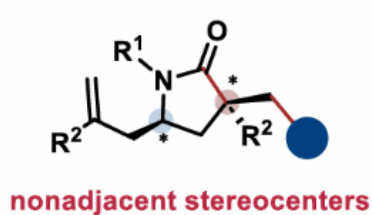
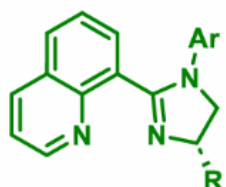
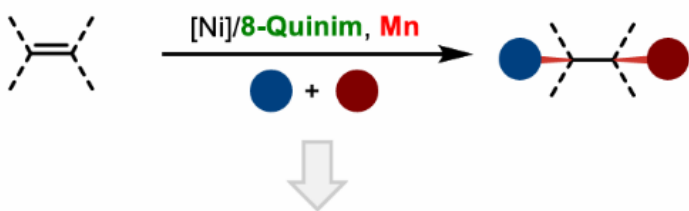
Present **Professor.** ECUST.

**Research
field**

**I. 廉价金属催化的羰化反应研究；
II. 通过手性配体的从头设计合成来发展不对称催化的新方法。**

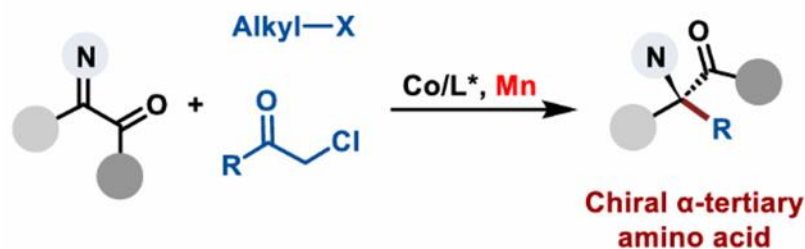
2. Catalytic Asymmetric Reductive Addition (CARA) Reaction

C=C bonds

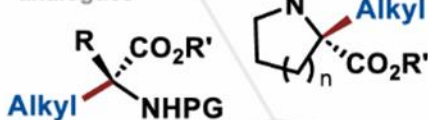


- rigid ring skeleton
- scalable synthesis
- facile modification

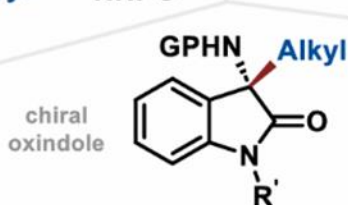
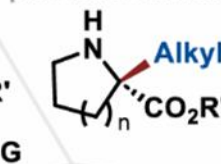
C=N bonds



amino acid
analogues

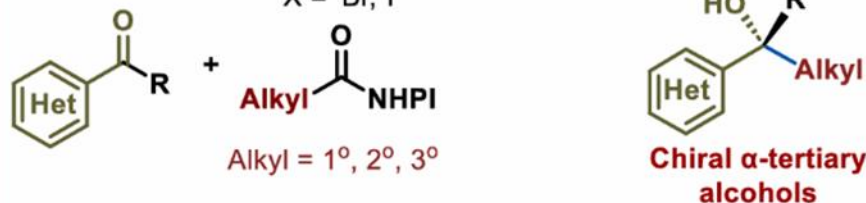
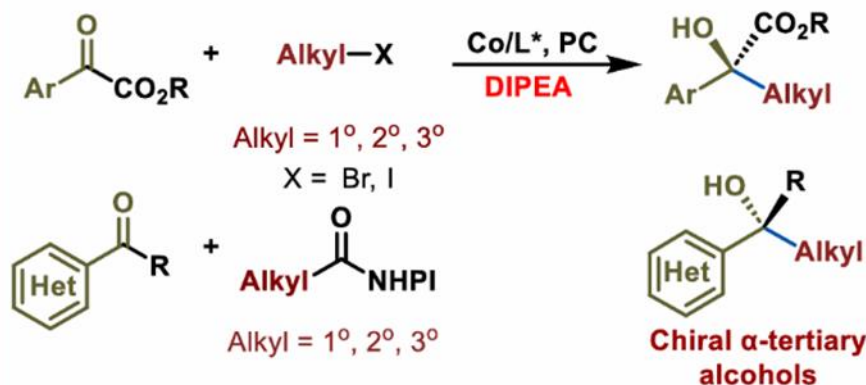


proline derivatives

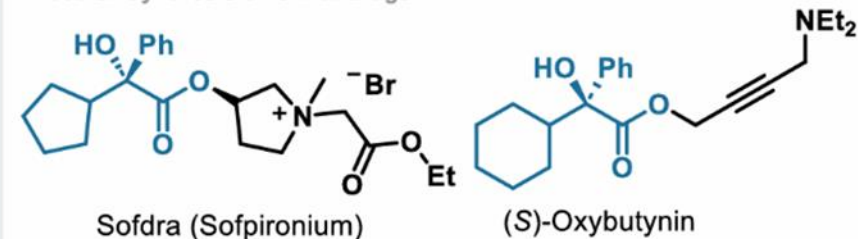


β -Tertiary Amino
Acid Analogues

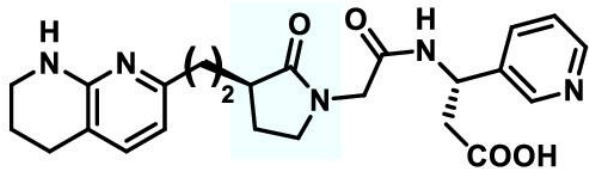
C=O bonds



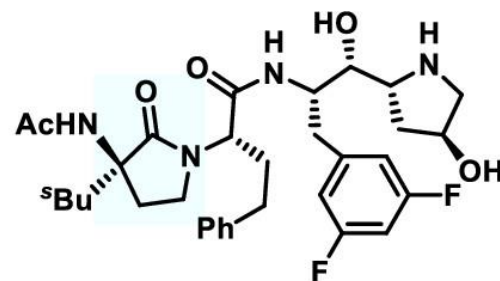
Modular Synthesis of chiral drugs



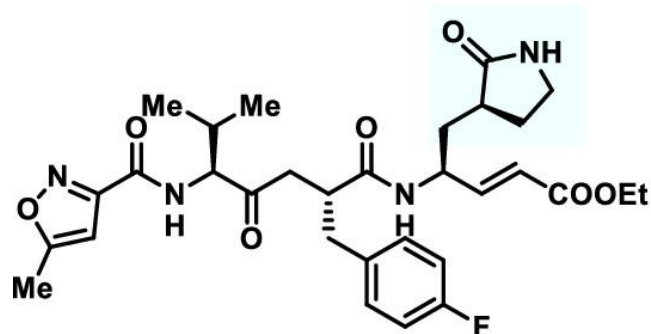
2.1 C=C bonds



$\alpha_v\beta_3$ integrin antagonists



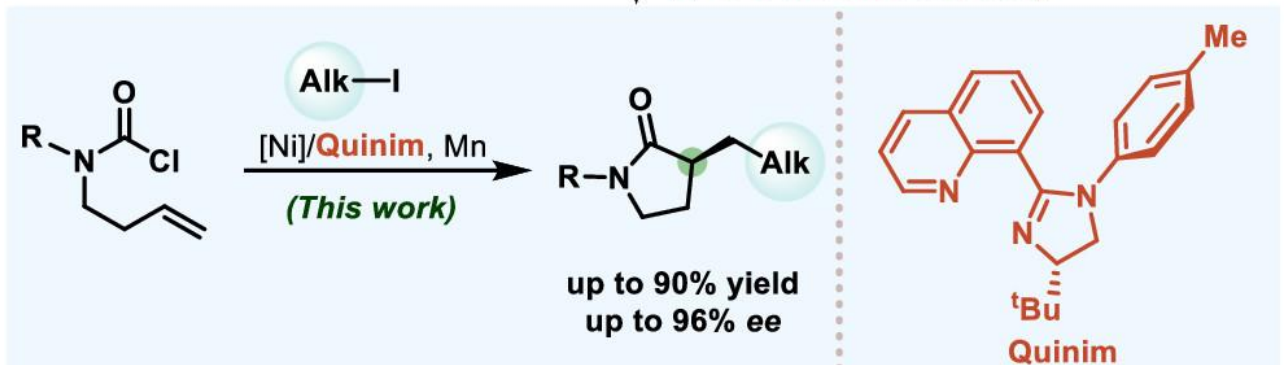
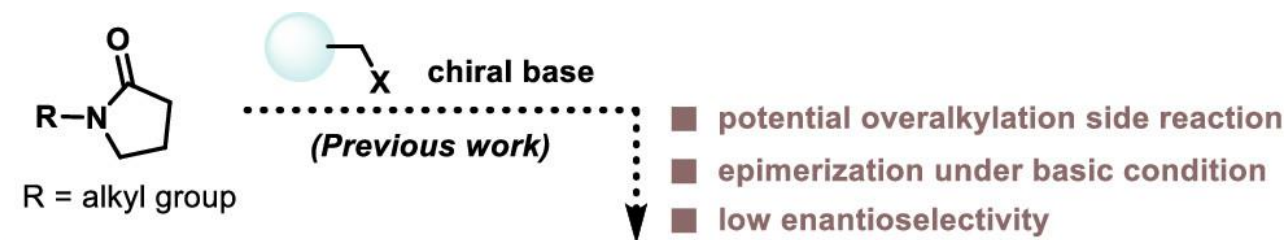
BACE-1 inhibitor



Rupintrivir



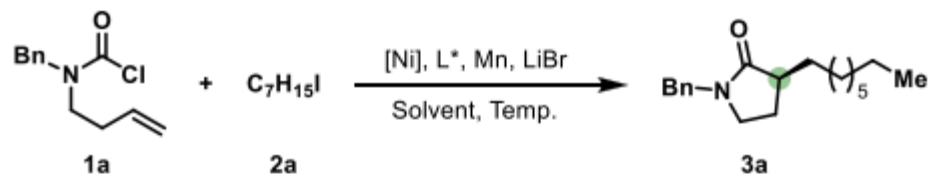
(-)-Salinosporamide A



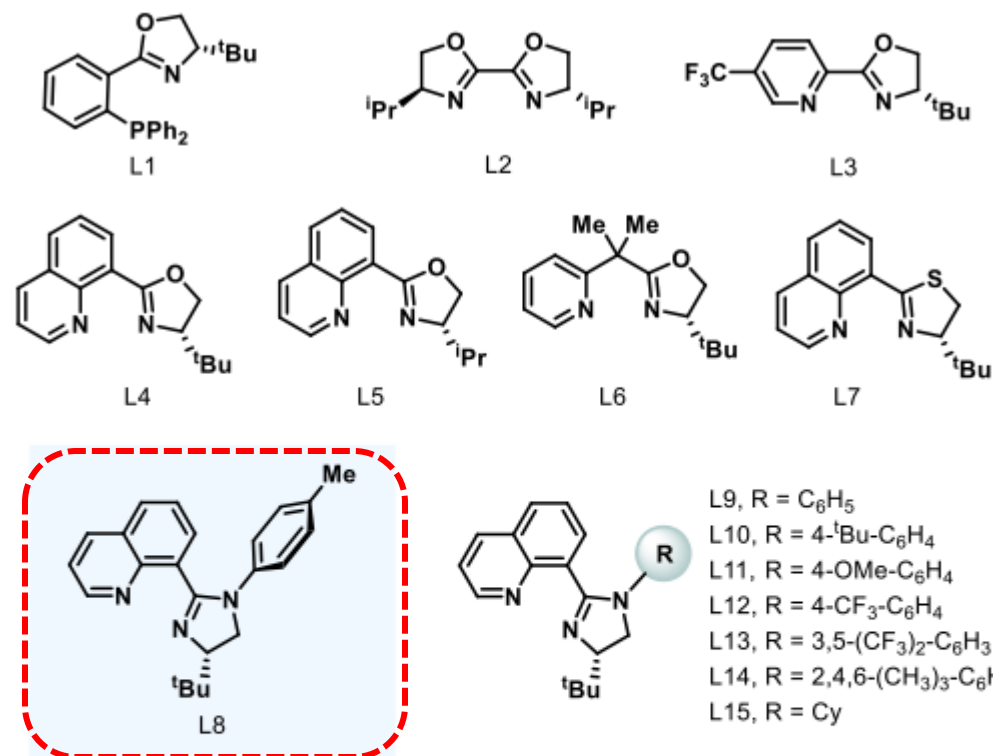
- newly synthesized, bidentate nitrogen ligand: Quinim
- first reductive cyclization for chiral non-aromatic heterocycles synthesis
- high enantioselectivity with broad functional group tolerance

J. Am. Chem. Soc. **2020**, *142*, 15654–15660.

2.1 C=C bonds

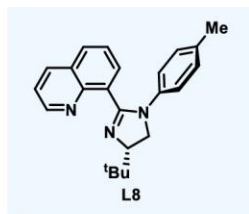
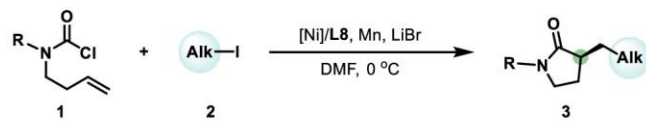


Entry	Ligand	Catalyst	Solvent	Temp. (°C)	3a (%) ^b	e.r. ^c
1	L1	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	0	-
2	L2	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	12	56:44
3	L3	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	33	68:32
4	L4	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	54	85:15
5	L5	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	17	51:49
6	L6	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	0	-
7	L7	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	0	-
8	L8	Ni(ClO ₄) ₂ •6H ₂ O	NMP	30	30	92:8
9	L8	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	55	96:4
10	L9	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	39	95:5
11	L10	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	46	96:4
12	L11	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	21	84:16
13	L12	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	36	91:9
14	L13	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	39	84:16
15	L14	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	9	96:4
16	L15	Ni(ClO ₄) ₂ •6H ₂ O	NMP	0	12	70:30
17	L8	Ni(ClO ₄) ₂ •6H ₂ O	DMF	0	64	97:3
18 ^d	L8	Ni(cod) ₂	DMF	0	92 (86)	97:3
19 ^{d,e}	L8	Ni(cod) ₂	DMF	0	0	-
20 ^{d,f}	L8	Ni(cod) ₂	DMF	0	16	83.5:16.5

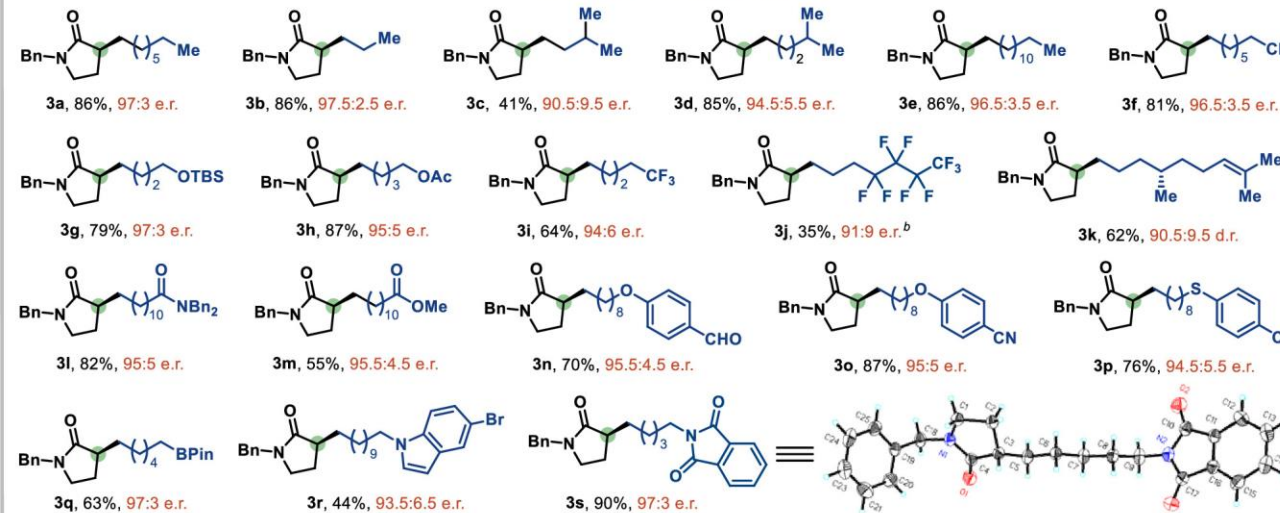


^aReaction conditions: 1a (0.2 mmol), 2a (3.0 equiv., 0.6 mmol), [Ni] (15 mol%, 0.03 mmol), ligand (22 mol%, 0.044 mmol), Mn (4.0 equiv., 0.8 mmol), LiBr (1.0 equiv., 0.2 mmol), solvent (1 mL) under the indicated temperature for 48 h. ^bCorrected GC yield. ^cDetermined by HPLC analysis. ^dNi(cod)₂ (15 mol%, 0.03 mmol); L8 (18 mol%, 0.036 mmol), 48 h, isolated yield in parentheses. ^eⁿC₇H₁₅Br (3.0 equiv) was used instead of 2a. ^fZn (4.0 equiv) was used instead of Mn.

2.1 C=C bonds

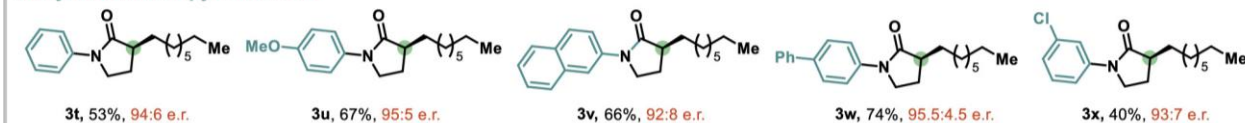


Variation of alkyl iodide

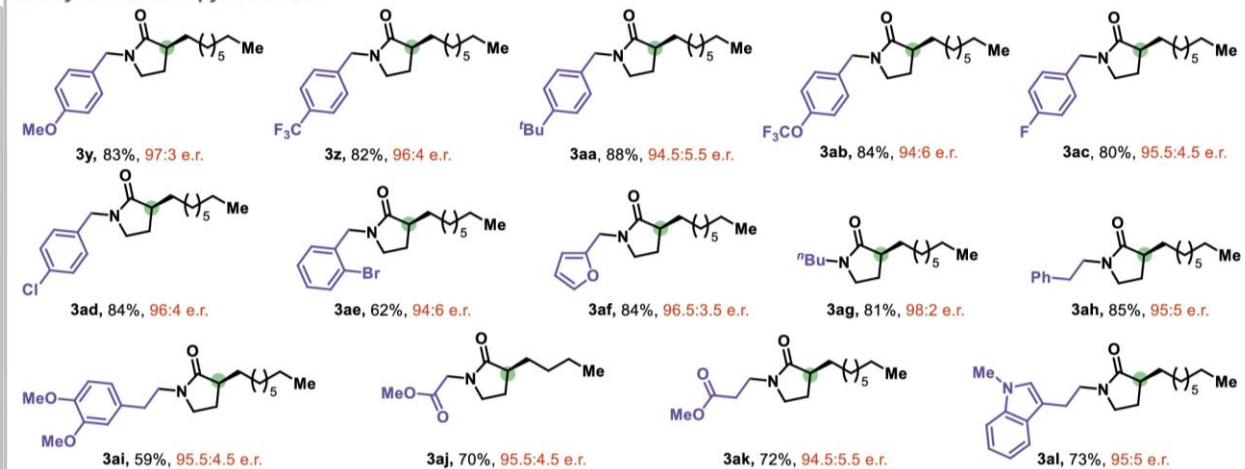


Variation of carbamoyl chloride

N-aryl substituted pyrrolidinone



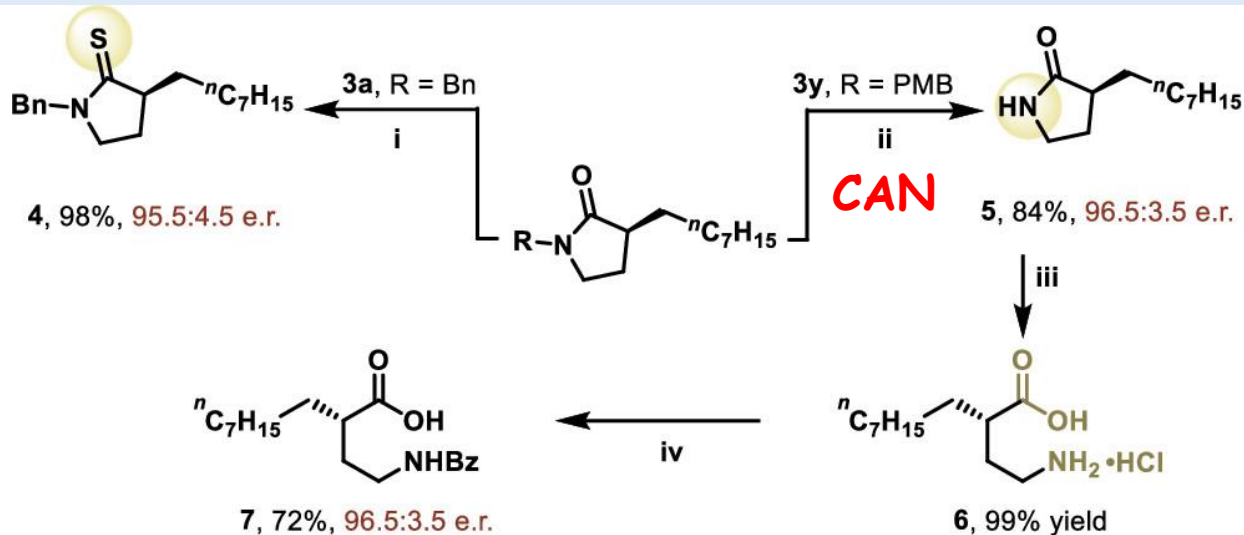
N-alkyl substituted pyrrolidinone



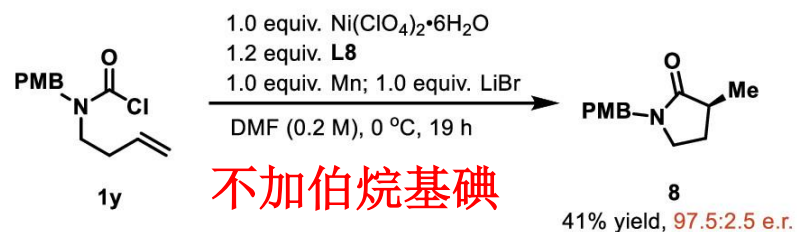
Limitations



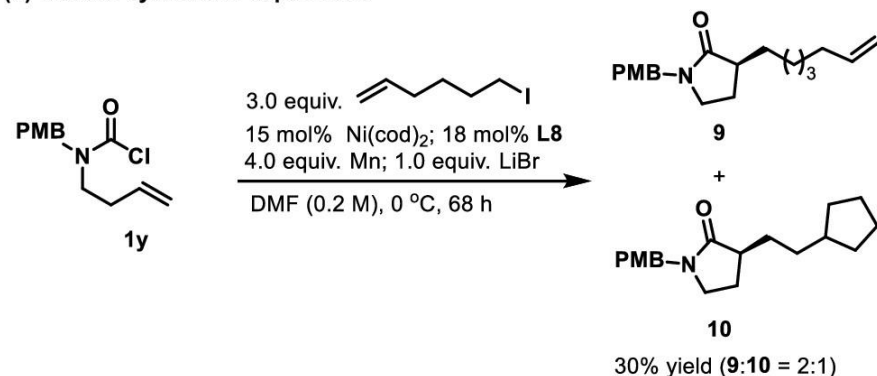
2.1 C=C bonds



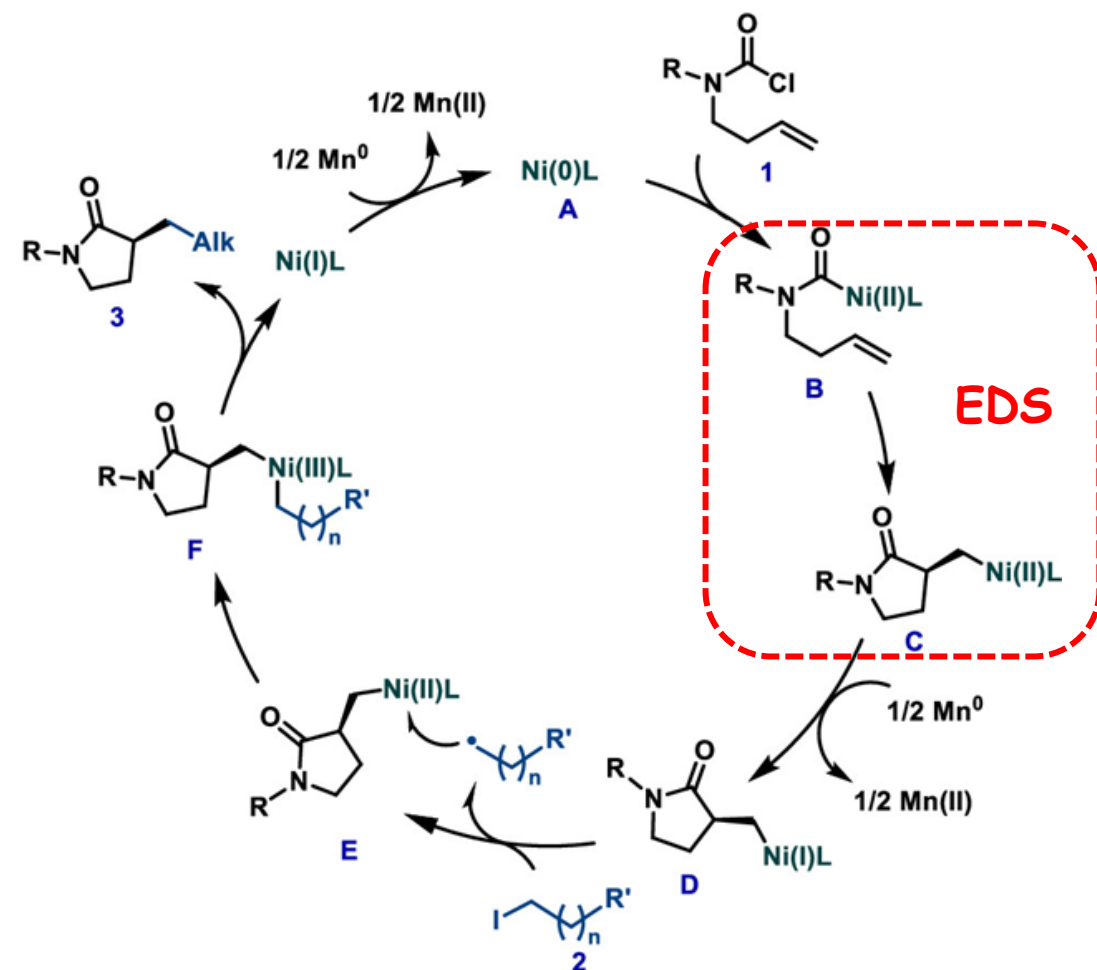
(a) Stoichiometric experiment



(b) Radical cyclization experiment



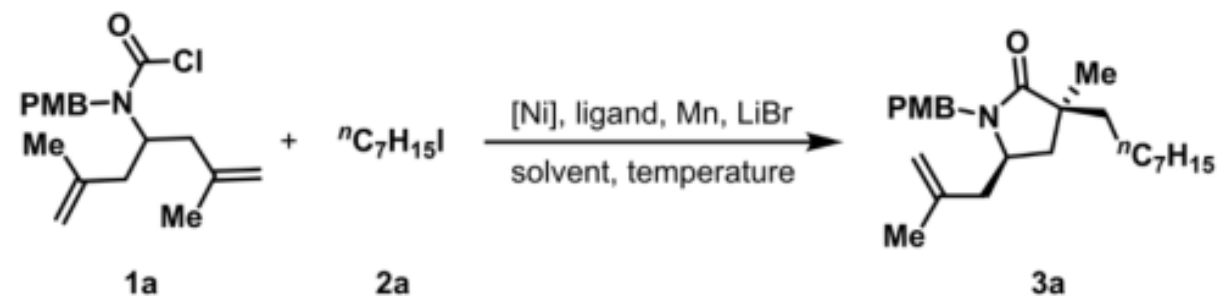
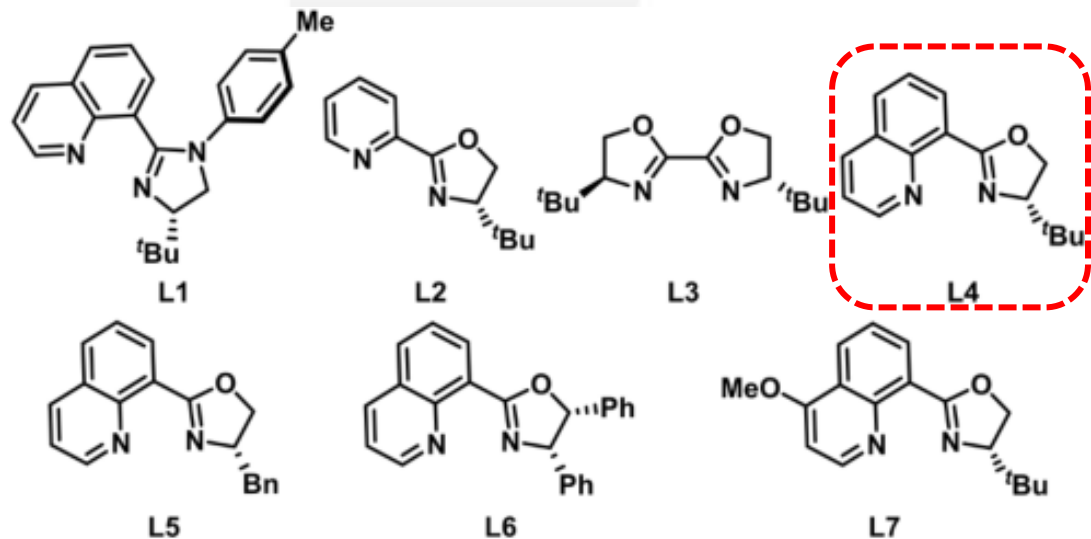
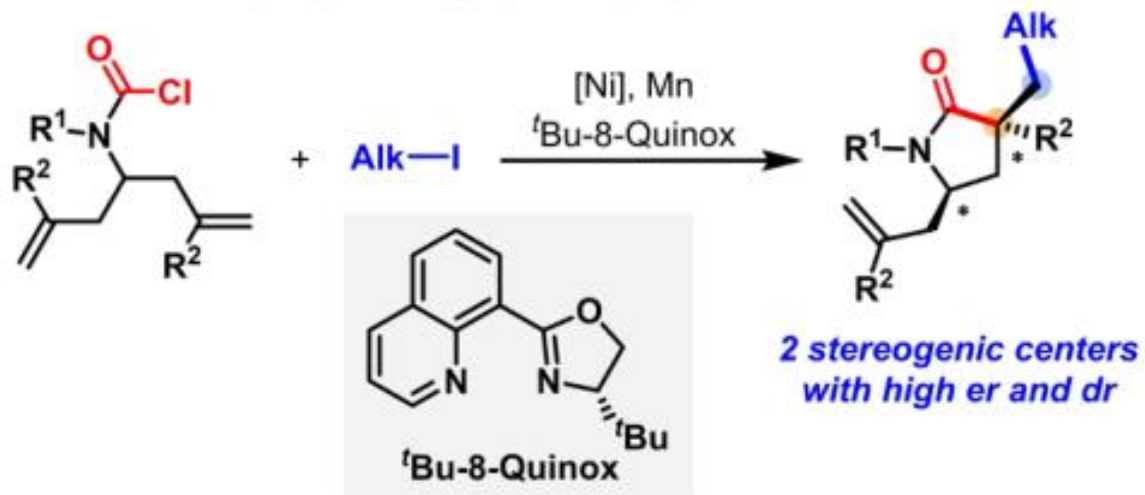
(c) Tentative mechanism



2.1 C=C bonds

c) Desymmetric dicarbofunctionalization of unactivated alkene

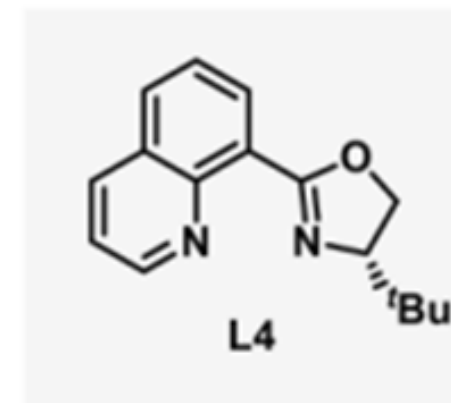
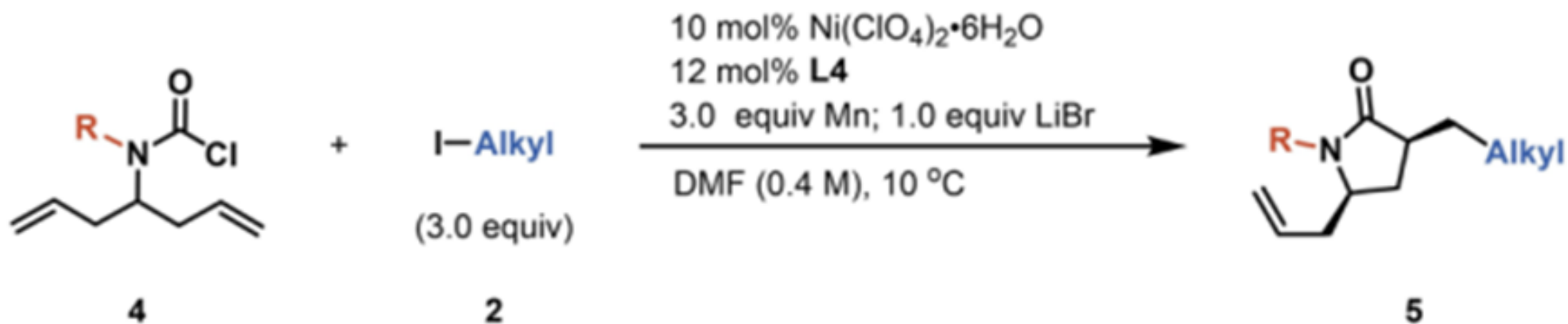
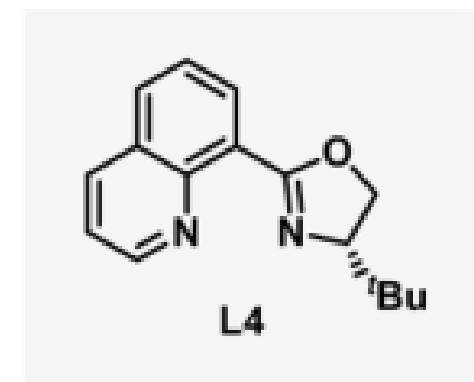
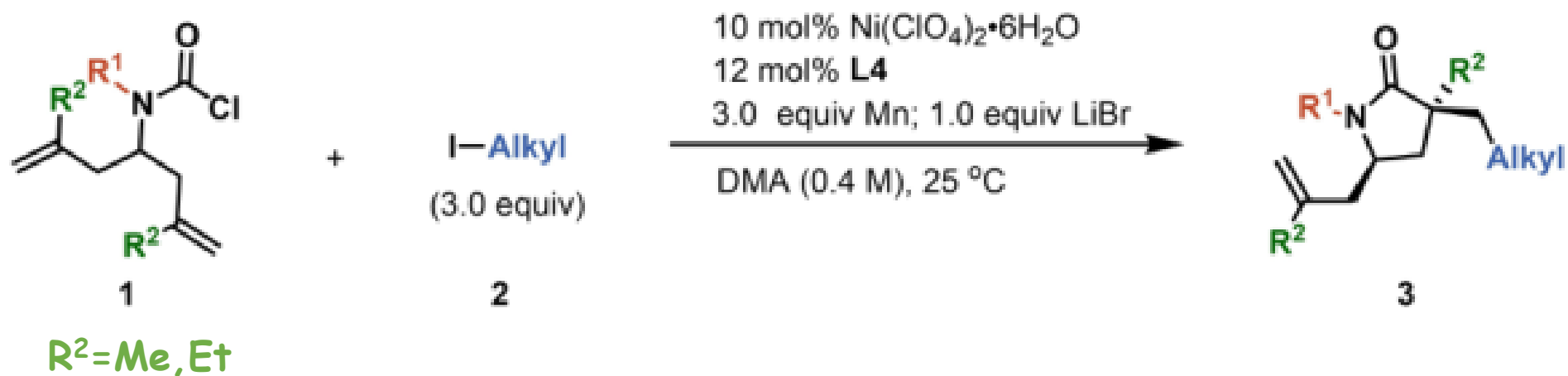
■ Cross-coupling strategy (this report)



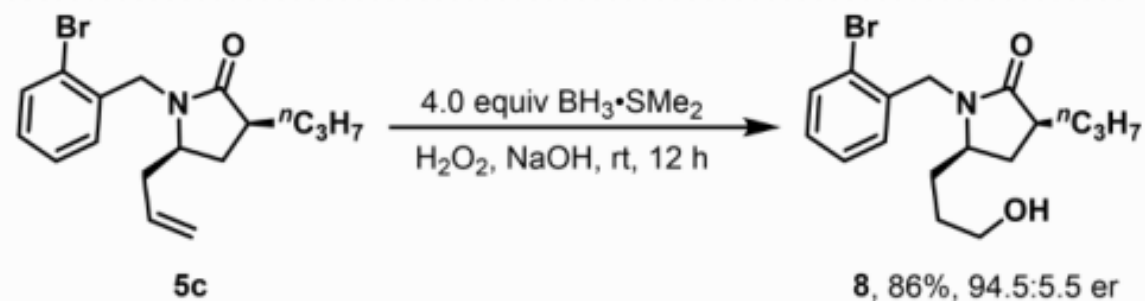
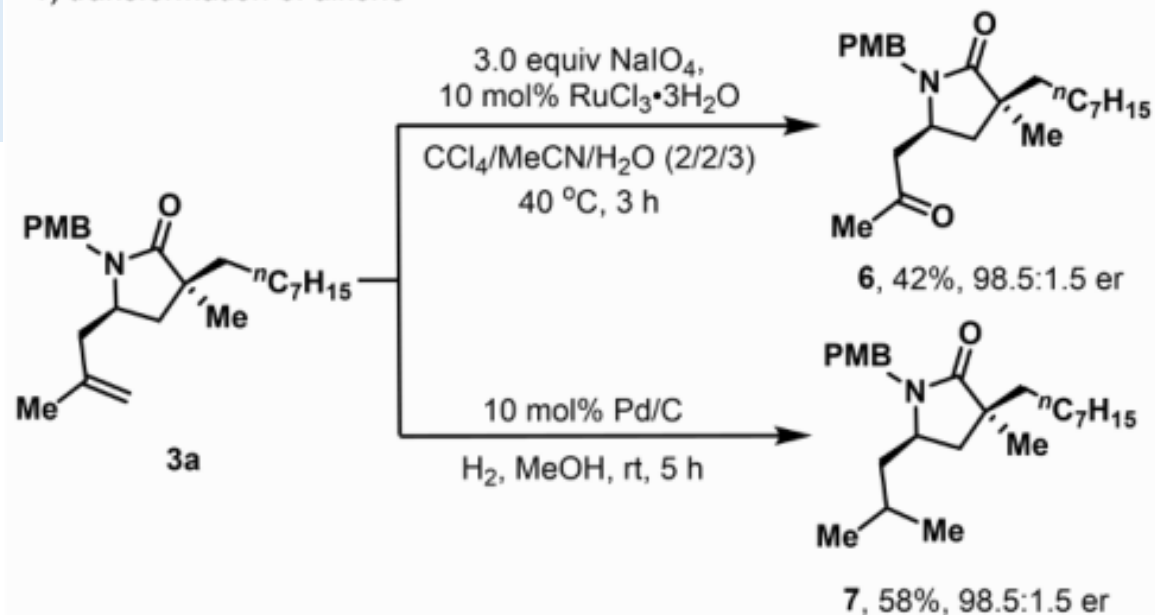
entry	ligand	catalyst	solvent	T (°C)	3a ^[b]	er ^[c]	dr ^[d]
1	L1	NiBr ₂ •DME	DMF (0.2 M)	10	27	90:10	>20:1
2	L2	NiBr ₂ •DME	DMF (0.2 M)	10	0	-	-
3	L3	NiBr ₂ •DME	DMF (0.2 M)	10	0	-	-
4	L4	NiBr ₂ •DME	DMF (0.2 M)	10	29	94:6	>20:1
5	L4	NiBr ₂ •DME	DMA (0.2 M)	10	34	94:6	>20:1
6	L4	NiBr ₂ •DME	DMA (0.2 M)	25	37	95:5	>20:1
7	L4	NiBr ₂ •DME	DMA (0.4 M)	25	53	96.5:3.5	>20:1
8	L4	Ni(ClO ₄) ₂ •6H ₂ O	DMA (0.4 M)	25	92	98.5:1.5	>20:1
9	L5	Ni(ClO ₄) ₂ •6H ₂ O	DMA (0.4 M)	25	78	75.5:24.5	17:1
10	L6	Ni(ClO ₄) ₂ •6H ₂ O	DMA (0.4 M)	25	43	67:33	3:1
11	L7	Ni(ClO ₄) ₂ •6H ₂ O	DMA (0.4 M)	25	92	98:2	>20:1

Angew. Chem. Int. Ed. 2022, 61, e202111598.

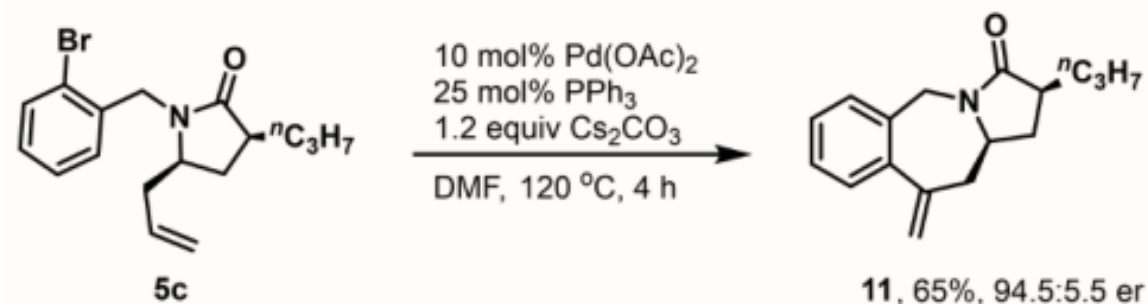
2.1 C=C bonds



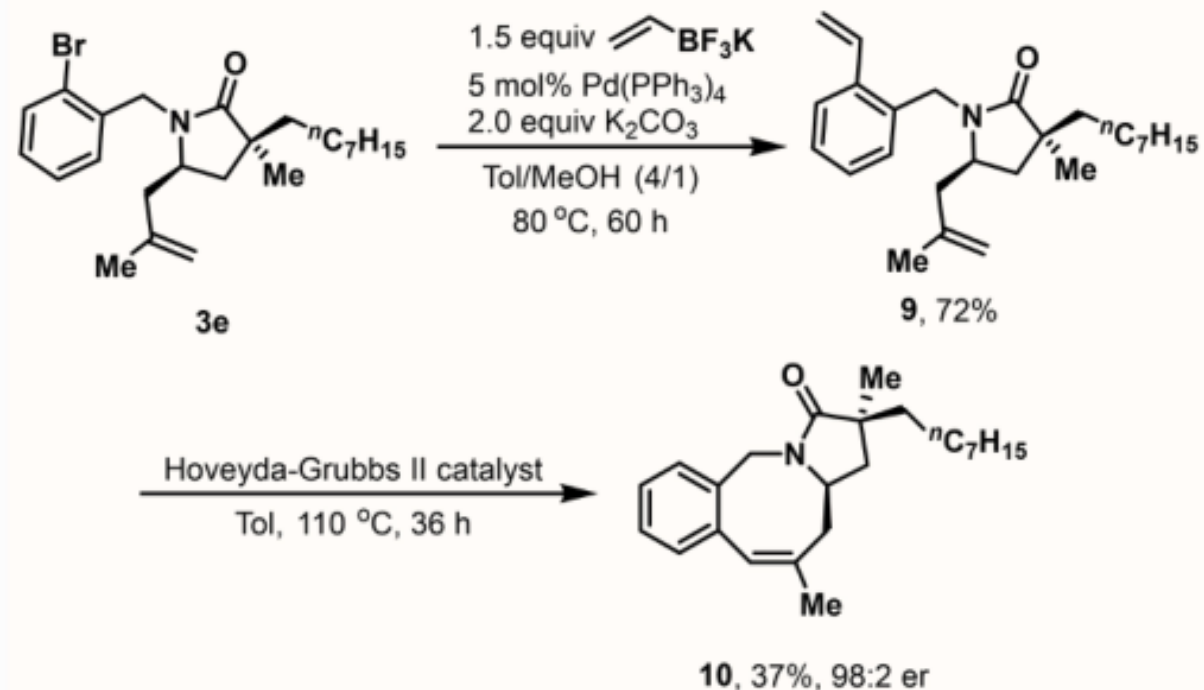
1) transformation of alkene



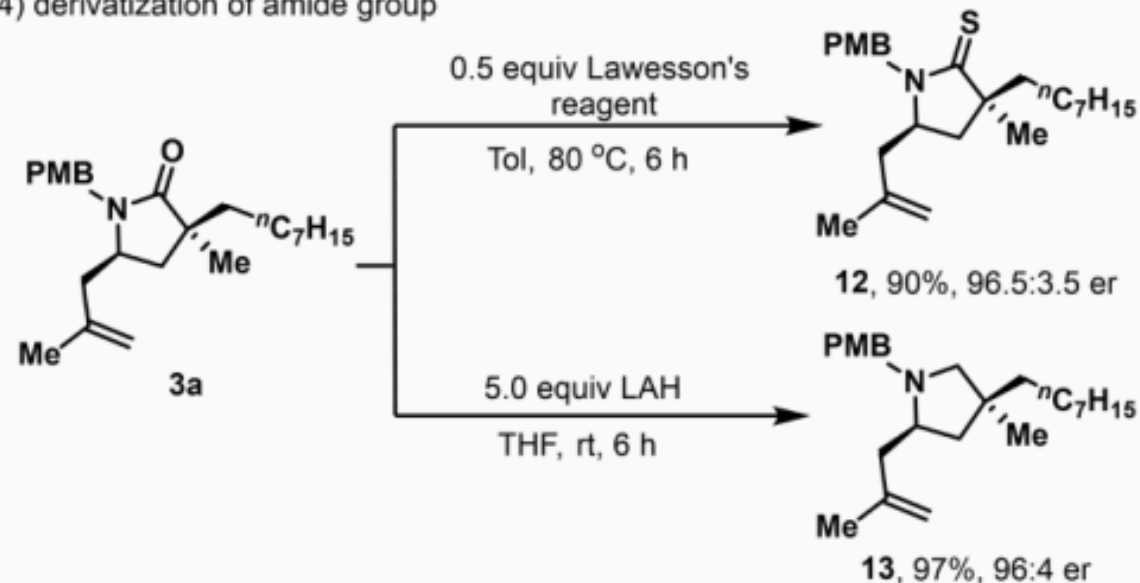
3) Heck reaction



2) RCM reaction

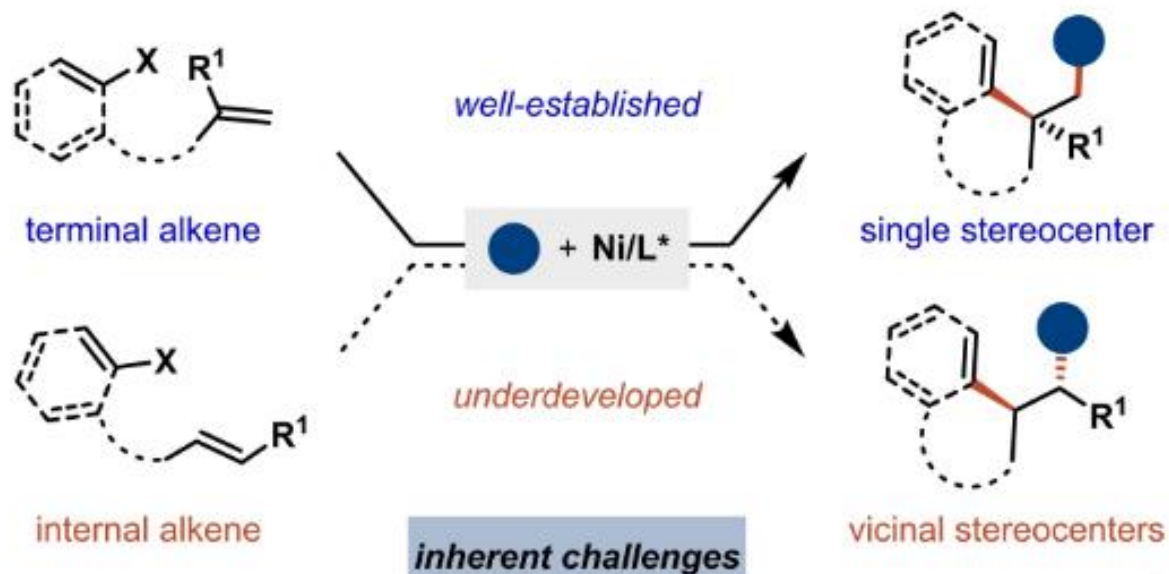


4) derivatization of amide group



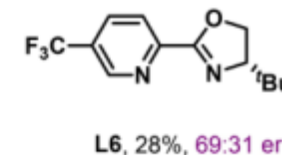
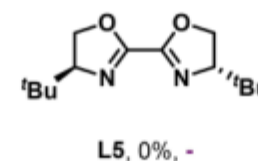
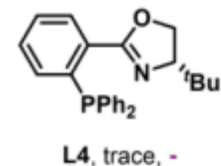
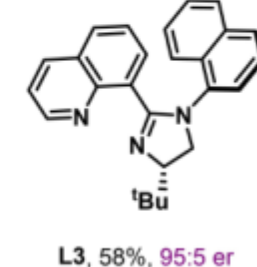
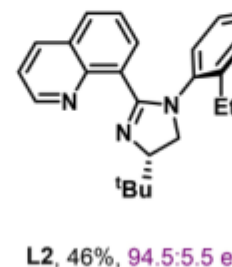
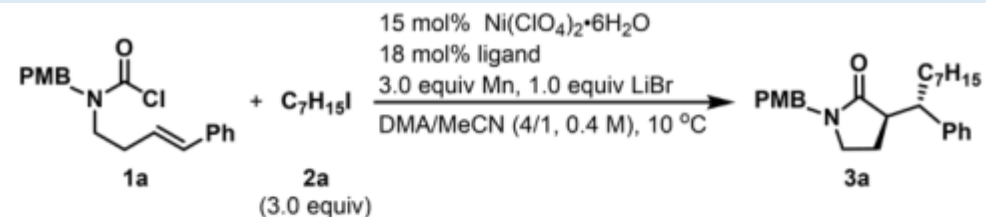
2.1 C=C bonds

a) Ni-catalyzed enantioselective difunctionalization of pendant alkenes



- low affinity of internal alkenes for migratory insertion
- diastereoselectivity issue for vicinal stereogenic centers
- potential Heck reaction through undesired β -H elimination

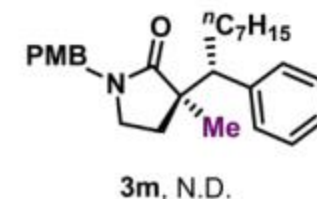
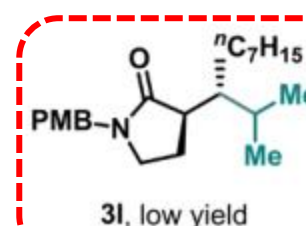
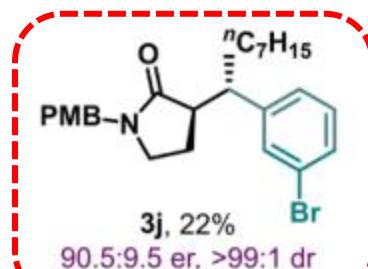
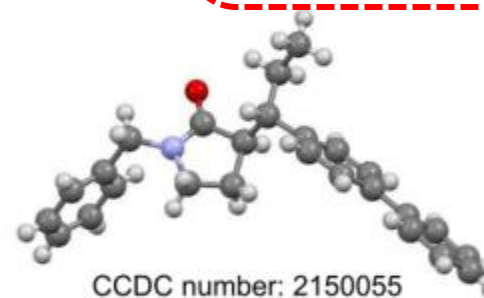
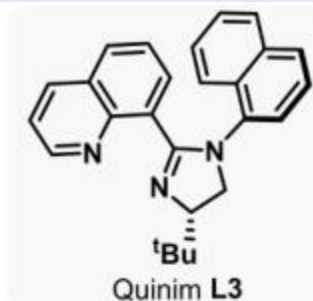
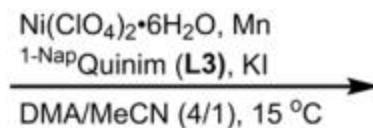
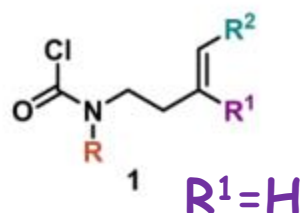
analysis. All the diastereoselectivities are $>20:1$. [d] $\text{Ni}(\text{ClO}_4)_2 \cdot 6\text{H}_2\text{O}$ (20 mol %, 0.020 mmol), L3 (24 mol %, 0.024 mmol) with 1.0 equiv KI (1.0 equiv, 0.1 mmol) as additive; isolated yield is reported in parentheses.



Entry	Deviation from standard condition	3a (%) ^[b]	e.r. ^[c]
1	none	58	95:5
2	DMA as solvent	51	89:11
3	MeCN as solvent	<5	-
4	$\text{Ni}(\text{cod})_2$ instead of $\text{Ni}(\text{ClO}_4)_2 \cdot 6\text{H}_2\text{O}$	30	85.5:14.5
5	$\text{NiBr}_2 \cdot \text{DME}$ instead of $\text{Ni}(\text{ClO}_4)_2 \cdot 6\text{H}_2\text{O}$	36	89:11
6	Zn instead of Mn	5	85.5:14.5
7	KI instead of LiBr	67	97:3
8 ^[d]	15 °C	77 (69)	97:3

Angew. Chem. Int. Ed. 2022, 61, e202207536.

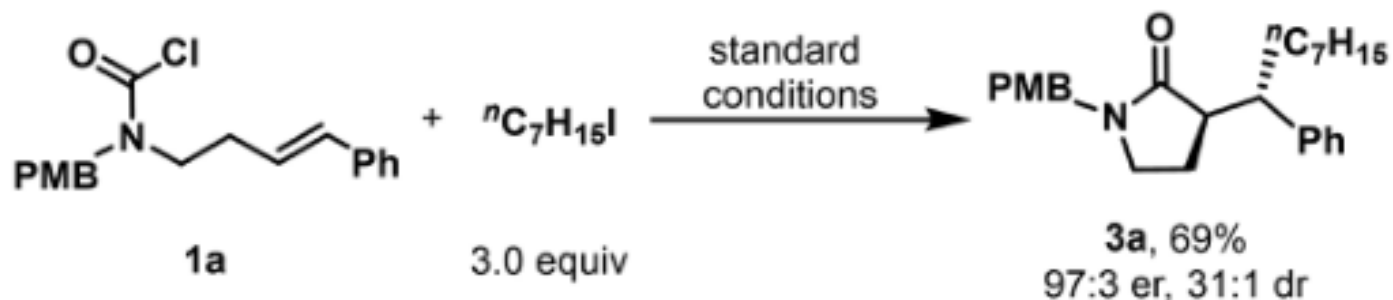
2.1 C=C bonds



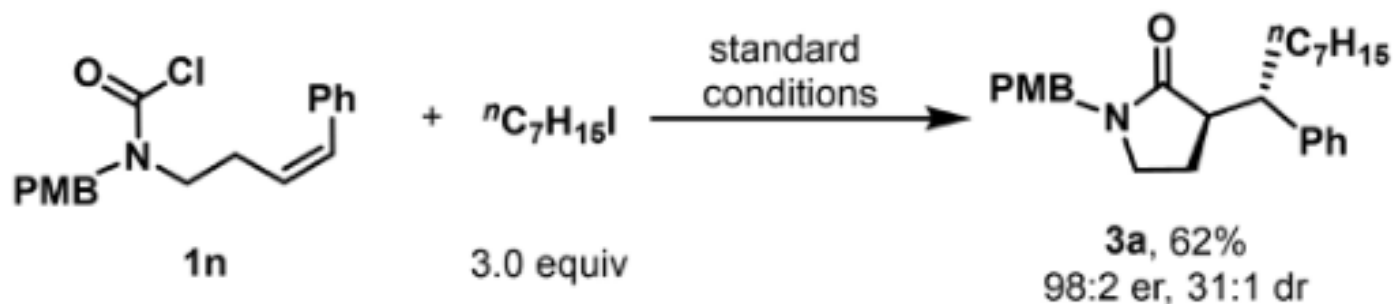
乙烯基芳烃可能用于稳定Ni中间体

2.1 C=C bonds

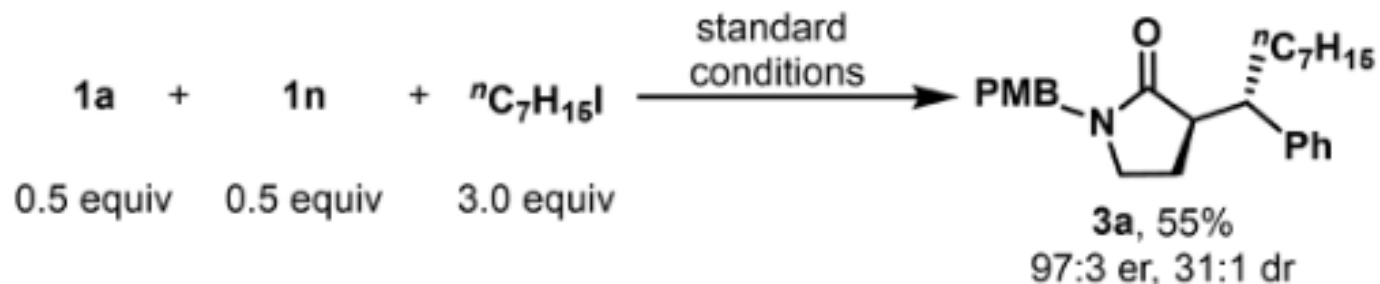
(a) Ni-catalyzed enantioselective alkyl-carbamoylation of *trans*-alkenes



(b) Ni-catalyzed enantioselective alkyl-carbamoylation of *cis*-alkenes



(c) crossover experiments of *trans*- and *cis*-alkenes



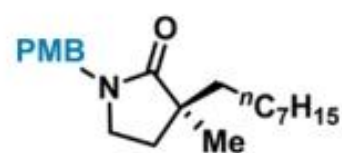
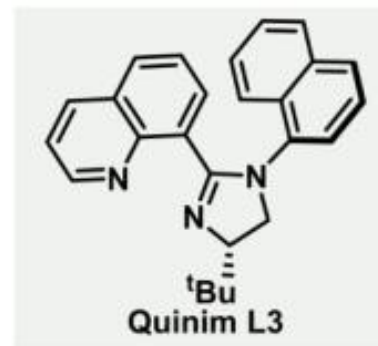
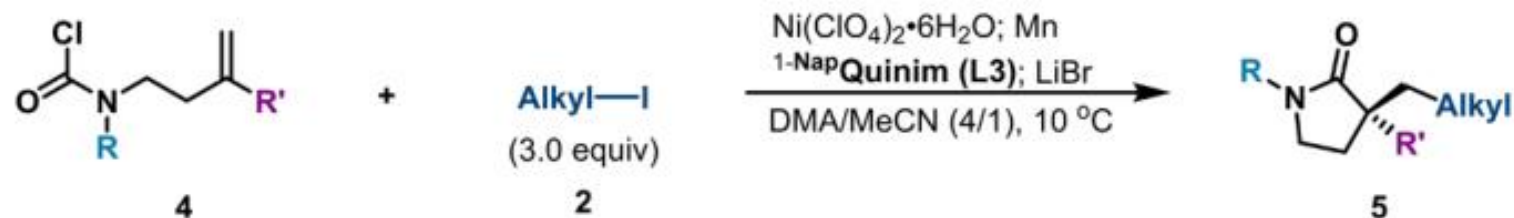
两种不同**Z/E**构型的烯烃底物



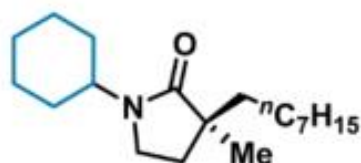
相同的主要非对映异构体

吡咯烷酮产物的立体化学与
烯烃起始原料的构型无关

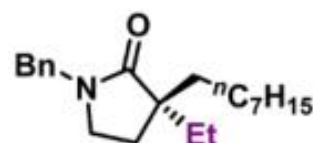
2.1 C=C bonds



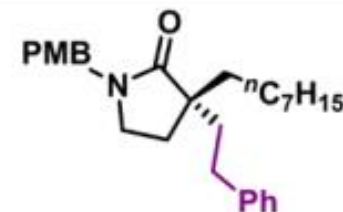
5a, 85%, 96.5:3.5 er
74%, 92:8 er^[b]



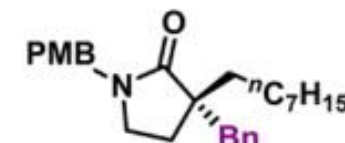
5b, 88%, 96.5:3.5 er



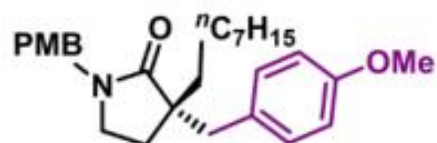
5c, 91%, 96.5:3.5 er



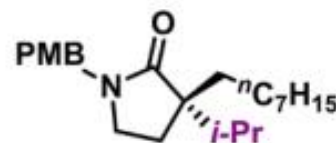
5d, 95%, 97.5:2.5 er



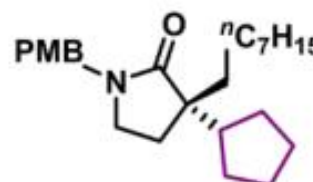
5e, 54%, 96:4 er



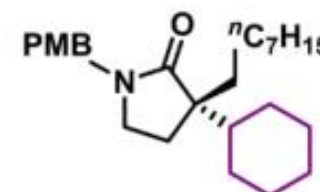
5f, 88%, 95.5:4.5 er



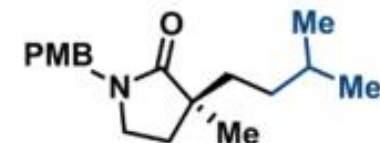
5g, 63%, 91.5:8.5 er



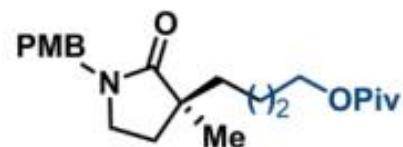
5h, 65%, 95.5:4.5 er



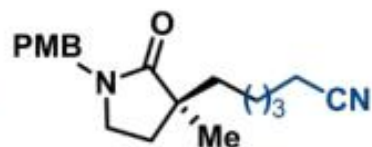
5i, 65%, 94.5:5.5 er



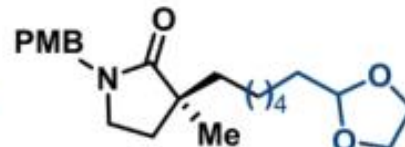
5j, 72%, 95.5:4.5 er



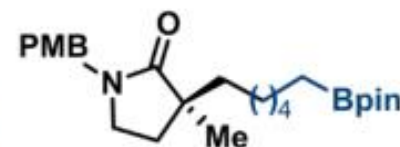
5k, 99%, 97:3 er



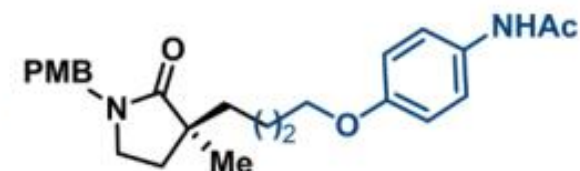
5l, 74%, 96:4 er



5m, 95%, 97:3 er



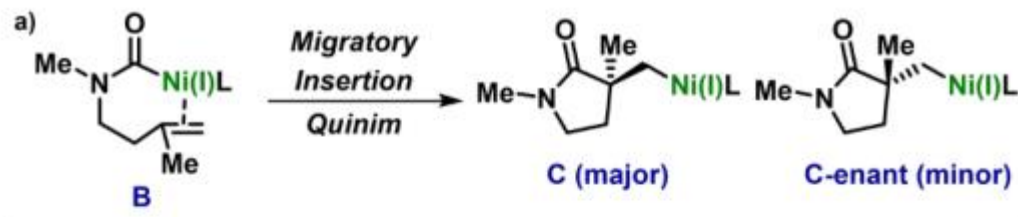
5n, 78%, 97.5:2.5 er



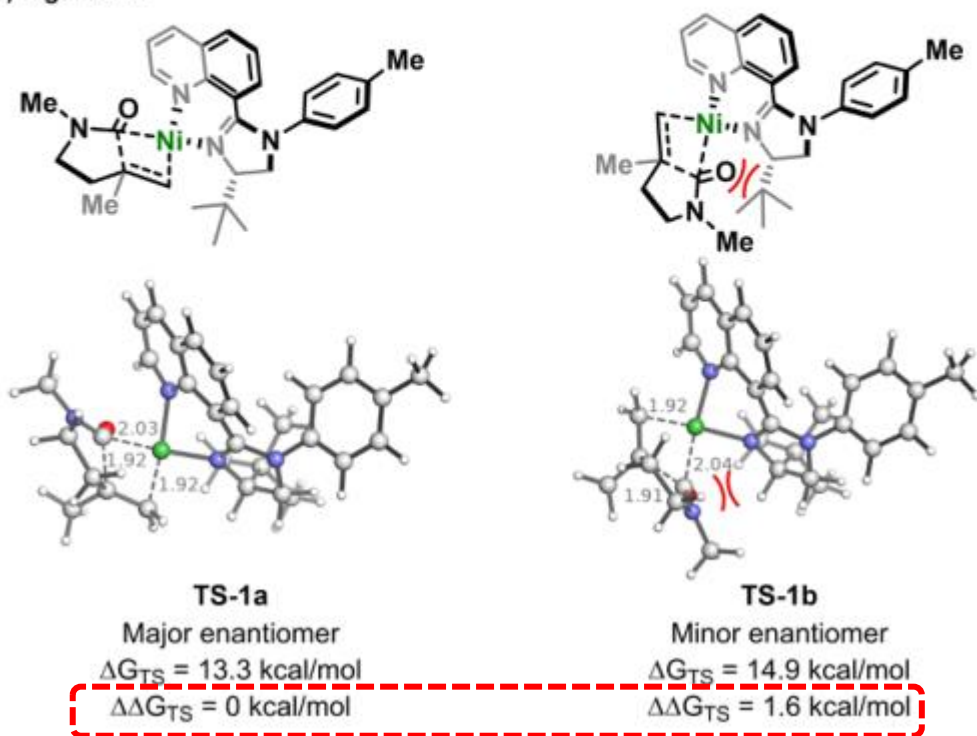
5o, 79%, 98.5:1.5 er

2.1 C=C bonds

对映选择性 (EDS)：迁移插入



b) Ligand: L1



c) Ligand: L3

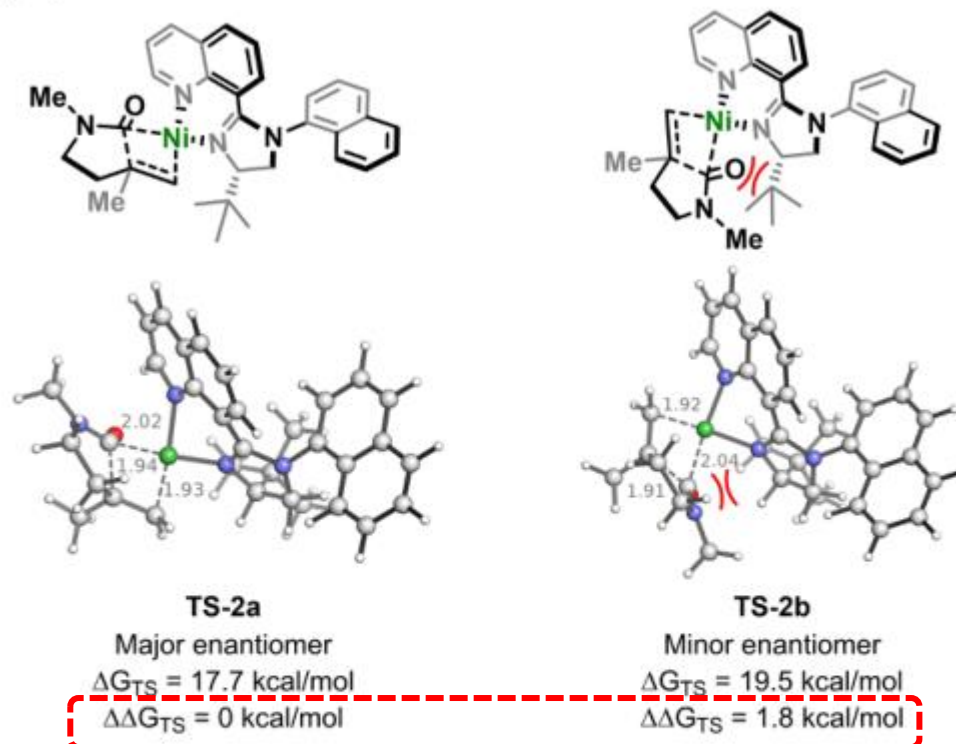


Figure 1. a) Migratory insertion step of model substrate B. b) Comparison of transition states for the formation of two enantiomers with ligand L1. c) Comparison of transition states for the formation of two enantiomers with ligand L3.

底物的取向方式：将羰基和N-取代基定位在配体的另一侧

修订反应机理

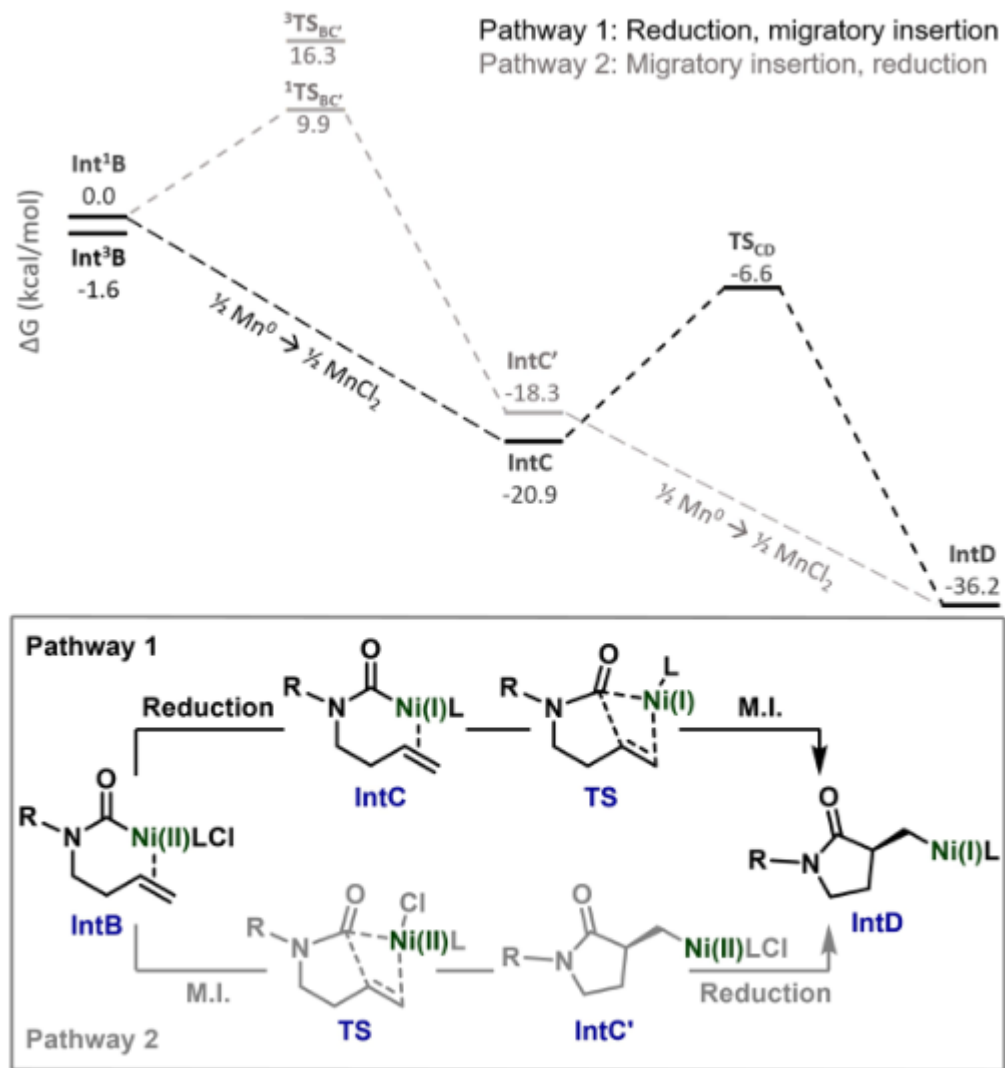


Figure 2. Gibbs free energy profile of two possible pathways for conversion of **IntB** to **IntD**. Energies in kcal mol⁻¹.

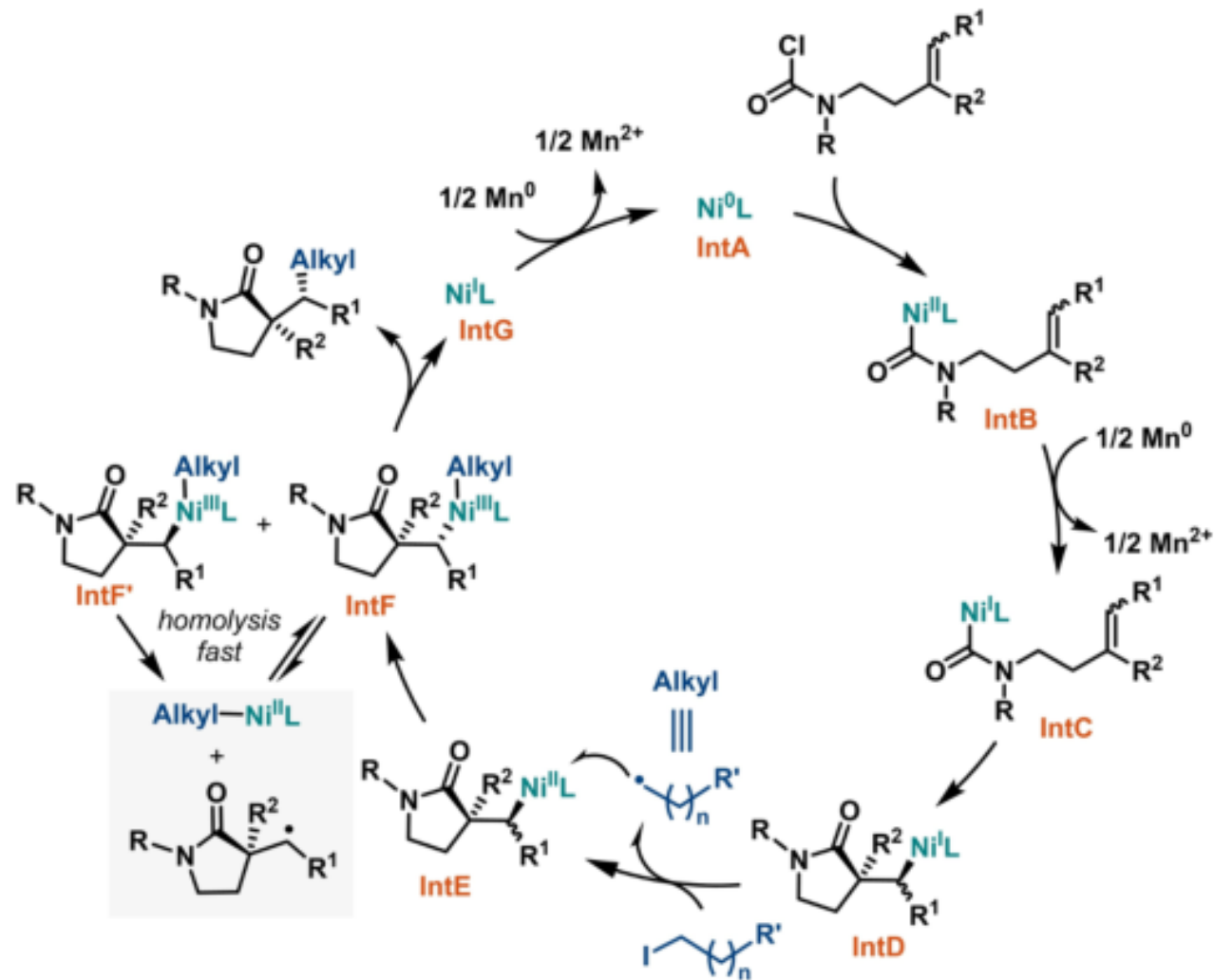
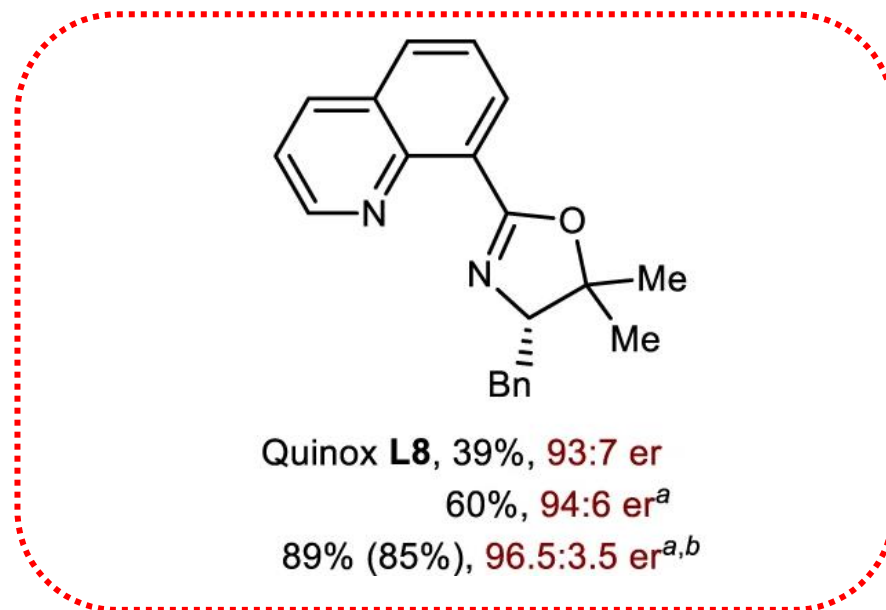
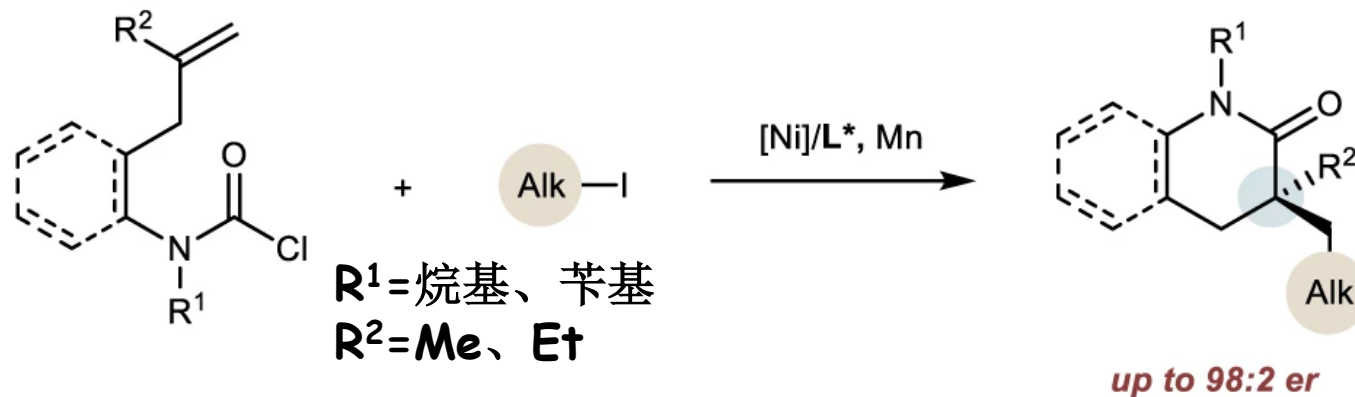


Figure 3. Proposed mechanistic cycle for reductive cyclization.

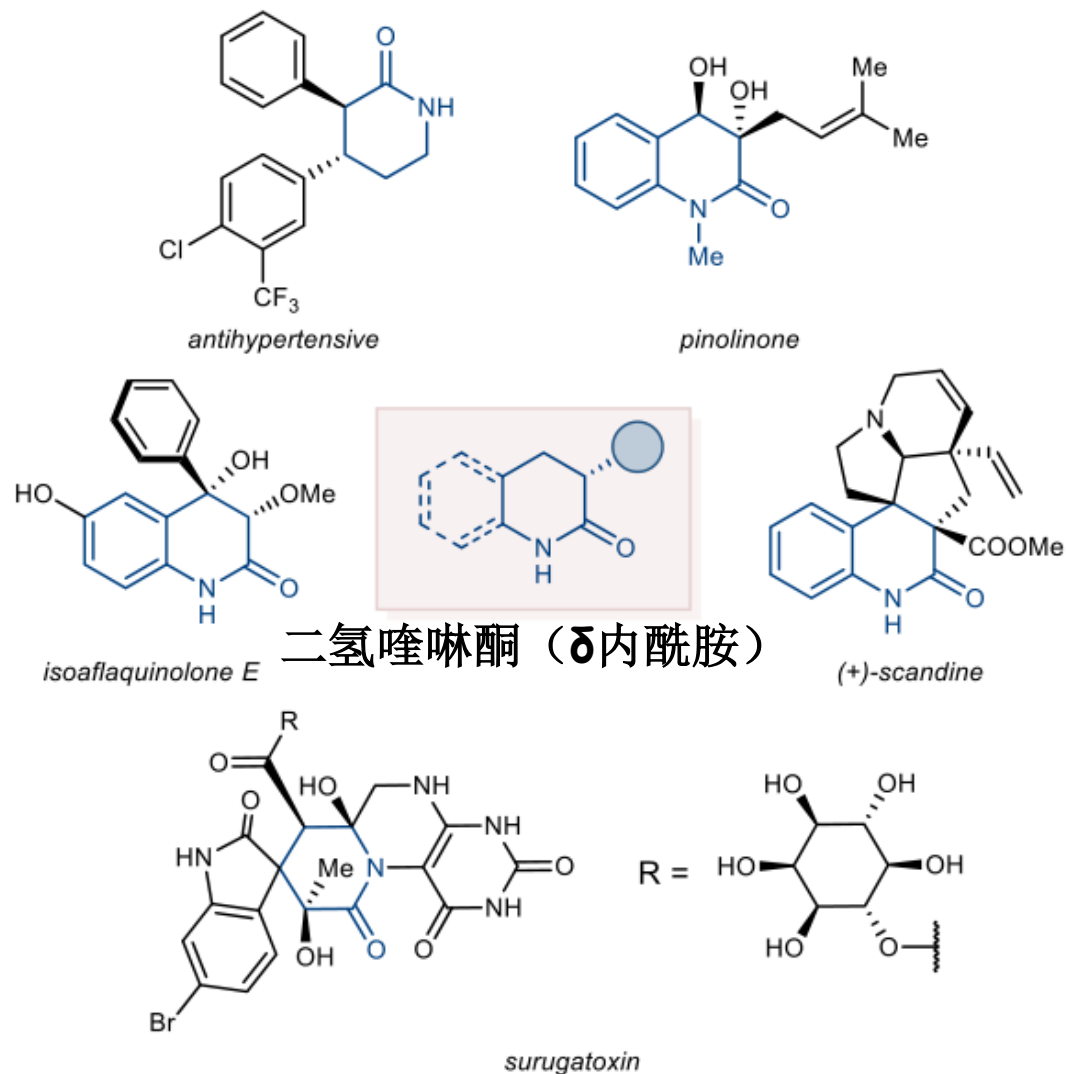
2.1 C=C bonds

c) synthetic gap: asymmetric carbamoylation for six-membered lactam synthesis



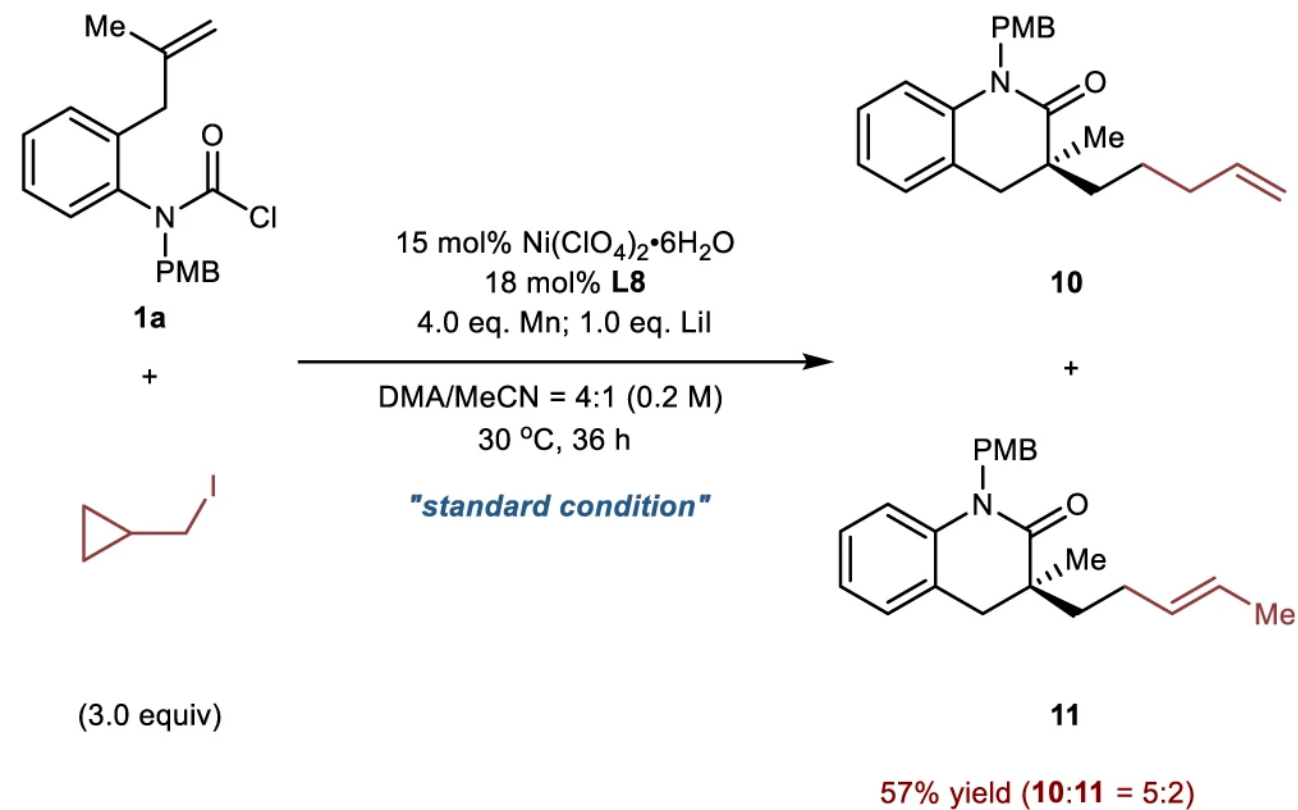
Nat. Commun. 2022, 13, 5964.

d) biologically active molecules bearing six-membered lactam scaffold



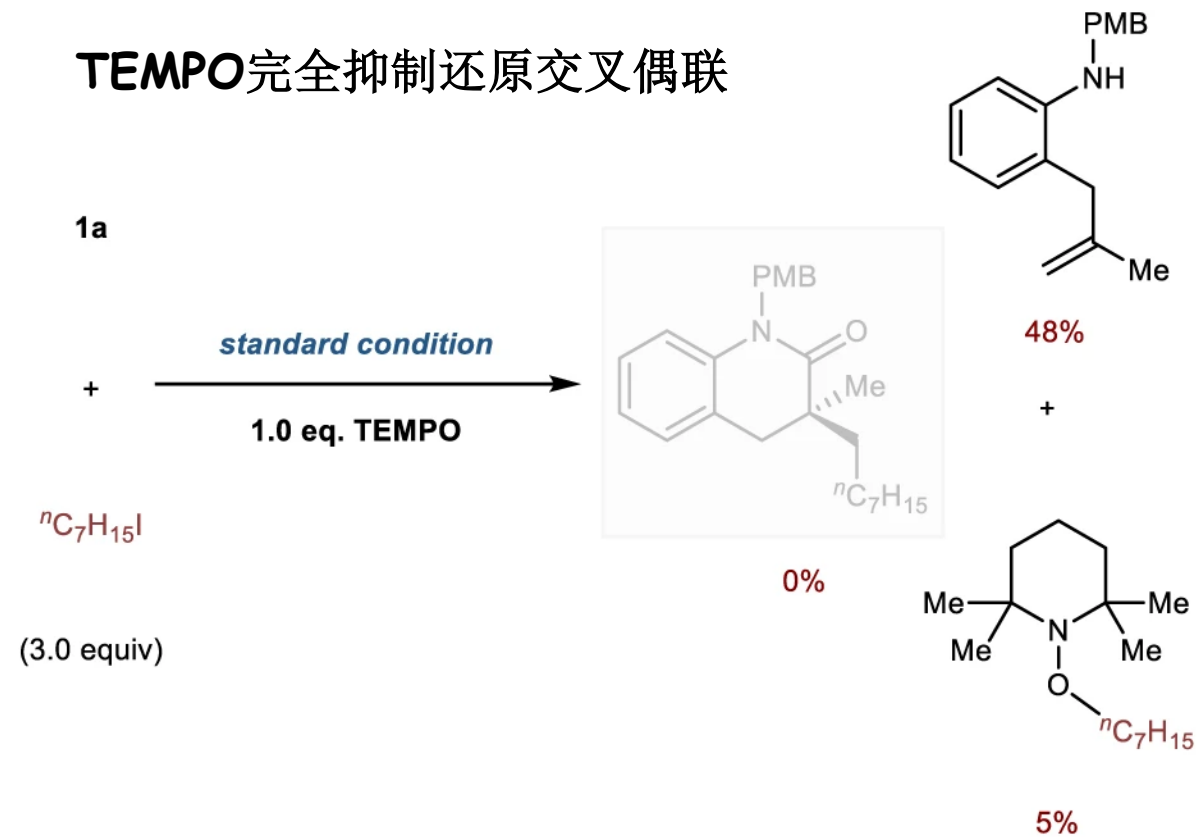
2.1 C=C bonds

a) Radical ring-opening experiment



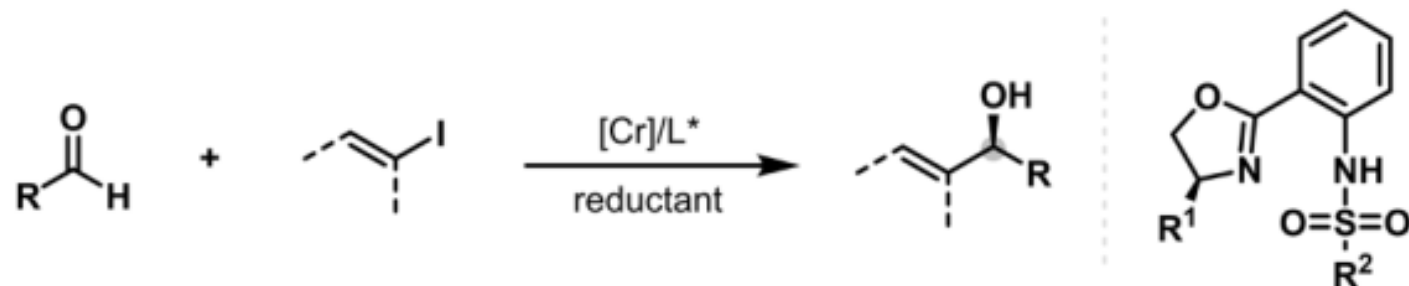
b) Radical trap reaction

TEMPO完全抑制还原交叉偶联

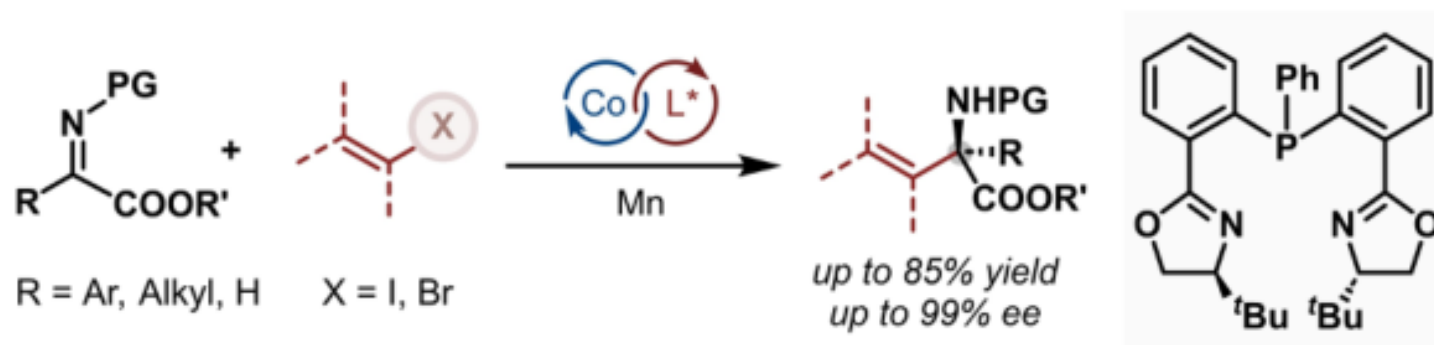


2.2 C=N bonds

(a) Asymmetric NHK (Nozaki–Hiyama–Kishi) reaction of aldehyde (Established protocol)



(b) Asymmetric aza-NHK reaction of α -imino ester with alkenyl halides (This work)

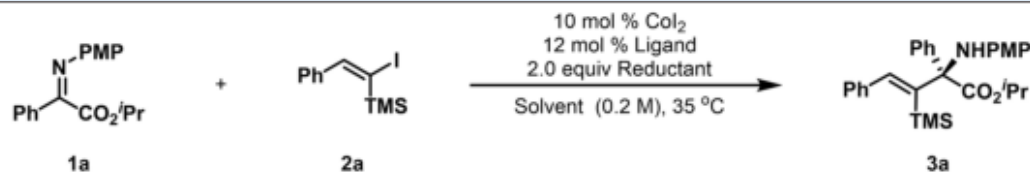


- expedient construction of chiral α -vinyl amino acid analogues
- asymmetric reductive addition reaction with alkenyl electrophiles
- excellent enantioselectivity with broad substrate scope

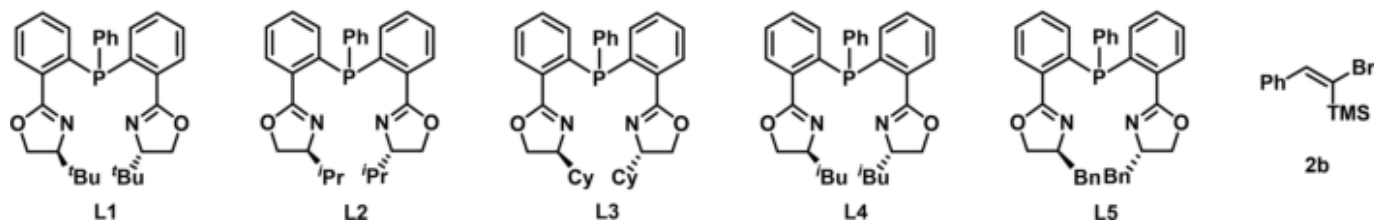
Scheme 1. The development of catalytic asymmetric aza-NHK reaction.

Angew. Chem. Int. Ed. **2024**, 63, e202316012.

2.2 C=N bonds



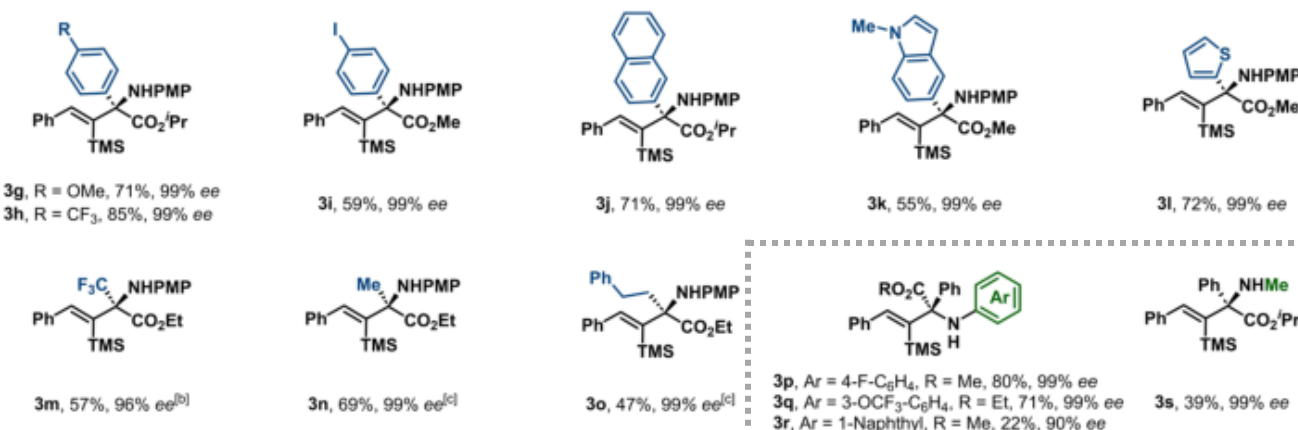
Entry	Ligand	Reductant	Solvent	Alkenyl halide	3a (%) ^[b]	ee (%) ^[c]
1	L1	Mn	THF	2a (1.5 equiv)	59	72
2	L1	Mn	THF with 1.0 equiv ⁱ PrOH	2a (1.5 equiv)	49	>99
3	L1	Mn	THF/ ⁱ PrOH = 4/1	2a (1.5 equiv)	51	>99
4	L2	Mn	THF/ ⁱ PrOH = 4/1	2a (1.5 equiv)	20	90
5	L3	Mn	THF/ ⁱ PrOH = 4/1	2a (1.5 equiv)	21	77
6	L4	Mn	THF/ ⁱ PrOH = 4/1	2a (1.5 equiv)	36	99
7	L5	Mn	THF/ ⁱ PrOH = 4/1	2a (1.5 equiv)	16	-
8	L1	Zn	THF/ ⁱ PrOH = 4/1	2a (1.5 equiv)	44	>99
9	L1	Mn	DME/ ⁱ PrOH = 4/1	2a (1.5 equiv)	79	>99
10	L1	Mn	MeCN/ ⁱ PrOH = 4/1	2a (1.5 equiv)	80	>99
11	L1	Mn	MeCN/ ⁱ PrOH = 4/1	2a (2.0 equiv)	85 (84) ^[de]	>99
12	L1	Mn	MeCN/ ⁱ PrOH = 4/1	2b (2.0 equiv)	85 ^[de]	>99



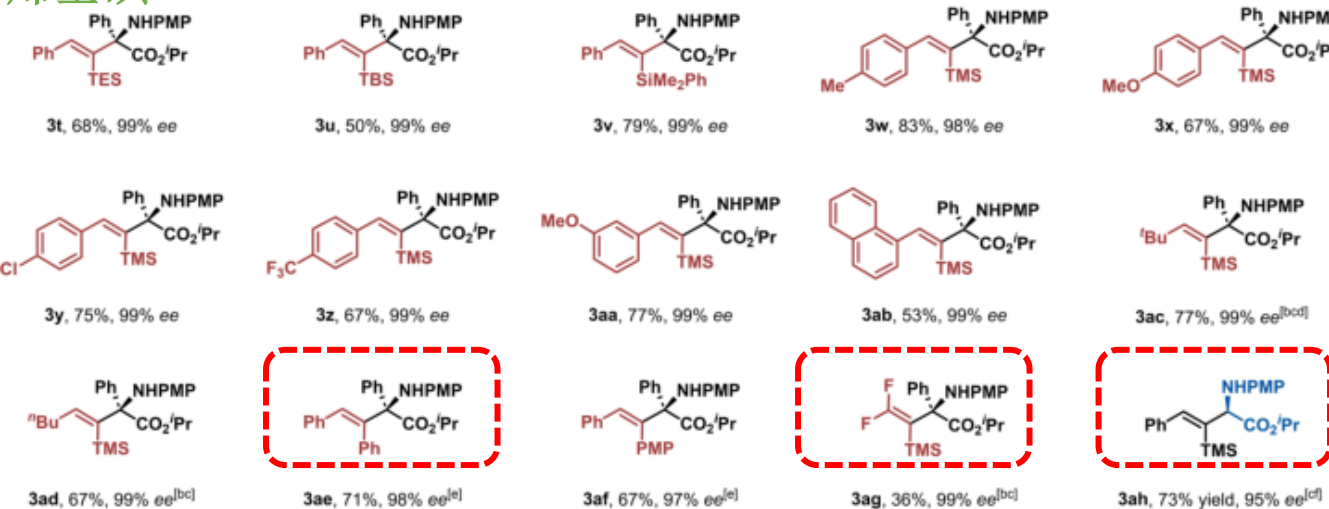
烯基溴化物有相似的效果

2.2 C=N bonds

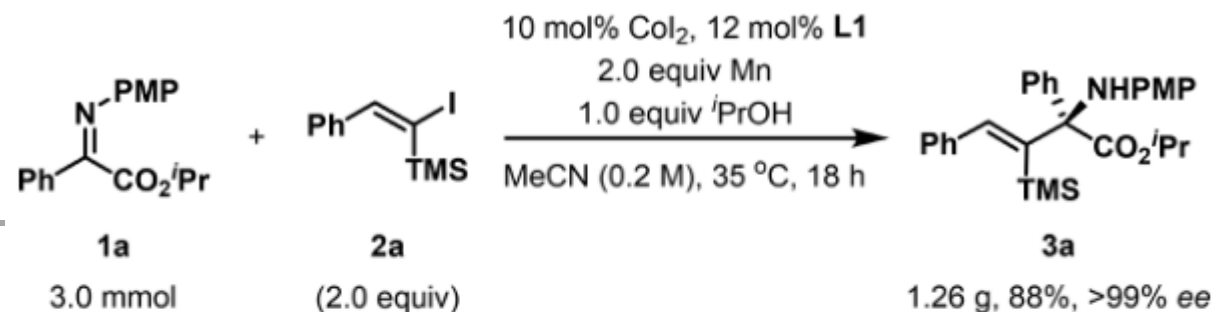
α亚胺酯



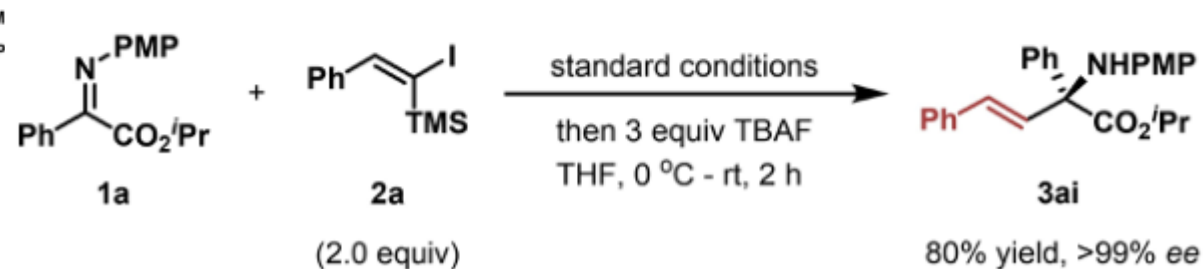
烯基碘



a) Gram-scale reaction



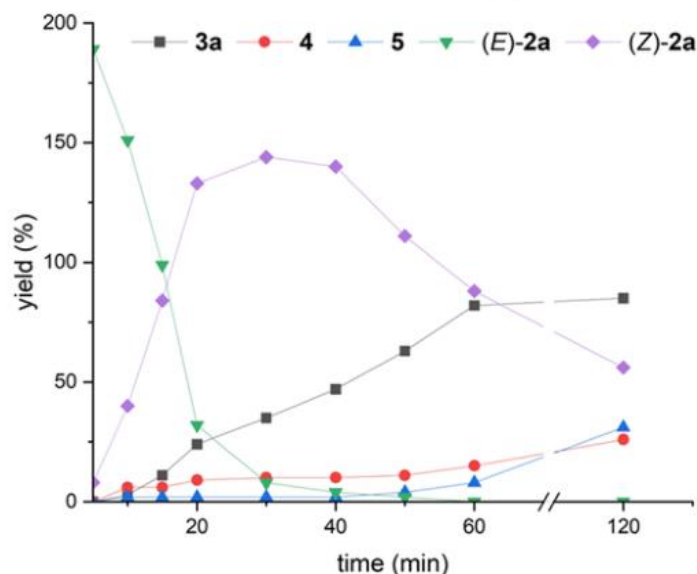
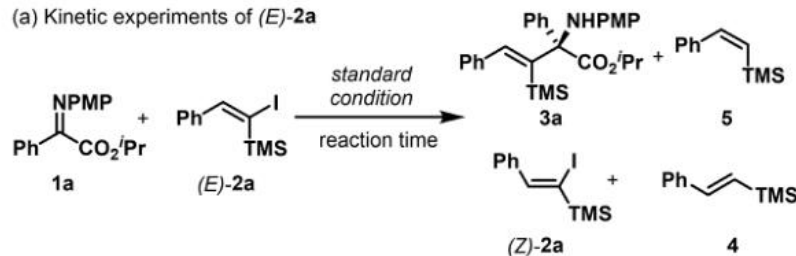
b) In-situ removal of TMS group



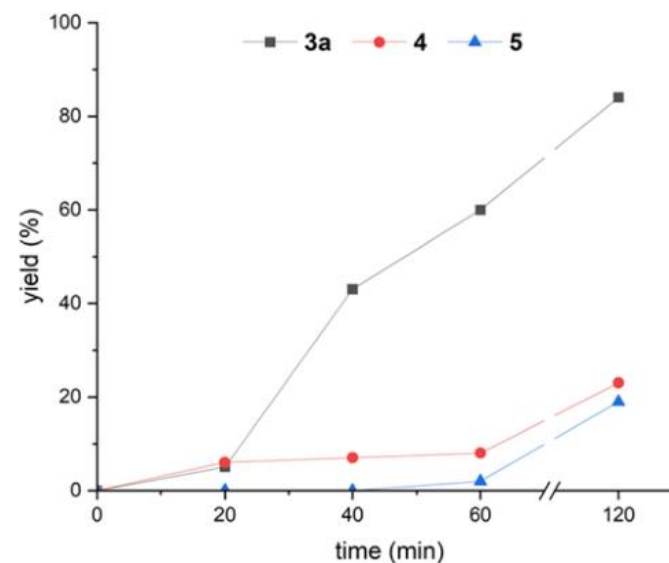
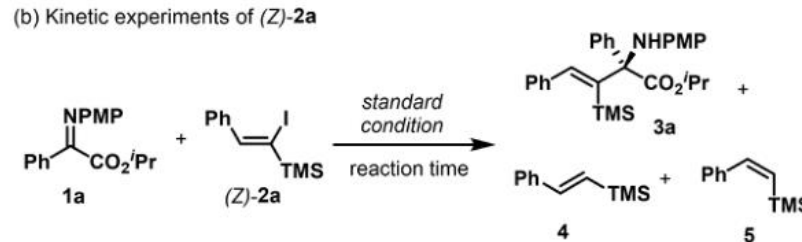
2.2 C=N bonds

实时示踪反应:

(a) Kinetic experiments of (E)-2a

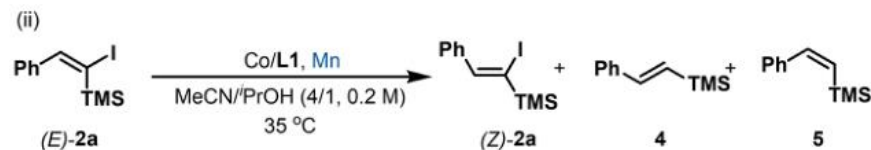
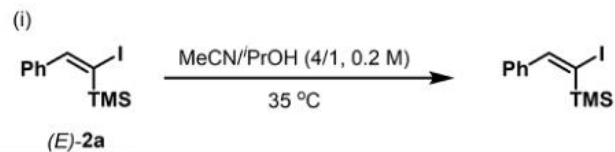


(b) Kinetic experiments of (Z)-2a



生成的(E)-烯基钴的异构化过程达到另一个Z构型的速度远快于加成!

(c) Control experiments



低价Co: 碘化烯基异构化的控制

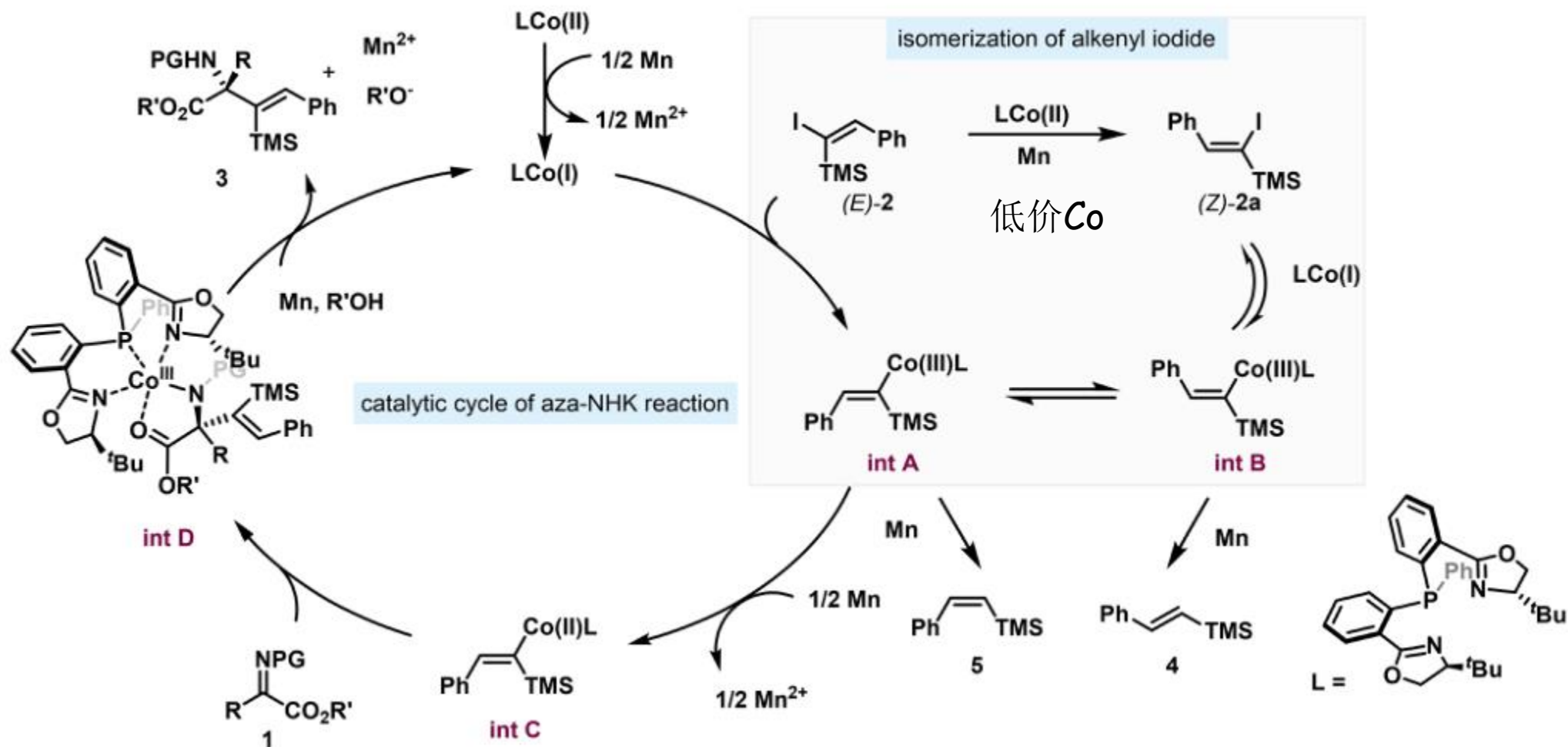
□ 不加CoI₂, L1 2.0 equiv Mn 98% of (E)-2a recovered

□ 不加Mn 10 mol% CoI₂; 12 mol% L1 98% of (E)-2a recovered

10 mol% CoI ₂ , 12 mol% L1, 10 mol% Mn	96%	5%	--
10 mol% CoI ₂ , 12 mol% L1, 2.0 equiv Mn	7%	52%	39%
1.0 equiv [^t Bu-NPN ^{Ph} -Co ^I], w/o Mn	46%	43%	7%

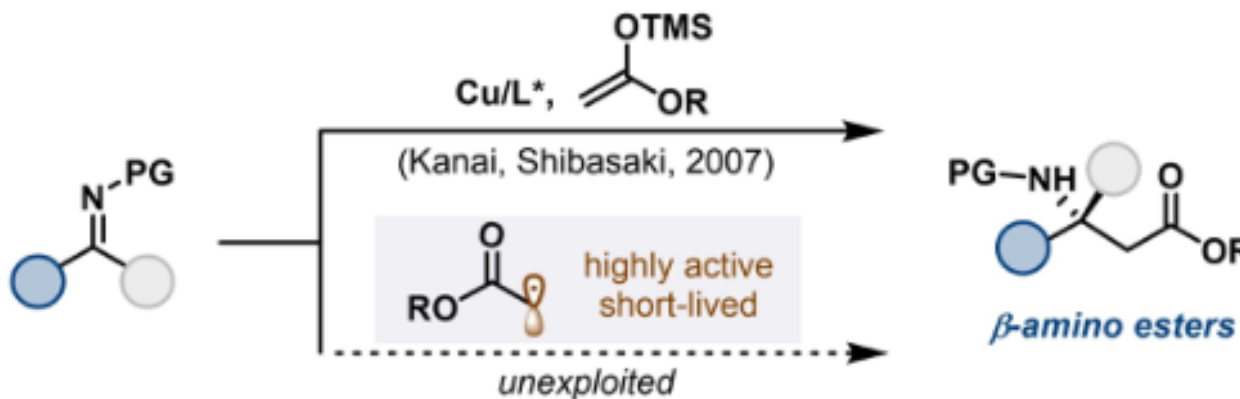
2.2 C=N bonds

(d) Plausible mechanism

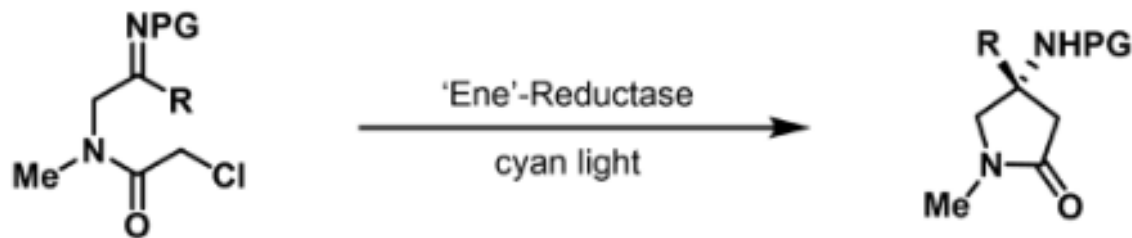


2.2 C=N bonds

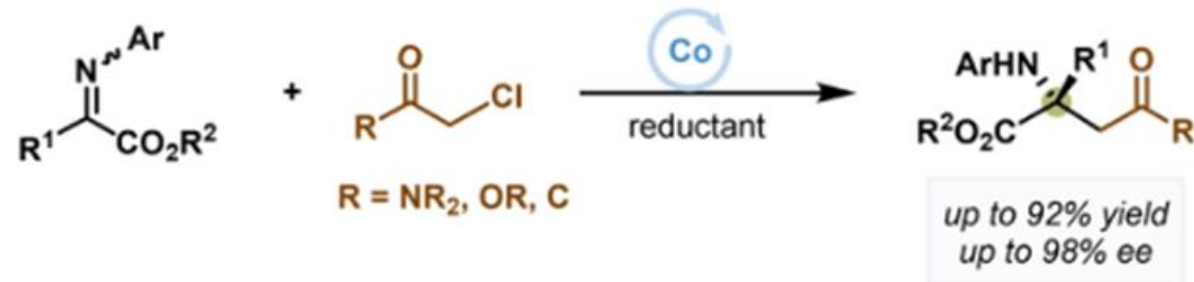
(a) Chiral β -tertiary amino ester synthesis enabled by the stereoselective addition



(b) Enzyme-catalyzed intramolecular asymmetric reductive addition of ketimine with α -chloro amide (Hyster, 2021)



(c) Cobalt-catalyzed intermolecular asymmetric reductive addition of ketimine with α -chloro carbonyl compound (This work)

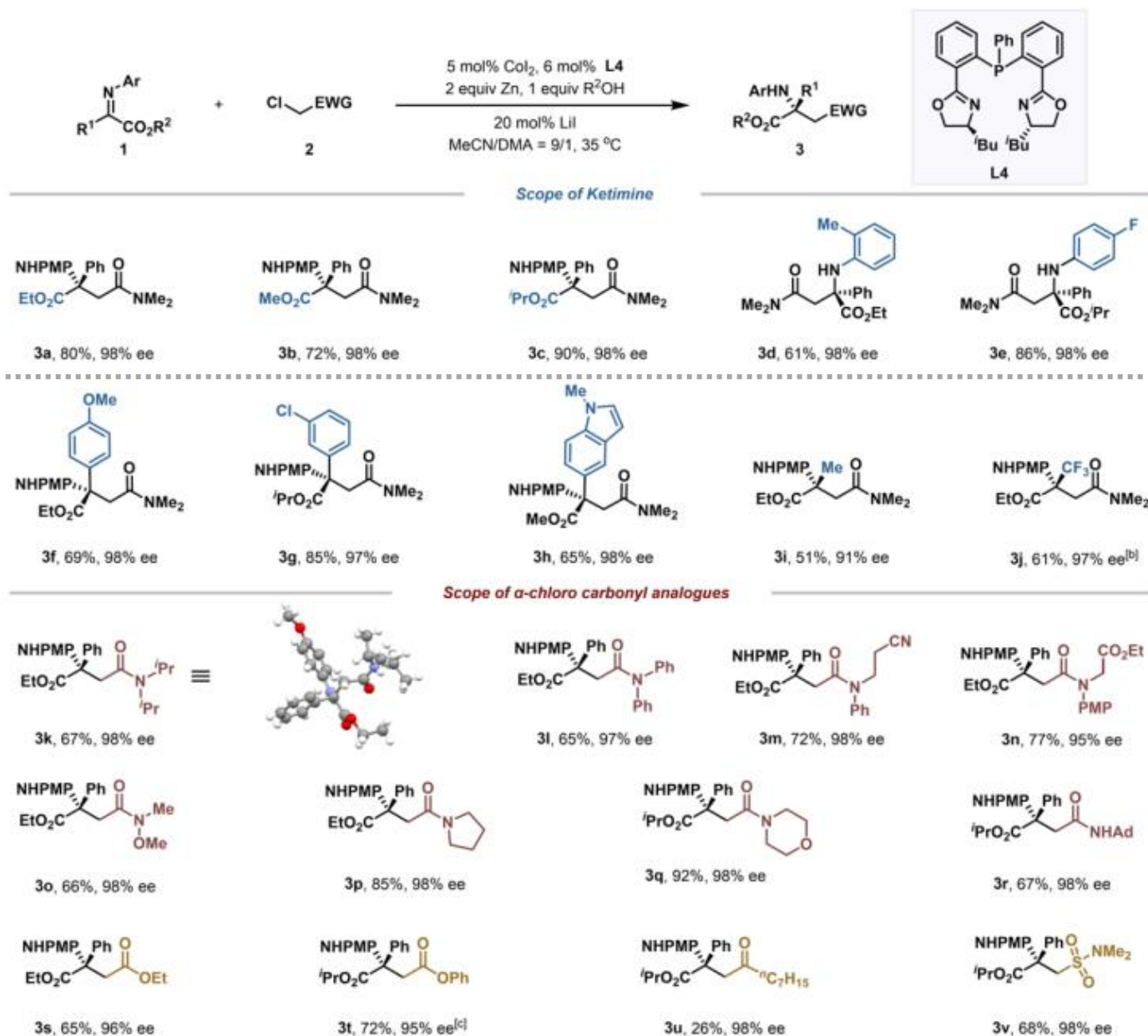


- transition metal-catalyzed stereoselective addition of α -carbonyl radical
- convenient synthesis of enantioenriched tertiary β -amino acid analogues
- excellent enantioselectivity with broad substrate scope under mild condition

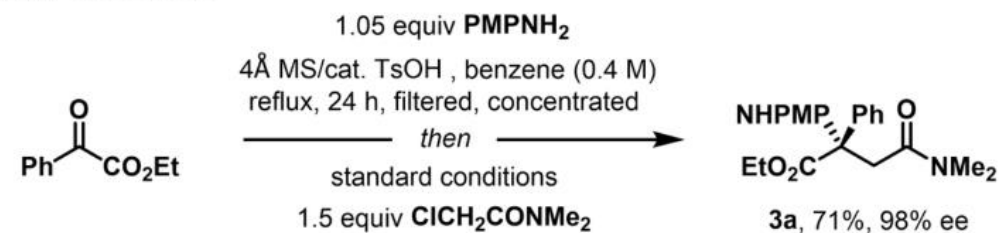
高反应活性的羰基 α 位自由基：易发生**Reformatsky**加成、自由基对亚胺酯直接加成，降低对映选择性。

Angew. Chem. Int. Ed. **2024**, 63, e202316012.

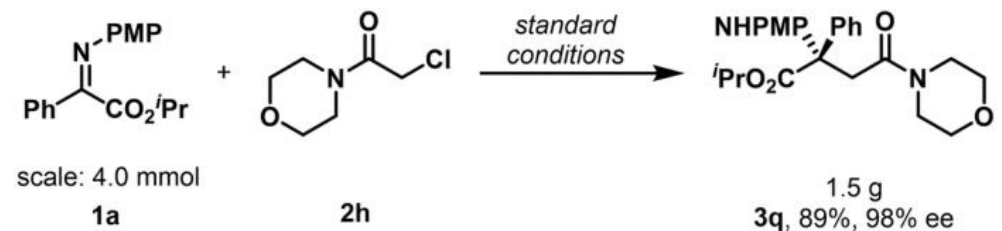
2.2 C=N bonds



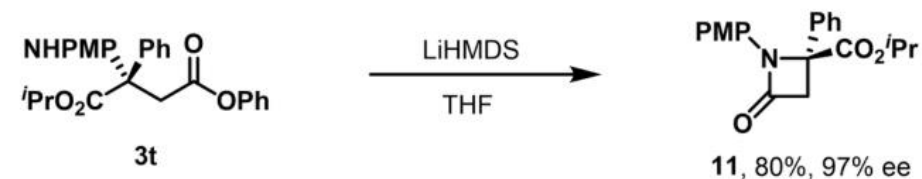
(a) *In-situ* reaction



(b) Gram-scale reaction



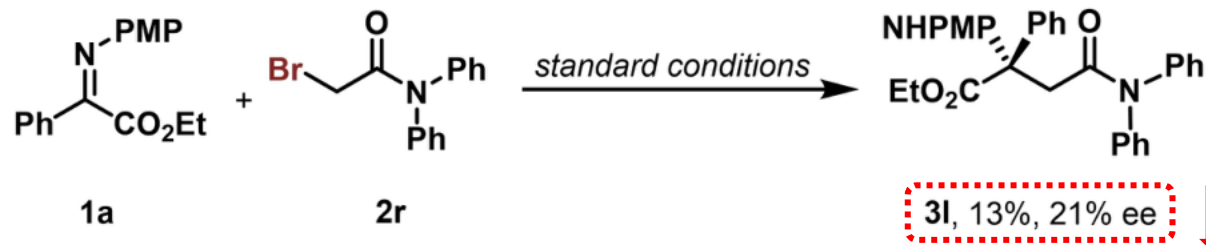
(c) Construction of chiral β -lactam



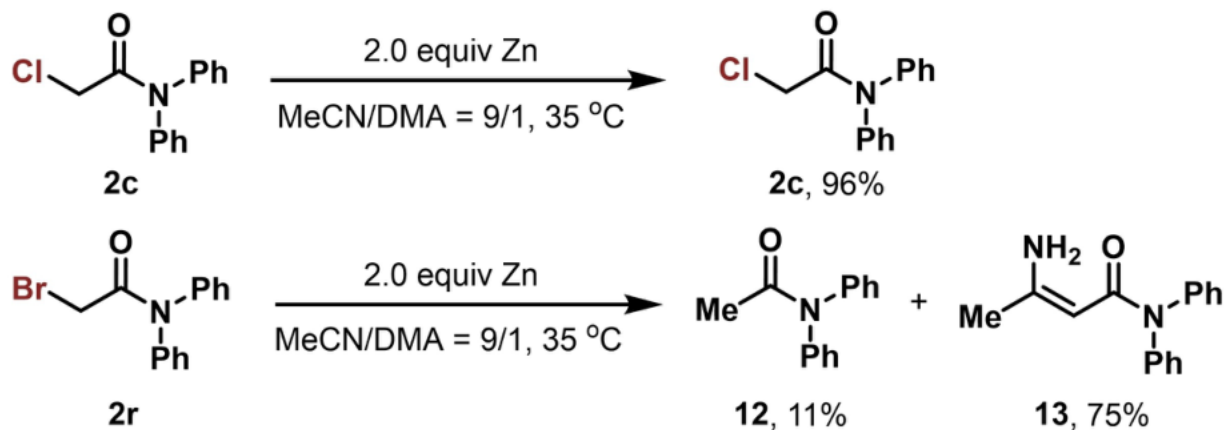
Scheme 4. Synthetic application.

2.2 C=N bonds

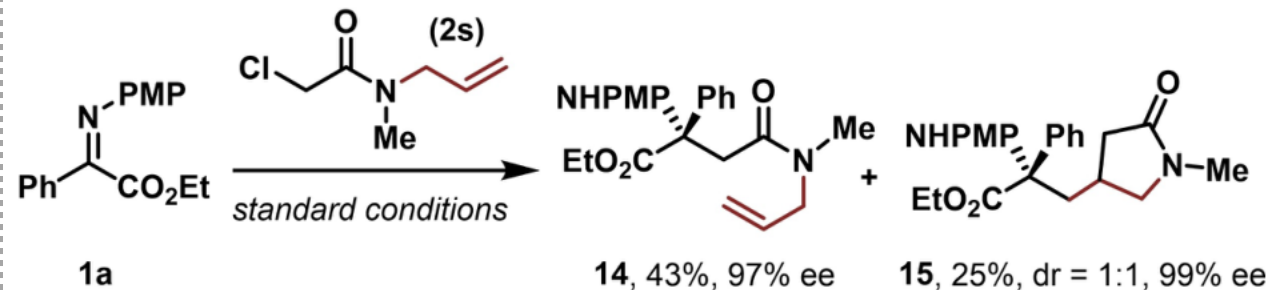
(a) Reaction with α -bromoamide as an electrophile



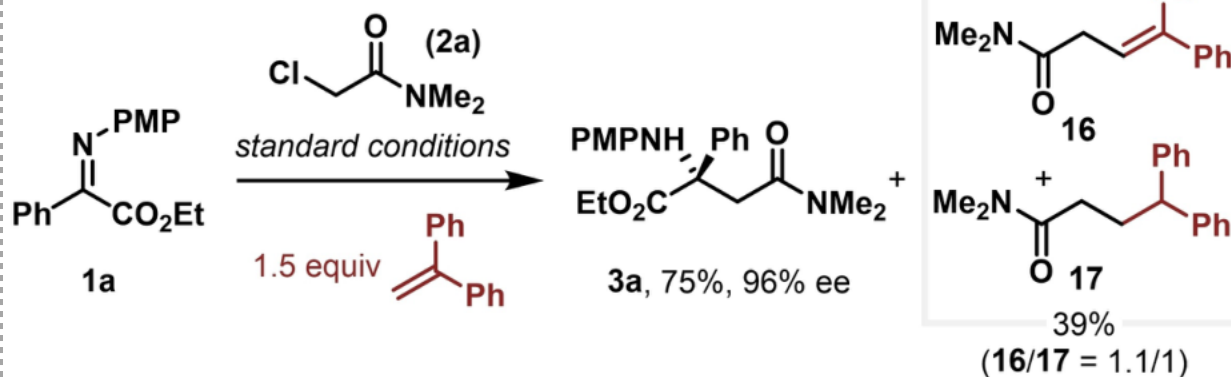
(b) Investigation of Reformatsky reagent formation



(c) Reaction with alkene substituted α -chloroamide



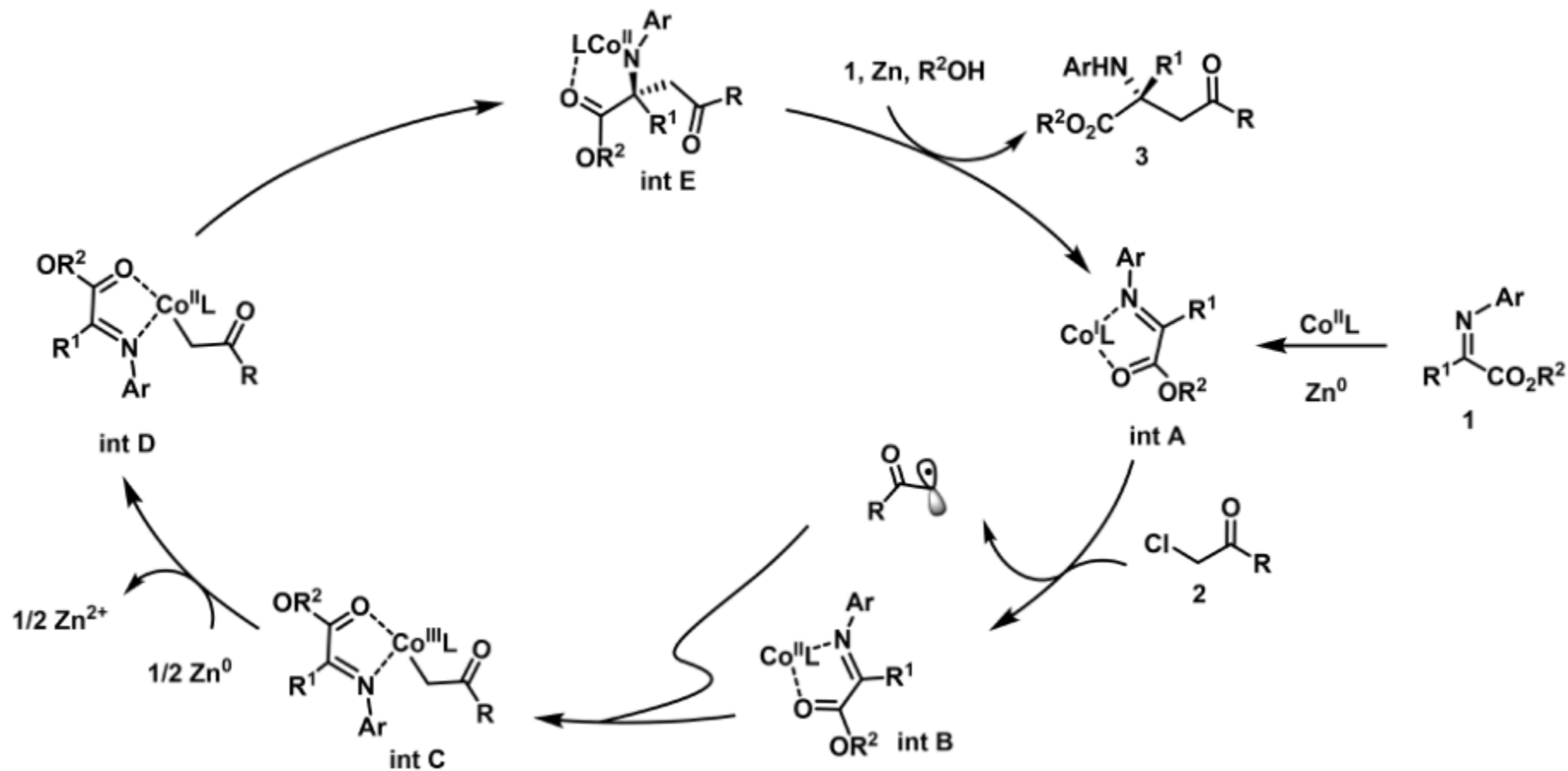
(d) Radical trapping reaction



还原自由基加成过程
和经典Reformatsky
加成界限的区别

2.2 C=N bonds

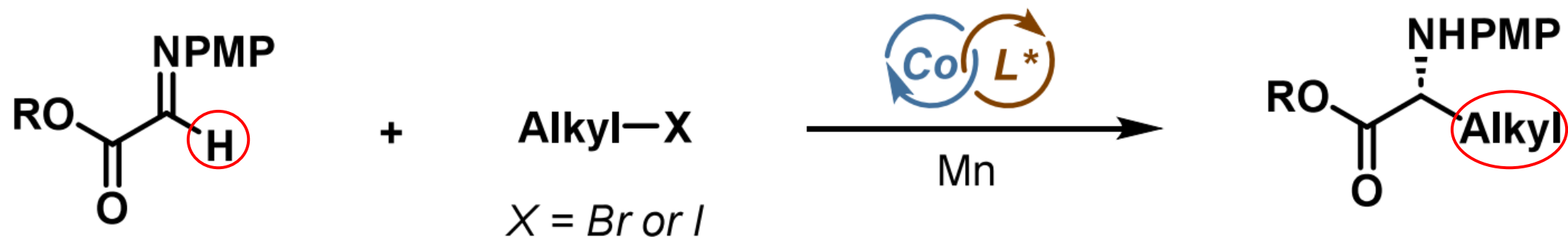
(e) Proposed mechanism



Scheme 5. Mechanistic studies.

2.2 C=N bonds

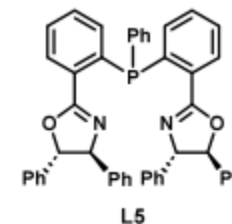
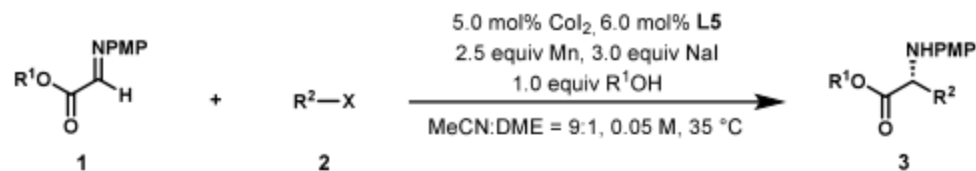
c) This work: cobalt-catalyzed asymmetric aza-Barbier alkylative addition



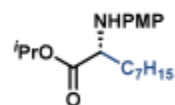
- ◆ *high functional group tolerance*
- ◆ *modular assembly of exotic α -AAs*

- ◆ *compatibility with 1°, 2°, and 3° alkyl halides*
- ◆ *good to excellent enantioselectivity and yield*

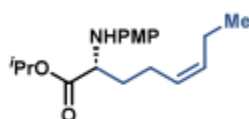
2.2 C=N bonds



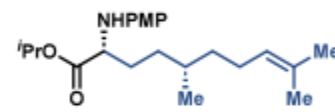
Primary Alkyl Iodide Scope



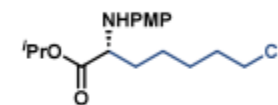
3a, 79%, 95% ee



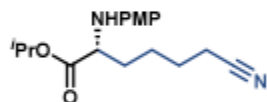
3b, 71%, 91% ee



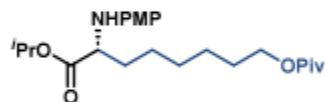
Citronellol analogue
3c, 76%, 96:4 dr



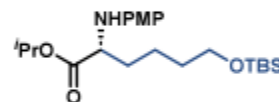
3d, 84%, 94% ee



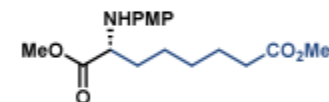
3e, 78%^b, 92% ee



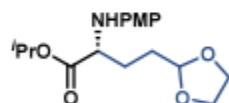
3f, 62%, 91% ee



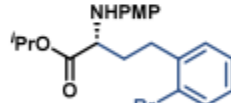
3g, 73%, 90% ee



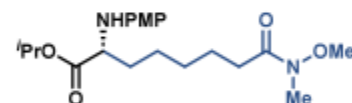
3h, 53%, 90% ee



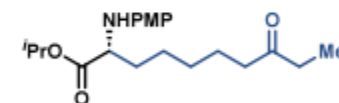
3i, 79%^b, 89% ee



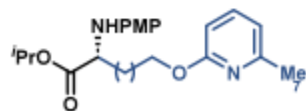
3j, 76%^b, 91% ee



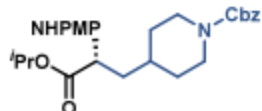
3k, 48%^b, 90% ee



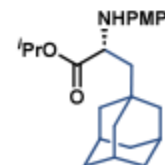
3l, 57%^b, 91% ee



3m, 56%^b, 90% ee



3n, 79%, 90% ee

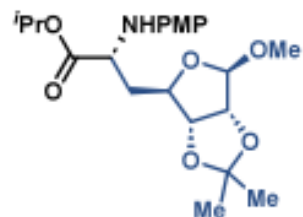


3o, 75%, 99% ee

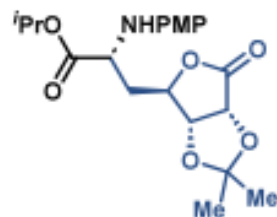


Gemfibrozil analogue
3p, 72%, 98% ee

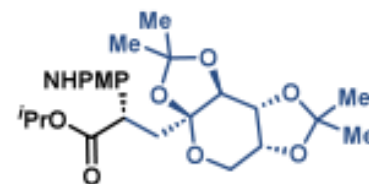
2.2 C=N bonds



Methyl-β-D-ribofuranoside analogue
3q, 63%, 98:2 dr

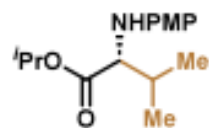


D-Ribonic-γ-lactone analogue
3r, 55%, 96:4 dr

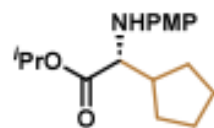


Diacetonefructose analogue
3s, 68%, 95:5 dr

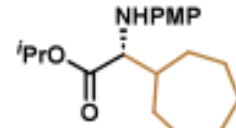
Secondary Alkyl Bromide Scope



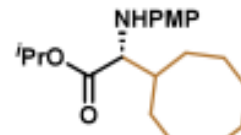
3t, 37%^b, 96% ee



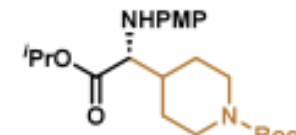
3u, 71%^b, 96% ee



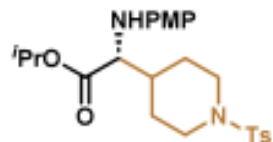
3v, 54%^b, 97% ee



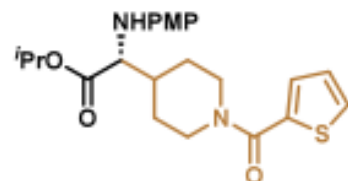
3w, 73%^b, 97% ee



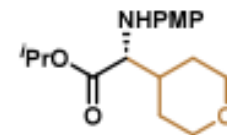
3x, 52%^b, 90% ee



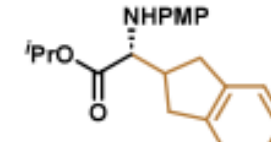
3y, 50%^b, 90% ee



3z, 66%^b, 90% ee

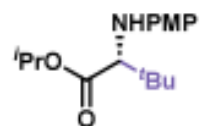


3aa, 44%^b, 93% ee

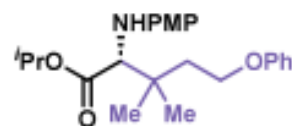


3ab, 64%^b, 91% ee

Tertiary Alkyl Bromide Scope



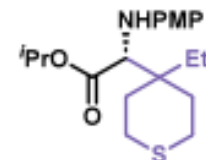
3ac, 58%^b, 92% ee



3ad, 79%^b, 94% ee

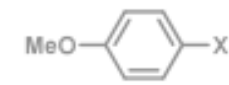


3ae, 80%^b, 94% ee



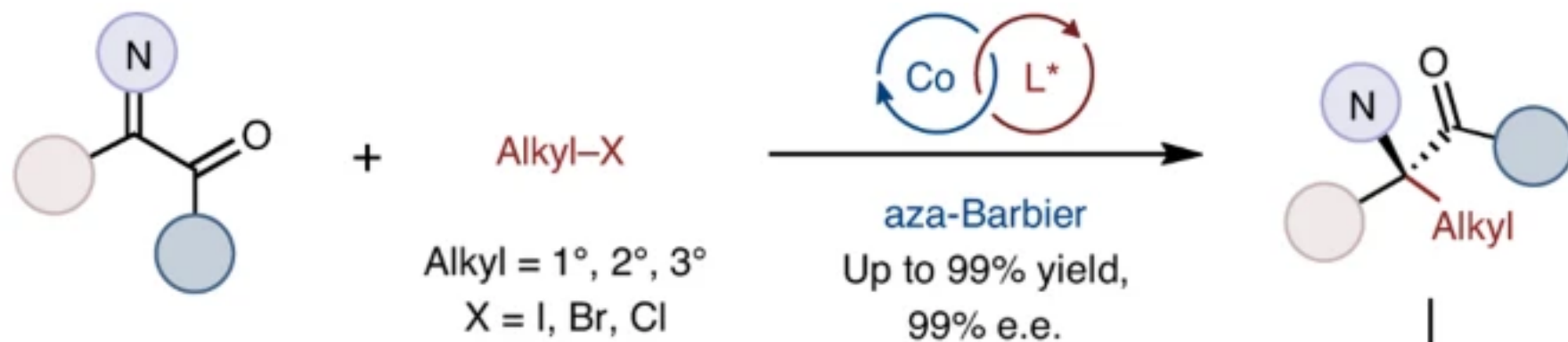
3af, 83%^b, 95% ee

Unsuccessful examples

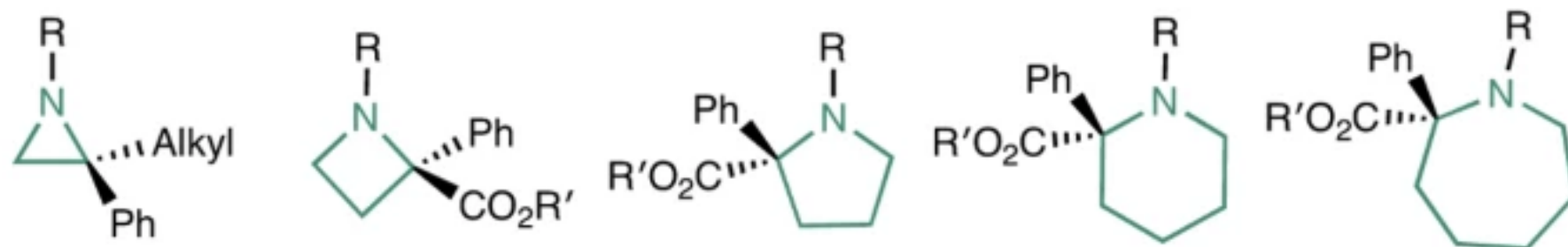


X = Br, I, OTf

2.2 C=N bonds



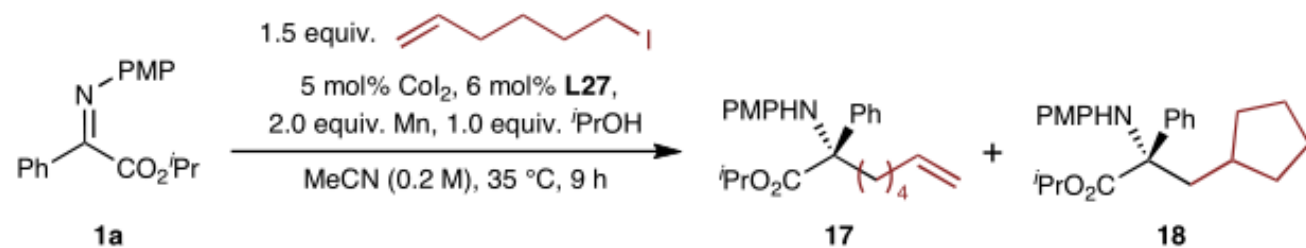
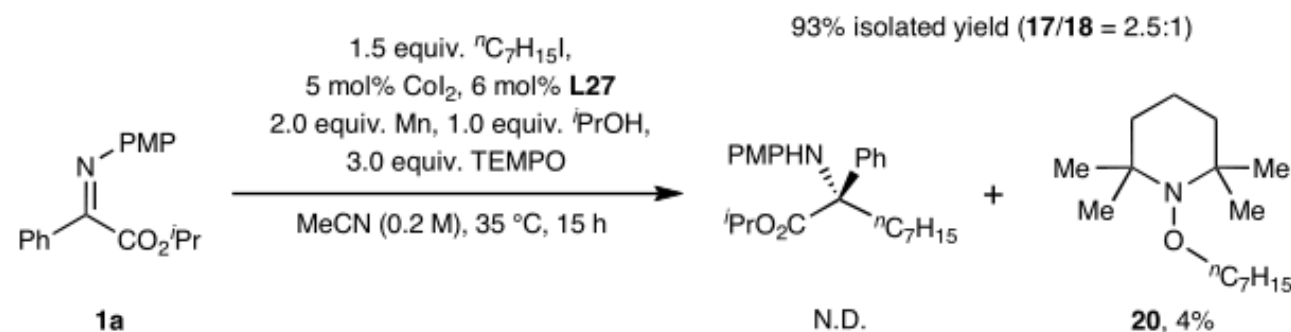
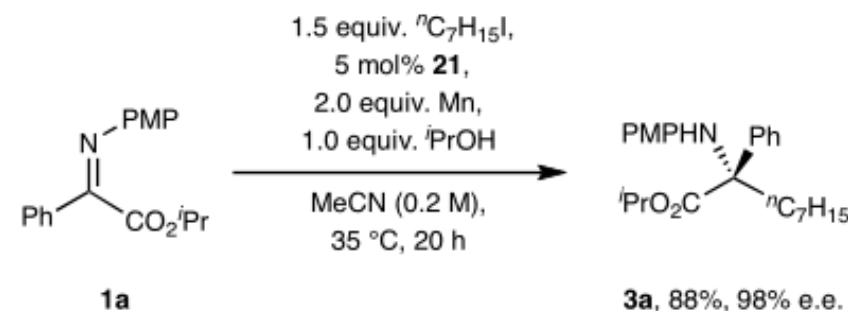
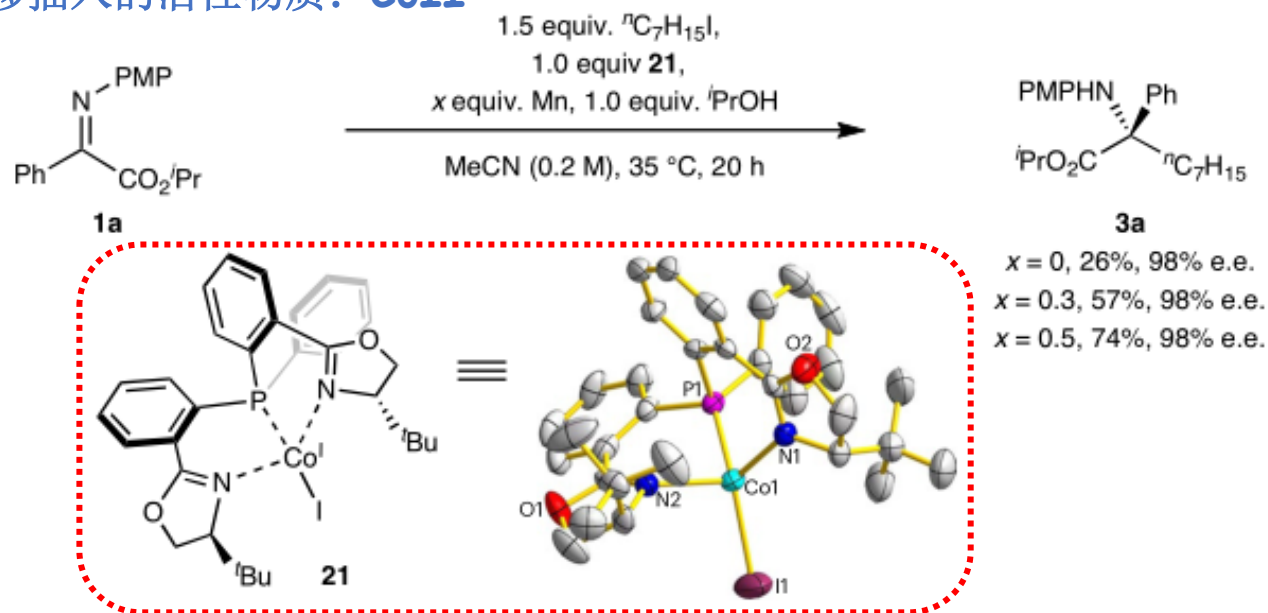
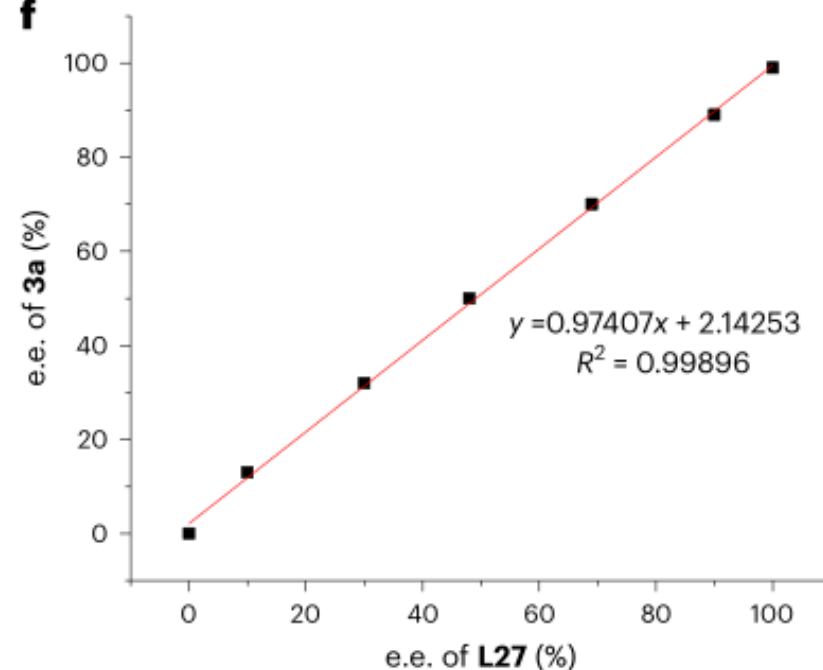
Modular chiral heterocycle synthesis



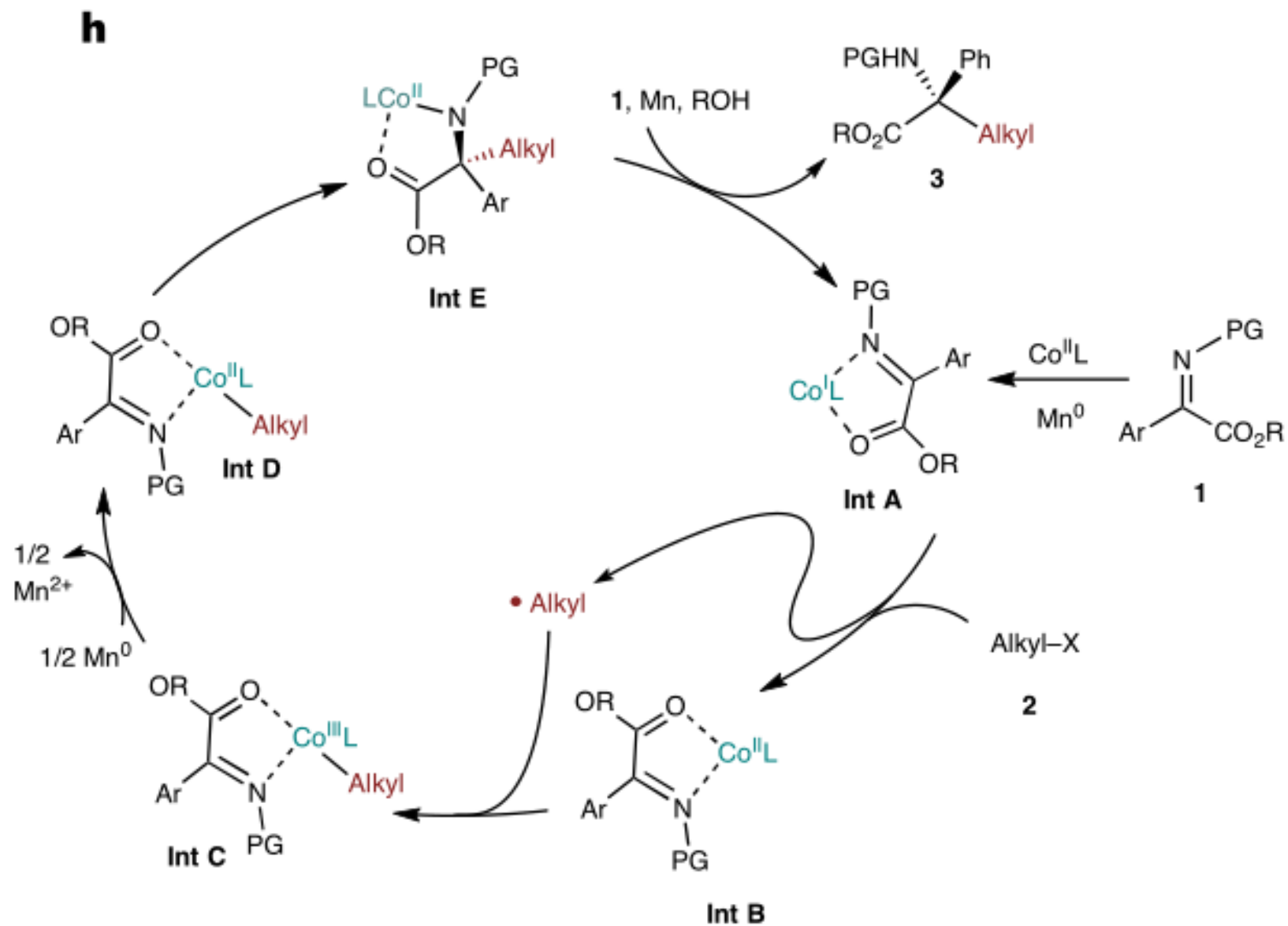
■ Facile construction of α -tertiary amino ester

■ Excellent functional group tolerance

Nat. Chem. **2024**, *16*, 398.

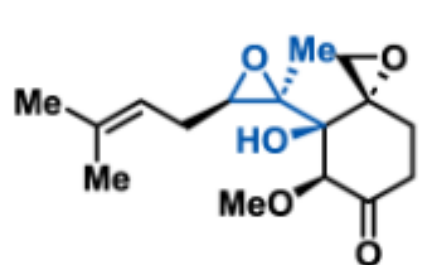
a**b****c****d** 反应中间体: **CoI**诱发反应**e** 迁移插入的活性物质: **CoII****f**

2.2 C=N bonds

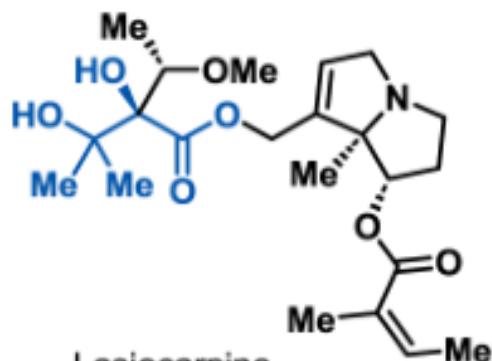


2.3 C=O bonds

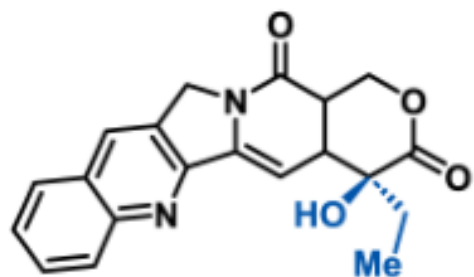
a The prevalence of α -tetrasubstituted chiral alcohol



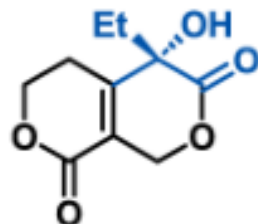
(-)-Ovalicin



Lasiocarpine

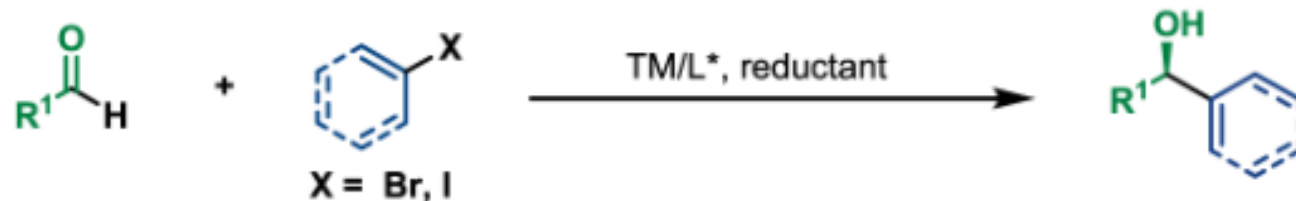


Camptothecin

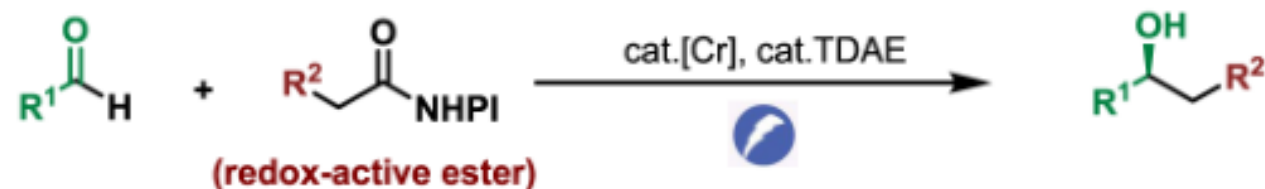


(+)-Gentiolactone

b Catalytic asymmetric addition of aldehyde with alkenyl or aryl halides



Electrocatalytic asymmetric Nozaki-Hiyama-Kishi reaction of aldehyde

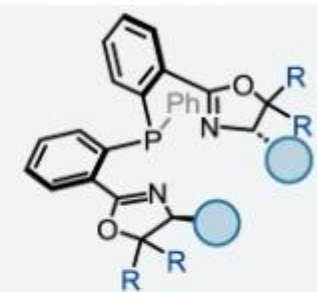


2.3 C=O bonds

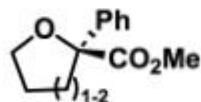
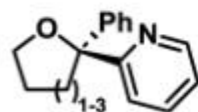
c This work



New NPN Ligand

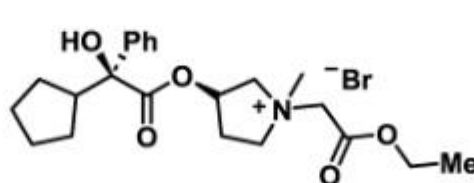


chiral heterocycles

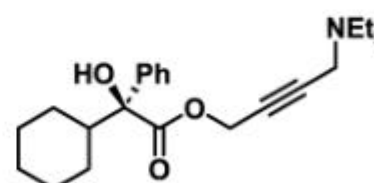


Diversified oxygen-containing heterocycles

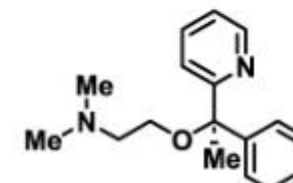
facile synthesis of chiral drugs



Sofdra (Sofpironium)



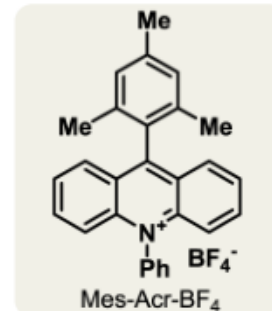
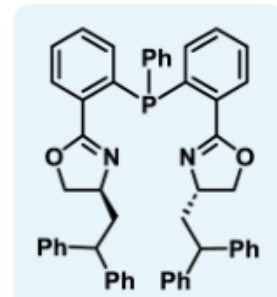
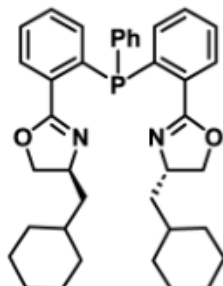
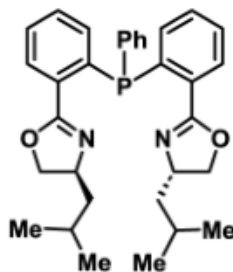
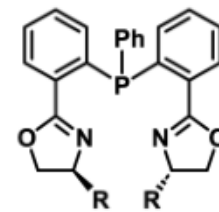
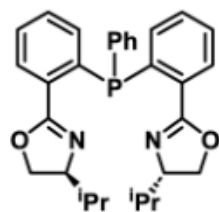
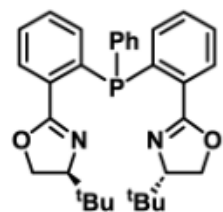
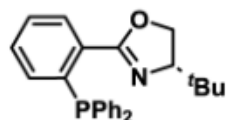
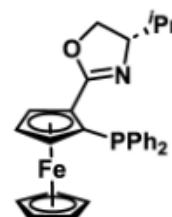
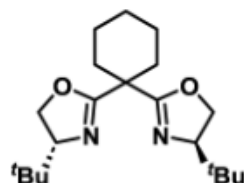
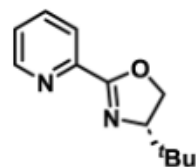
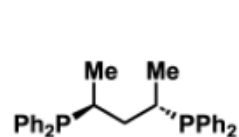
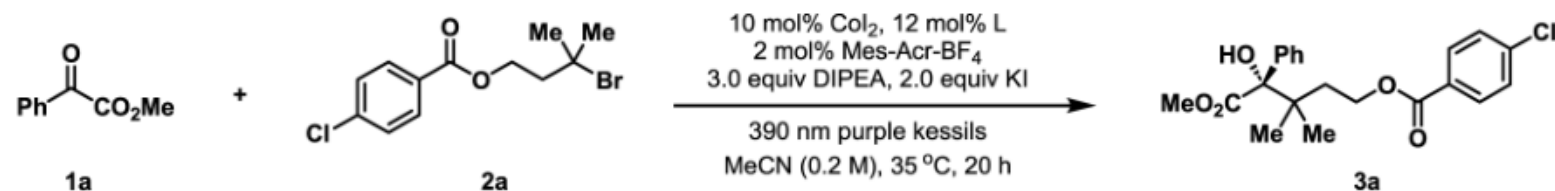
(S)-Oxybutynin



Doxylamine

2.3 C=O bonds

Table 1. Optimization of Reaction Conditions^{ab}



Variations from above

4CzIPN instead of Mes-Acr- BF_4

$\text{Ir}(\text{ppy})_2(\text{dtbbpy})\text{PF}_6$ instead of Mes-Acr- BF_4

$\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ instead of Mes-Acr- BF_4

465 nm blue LED instead of 390 nm purple LED
 without KI

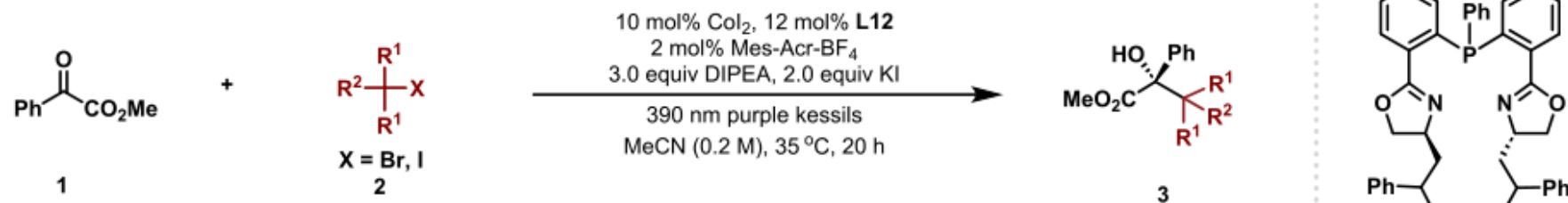
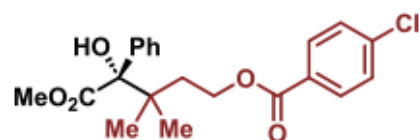
7.5 mol % CoI_2 , 9 mol % **L12**

alkyl iodide instead of alkyl bromide **2a**

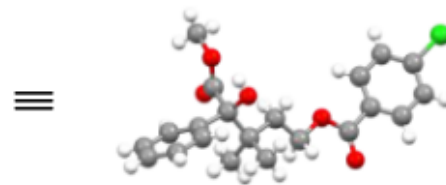
Entry	Yield (%)
1	78
2	68
3	69
4	19
5	20
6	70
7 ^c	82

烷基碘化物有相似的效果。

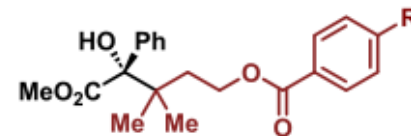
2.3 C=O bonds

Scheme 2. Substrate Scope of Unactivated Alkyl Halides^{ab}Tertiary alkyl bromide^a

3a, 85%, 96:4 er



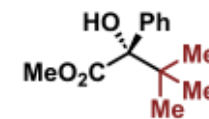
CCDC: 2326190



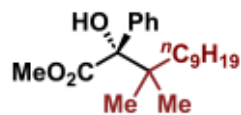
3b, R = CO₂Me, 94%, 96:4 er

3c, R = H, 56%, 95.5:4.5 er

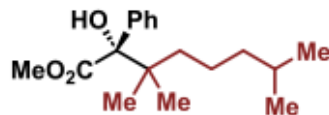
3d, R = OMe, 92%, 95:5 er



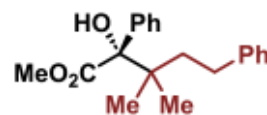
3e, 65%, 95:5 er



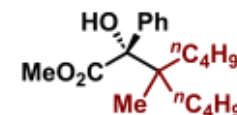
3f, 78%, 95:5 er



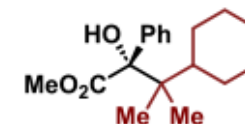
3g, 78%, 96.5:3.5 er



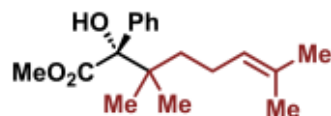
3h, 70%, 94:6 er



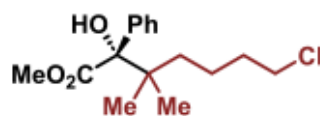
3i, 72%, 96.5:3.5 er



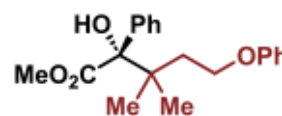
3j, 53%, 96:4 er



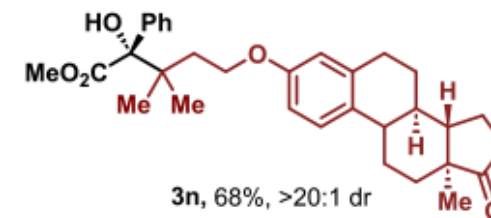
3k, 72%, 95.5:4.5 er



3l, 67%, 96:4 er

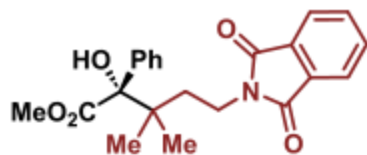


3m, 65%, 96:4 er

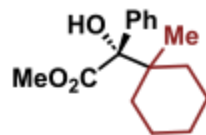


3n, 68%, >20:1 dr
from estrone

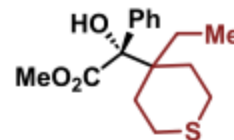
2.3 C=O bonds



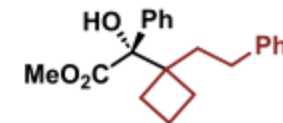
3o, 64%, 96.5:3.5 er



3p, 65%, 95.5:4.5 er

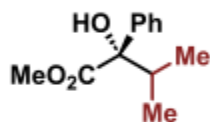


3q, 70%, 95.5:4.5 er

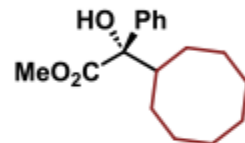


3r, 82%, 95.5:4.5 er

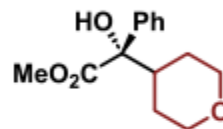
Secondary alkyl iodides^b



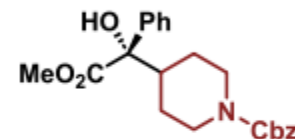
3s, 96%, 93:7 er



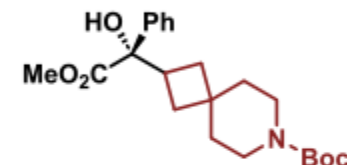
3t, 84%, 92:8 er



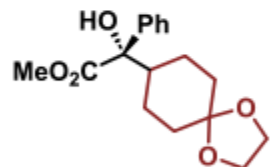
3u, 88%, 93.5:6.5 er



3v, 78%, 94.4:5.5 er

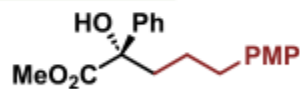


3w, 82%, 92.5:7.5 er

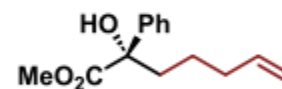


3x, 78%, 91.5:8.5 er

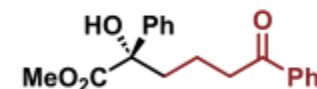
Primary alkyl iodides^b



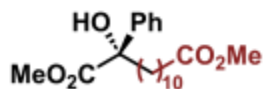
3y, 82%, 93.5:6.5 er



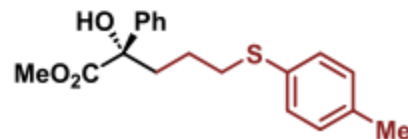
3z, 89%, 93:7 er



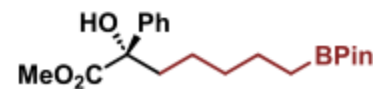
3aa, 74%, 96:4 er



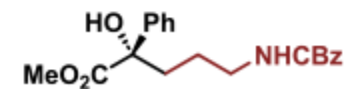
3ab, 78%, 92:8 er



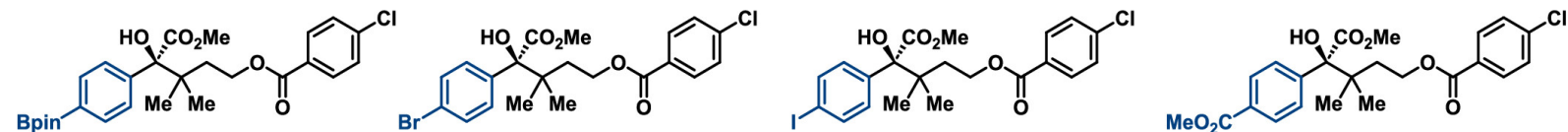
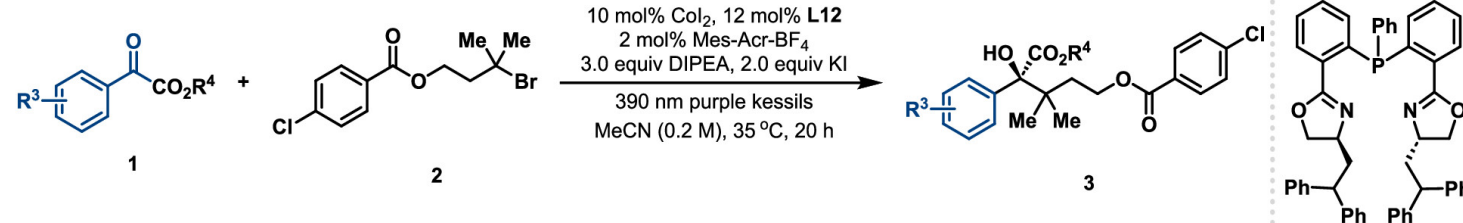
3ac, 71%, 91:9 er



3ad, 76%, 92:8 er



3ae, 63%, 92:8 er

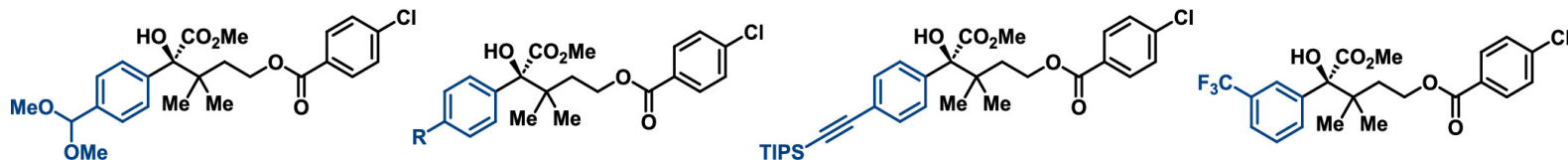


3af, 83%, 96:4 er

3ag, 76%, 95:5 er

3ah, 89%, 95:5 er

3ai, 85%, 94:6 er

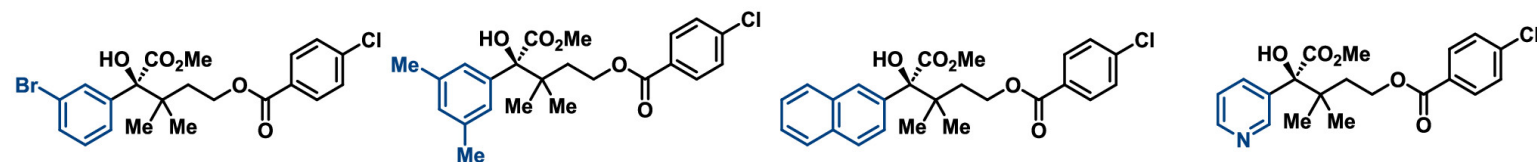


3aj, 82%, 95.5:4.5 er

3ak, R = TMS, 81%, 95:5 er
3al, R = SMe, 87%, 95.5:4.5 er

3am, 87%, 97:3 er

3an, 61%, 95:5 er

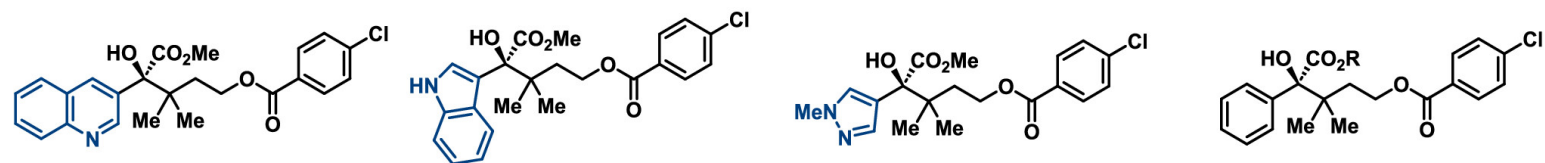


3ao, 54%, 95:5 er

3ap, 78%, 96:4 er

3aq, 89%, 95:5 er

3ar, 75%, 95:5 er



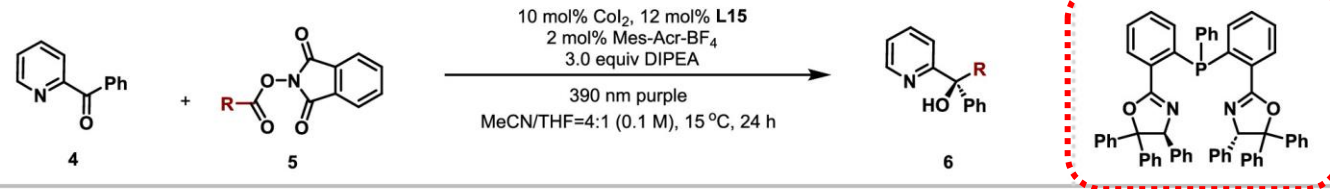
3as, 53%, 93.5:6.5 er

3at, 65%, 96.5:3.5 er

3au, 71%, 95.5:4.5 er

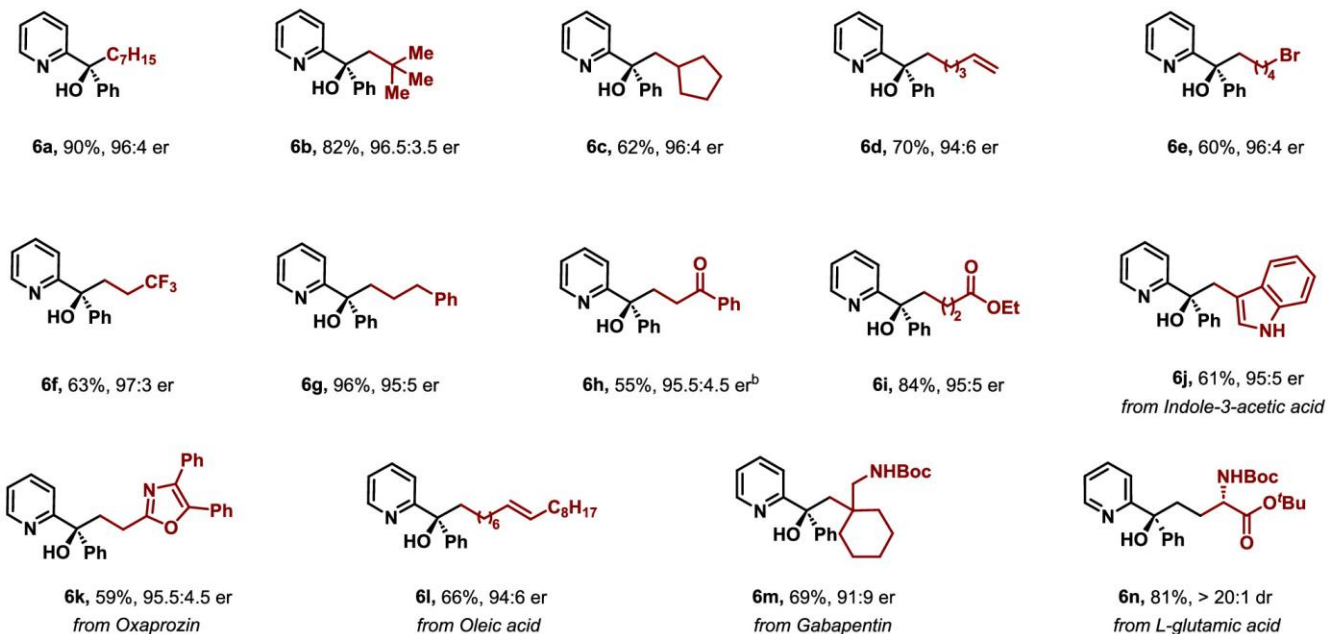
3av, R = ⁱPr, 83%, 90.5:9.5 er

3aw, R = Bn, 74%, 94:6 er

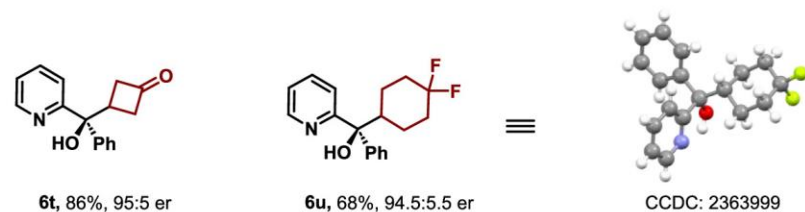


Primary alkyl NHPI esters

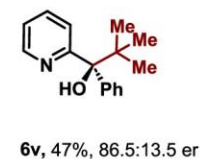
简单更换配体和溶剂



Secondary alkyl NHPI esters

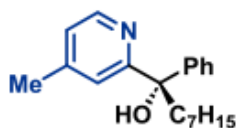
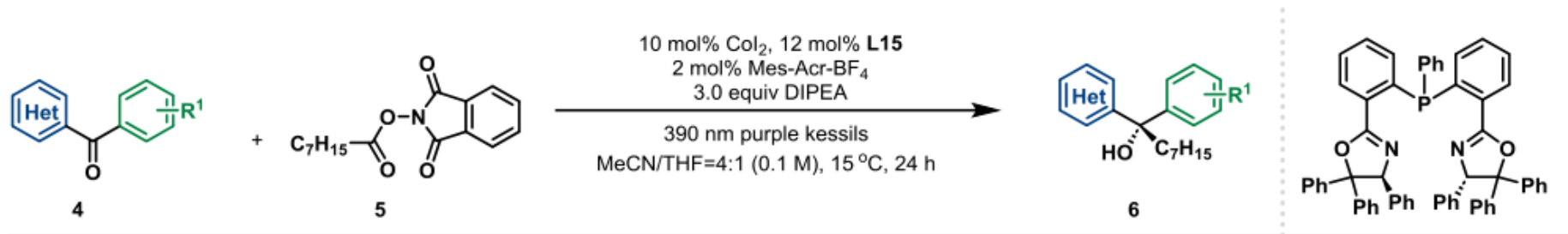


Tertiary alkyl NHPI esters

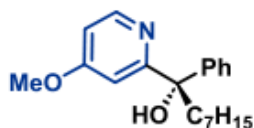


2.3 C=O bonds

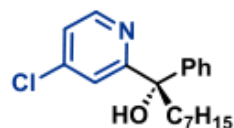
Scheme 5. Substrate Scope of Pyridine-Based Ketones^a



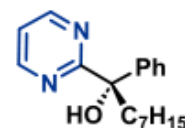
6w, 67%, 96:4 er



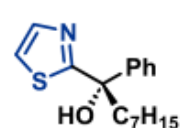
6x, 81%, 95:5 er



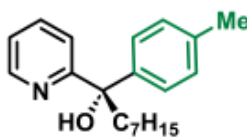
6y, 67%, 97:3 er



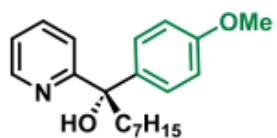
6z, 63%, 97:3 er



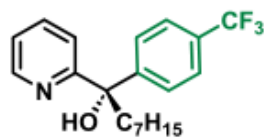
6aa, 62%, 95.5:4.5 er



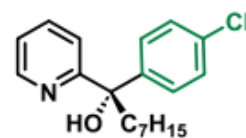
6ab, 79%, 95:5 er



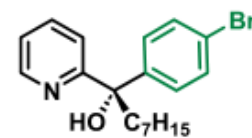
6ac, 74%, 95.5:4.5 er



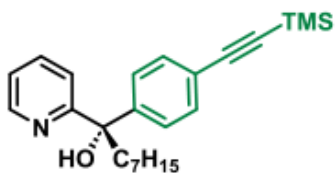
6ad, 51%, 93:7 er



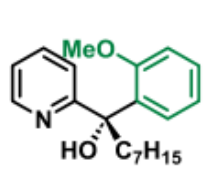
6ae, 46%, 95:5 er



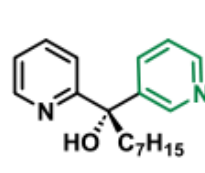
6af, 47%, 94.5:5.5 er



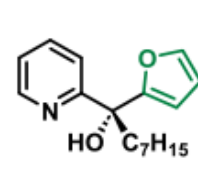
6ag, 53%, 94:6 er



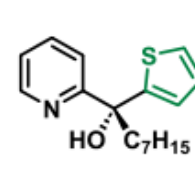
6ah, 46%, 84:16 er



6ai, 68%, 92.5:7.5 er



6aj, 47%, 96:4 er

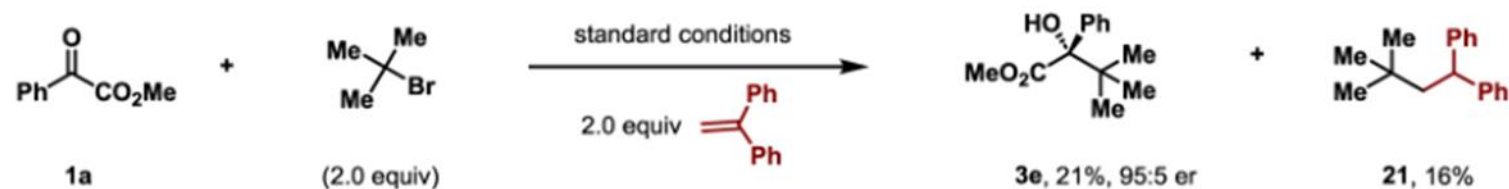


6ak, 76%, 95:5 er

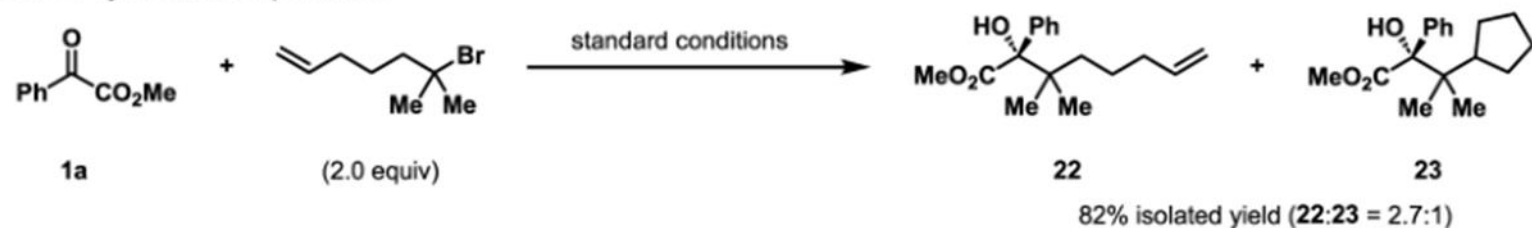
2.3 C=O bonds

Scheme 7. Preliminary Mechanistic Studies

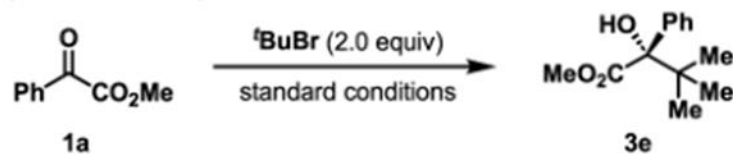
(a) Radical trap reaction



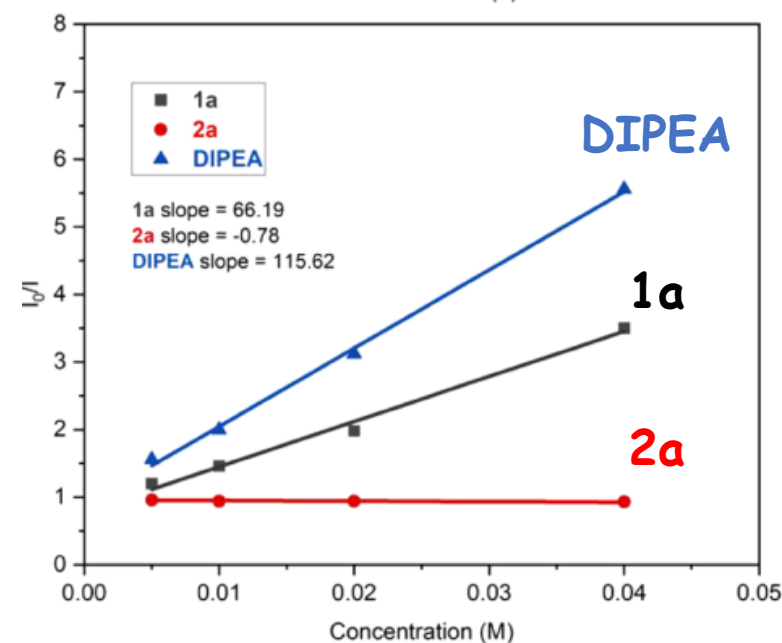
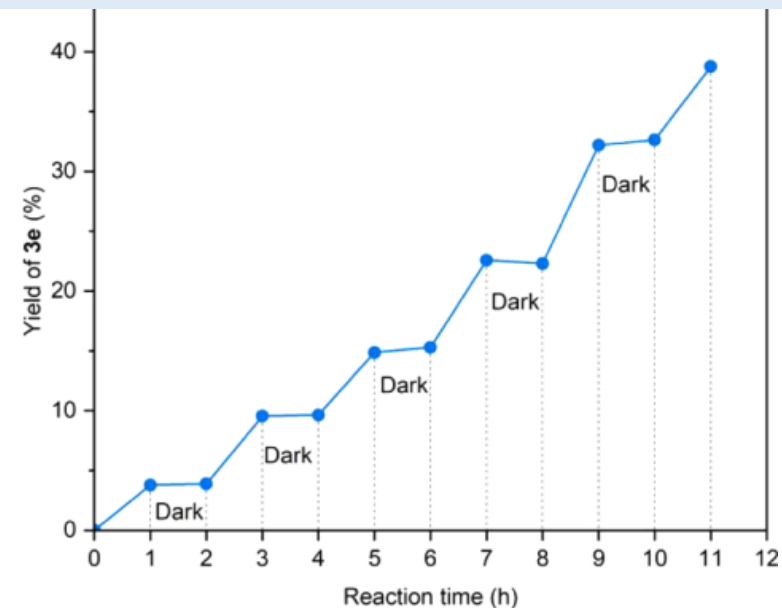
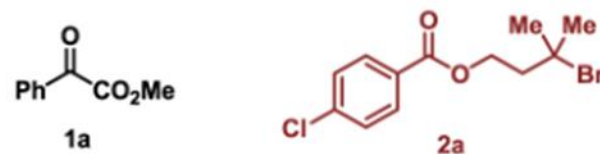
(b) Radical cyclization experiment



(c) Light on/off study

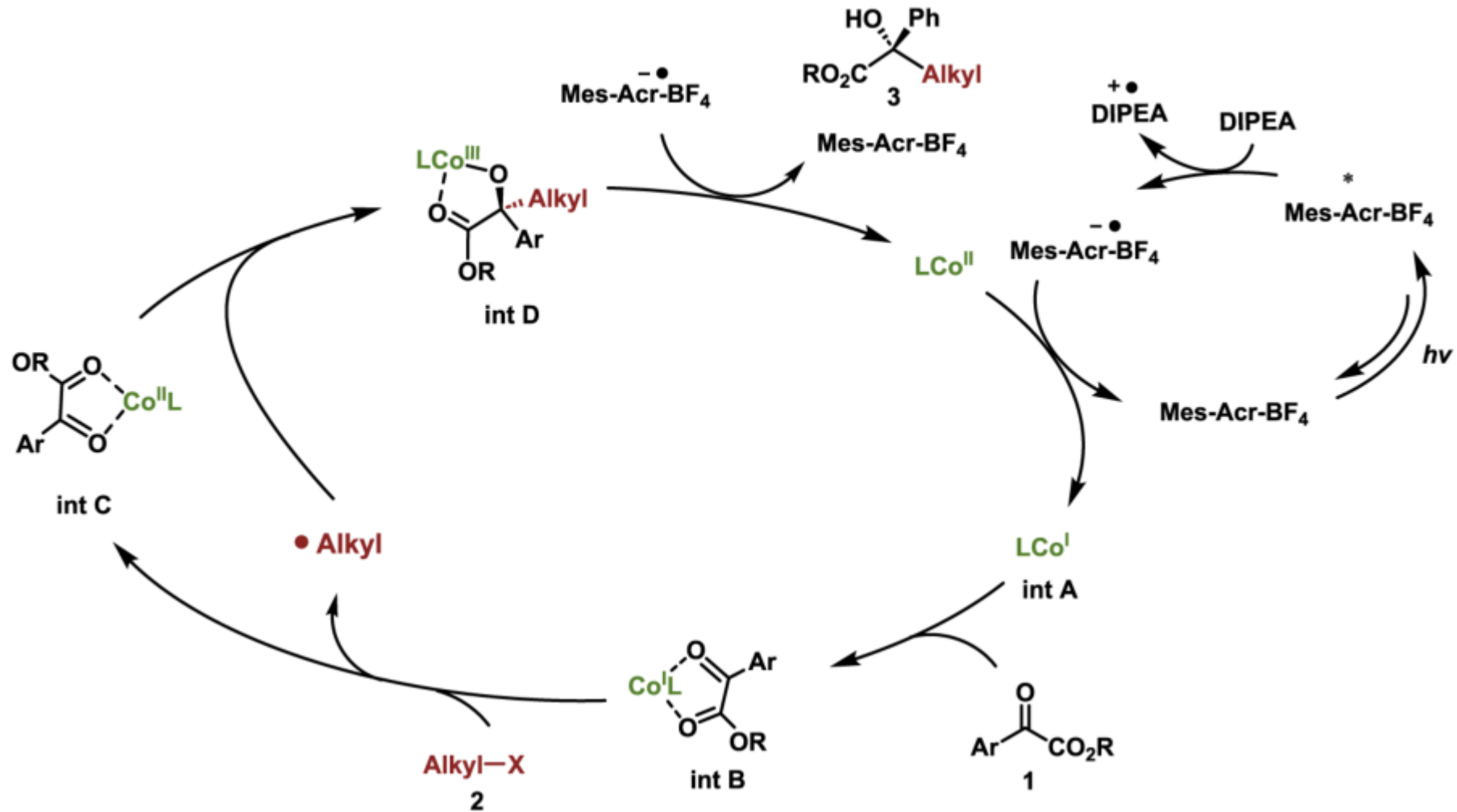


(d) The Stern–Volmer plots



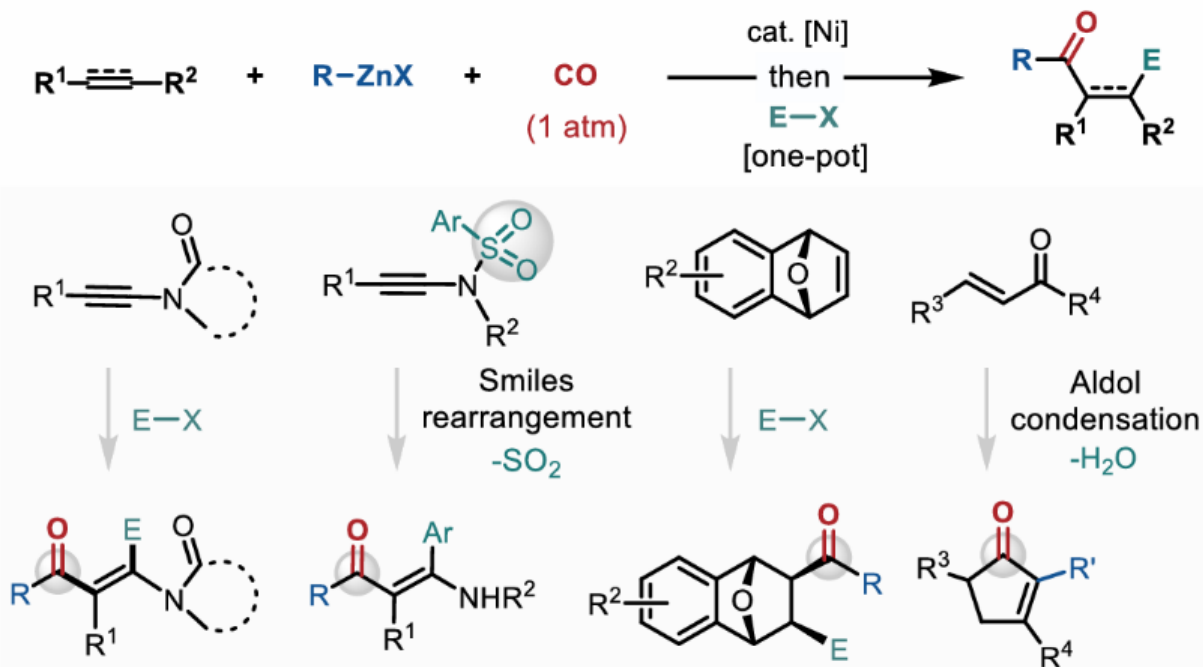
2.3 C=O bonds

(e) Proposed mechanism



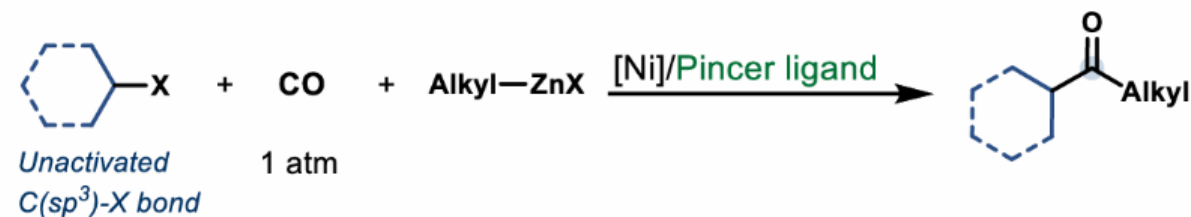
3. Catalytic Carbonylation of 1 atm CO by Nickel

Acylzincation of unsaturated C-C bond

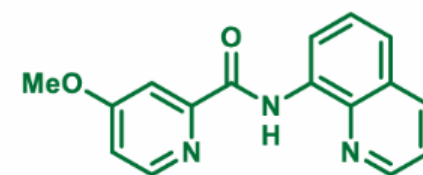
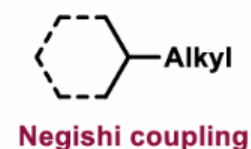


- *in situ* generation of catalytic acyl organometallics
- a new paradigm of carbometallation chemistry

Carbonylative cross-coupling reaction



Pitfall



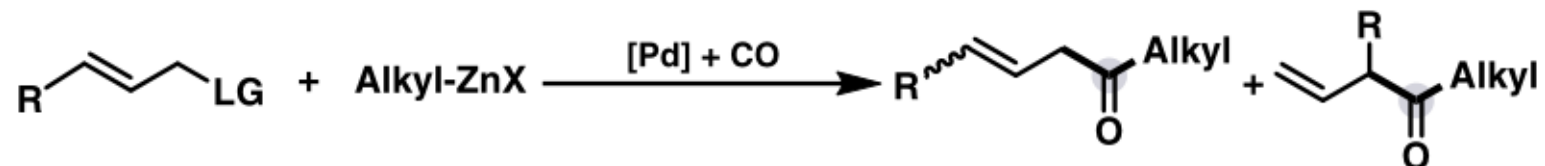
newly developed NN_2 -pincer ligand

the chemoselective three-component carbonylative Negishi cross-coupling reaction.

overcoming the competing Negishi coupling, the β -hydride elimination, dehalogenation

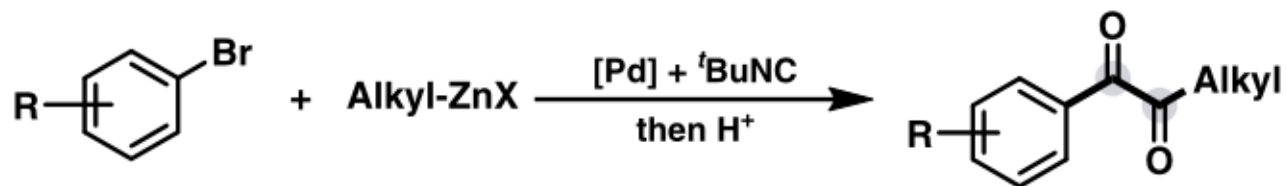
3. Catalytic Carbonylation of 1 atm CO by Nickel

a Pd-catalyzed allylic carbonylative Negishi reaction with CO (Tamaru)



■ Poor regioselectivity

b Pd-catalyzed aryl carbonylative Negishi reaction (Dechert–Schmitt and Blackmond)

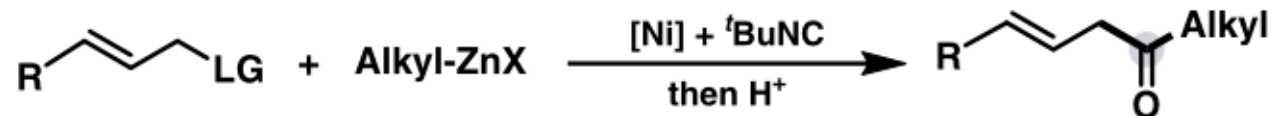


■ Double insertion of ^tBuNC

■ Challenge for monocarbonylation

Nat. Commun. **2020**, *11*, 392.

c Ni-catalyzed allylic carbonylative Negishi reaction with ^tBuNC (this work)



■ High regioselectivity

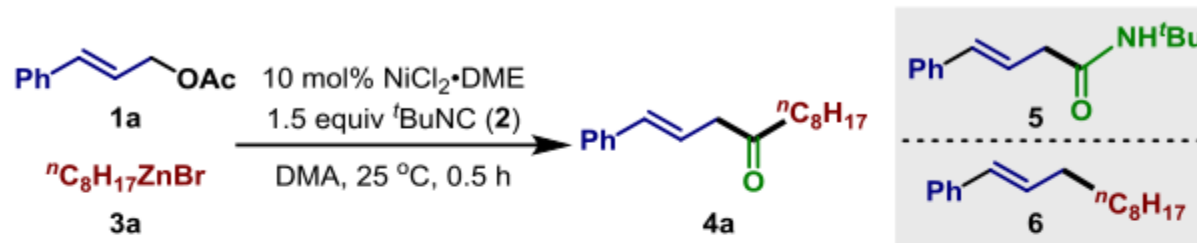
■ No polycarbonylation of isocyanide

■ High *E/Z* selectivity

■ No alkene isomerization

3. Catalytic Carbonylation of 1 atm CO by Nickel

Table 1 Optimization of the reaction conditions.



Entry ^a	Catalyst	Ligand	Yield (%) ^b		
			4a	5	6
1	$\text{NiCl}_2\cdot\text{DME}$	—	90 (86) ^c	ND	1
2	$\text{NiCl}_2\cdot\text{DME}$	PPh_3	27	10	ND
3	$\text{NiCl}_2\cdot\text{DME}$	dppf	5	ND	ND
4	$\text{NiCl}_2\cdot\text{DME}$	Phen	82	3	3
5	$\text{NiCl}_2\cdot\text{DME}$	$\text{SIPr}\cdot\text{HCl}$	87	2	3
6	$\text{Ni}(\text{cod})_2$	—	86	6	ND
7	$\text{Pd}(\text{dba})_2$	—	4	ND	ND
8	—	—	ND	ND	ND
9 ^d	$\text{NiCl}_2\cdot\text{DME}$	—	ND	ND	ND

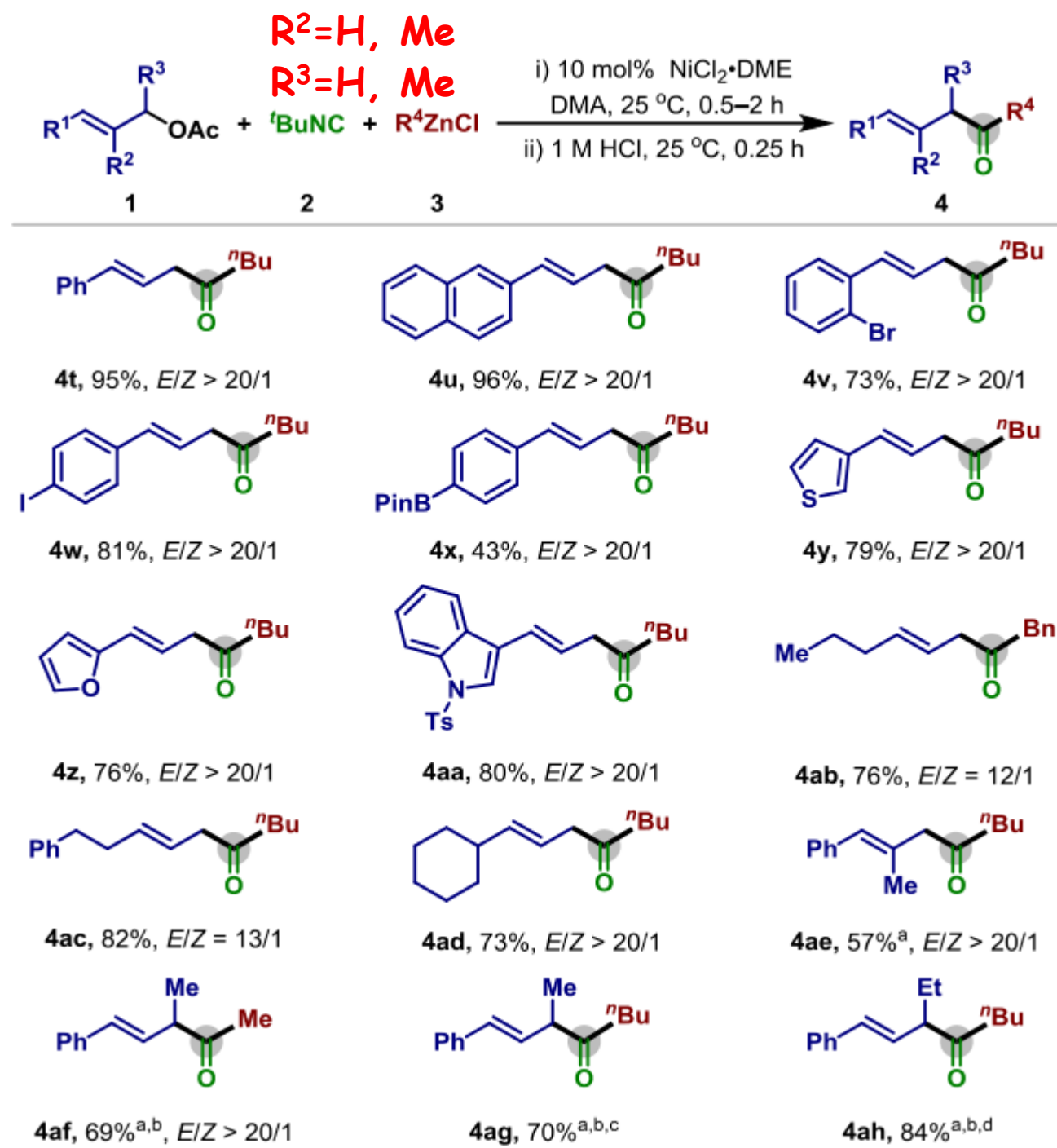
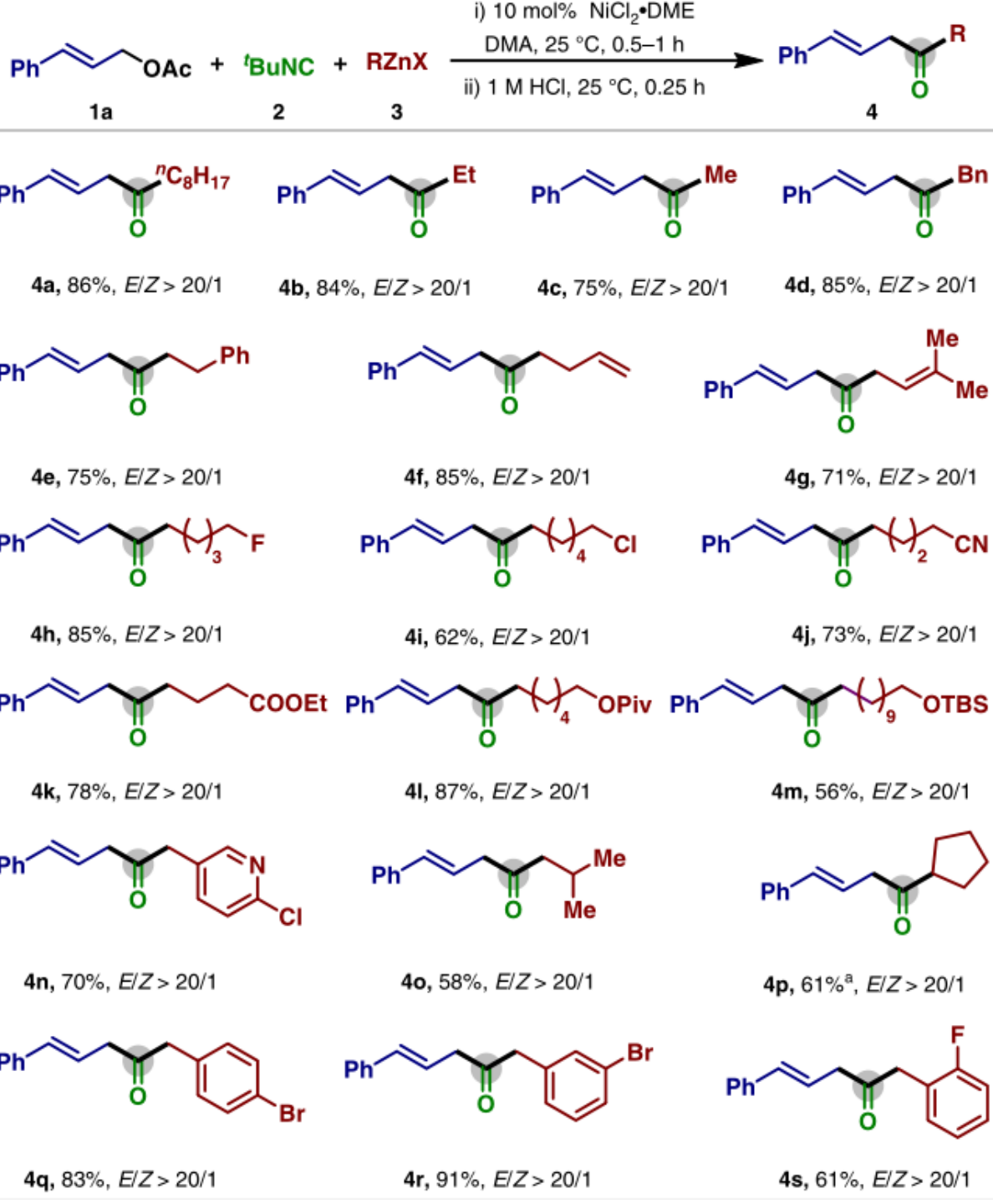
dppf 1,1'-Ferrocenediyl-bis(diphenylphosphine), $\text{SIPr}\cdot\text{HCl}$ 1,3-Bis-(2,6-diisopropylphenyl)imidazolium chloride, ND not determined

^aReaction conditions: **1a** (0.2 mmol), **2** (0.3 mmol), **3a** (0.3 mmol), catalyst (0.02 mmol), ligand (0.04 mmol), DMA (2 mL), 25 °C, 0.5 h, and then 1 M HCl (2 mL) was added, 25 °C, 0.25 h

^bCorrected GC yield

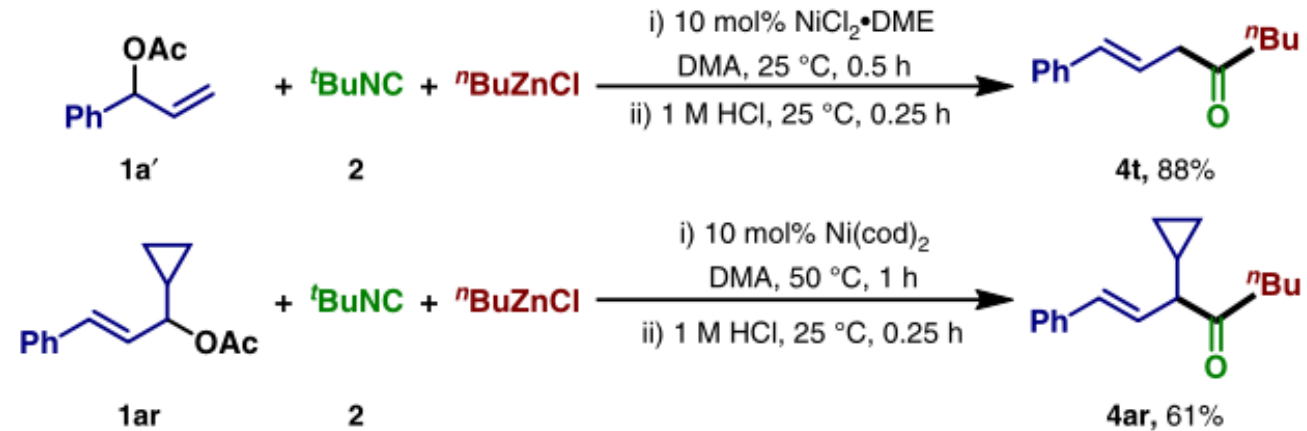
^cIsolated yield within parentheses in 0.2 mmol scale

^dThe reaction in CO (1 atm) instead of $t\text{BuNC}$



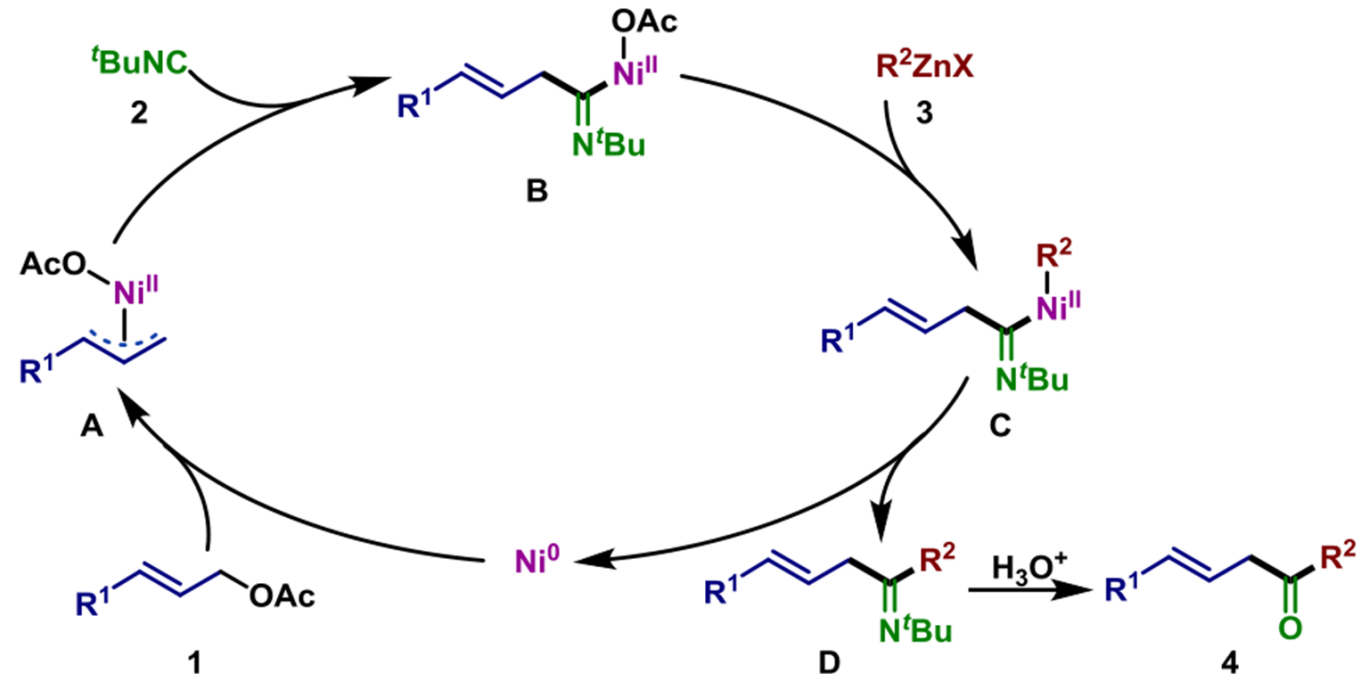
3. Catalytic Carbonylation of 1 atm CO by Nickel

a The reaction proceeds with π -allyl nickel intermediate



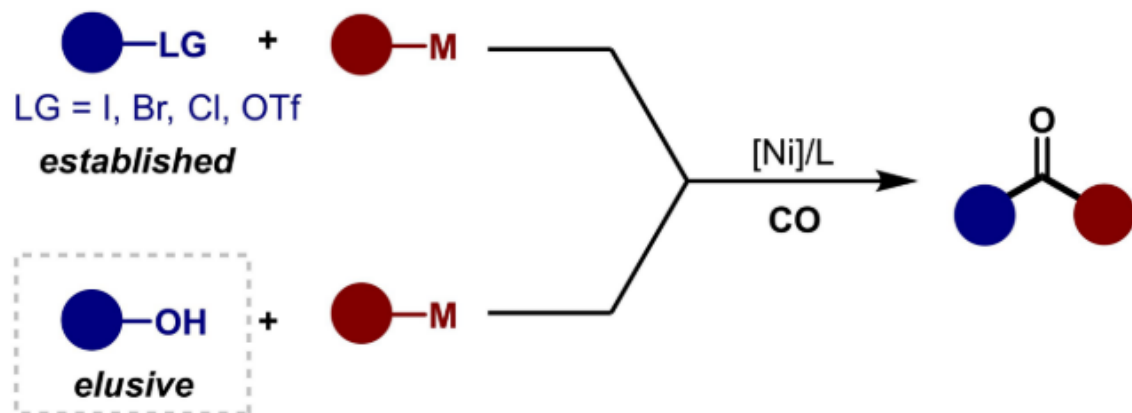
π 烯丙基镍中间体进行

自由基中间体



3. Catalytic Carbonylation of 1 atm CO by Nickel

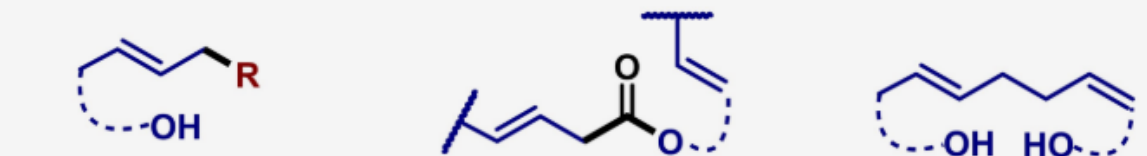
a) Ni-catalyzed carbonylative cross-coupling reactions



b) This work: Ni-catalyzed carbonylative reaction of allylic alcohols

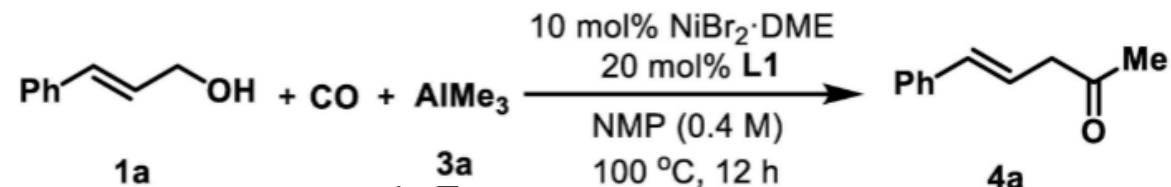


Potential side reaction

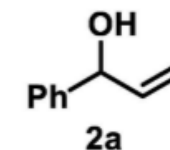
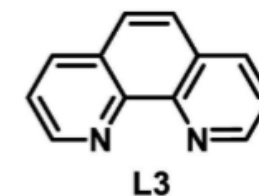
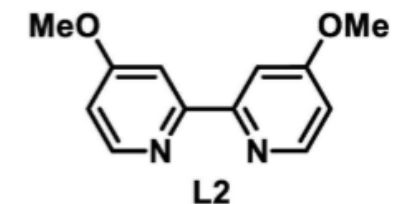
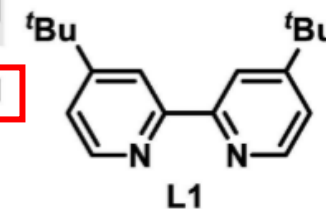


direct allylic coupling alkoxy carbonylation homodimerization

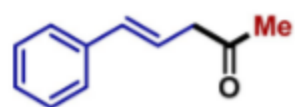
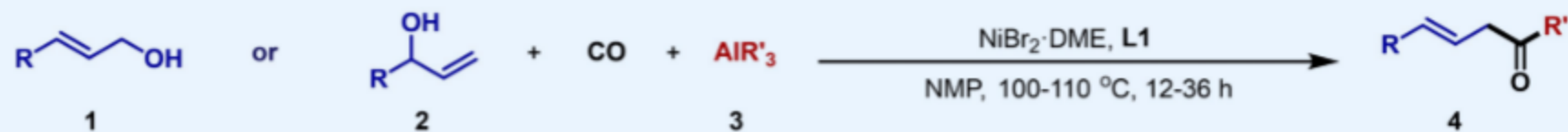
- Two C-C bonds formation
- chemoselective carbonylation
- Dual roles of organoalanes
- Investigation of reactive ADMAL



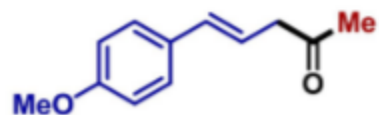
Entry	Deviation from above	4a (%) ^[b]
1	none	74 (72) ^[c]
2	NiCl ₂ ·DME instead of NiBr ₂ ·DME	65
3	Ni(COD) ₂ instead of NiBr ₂ ·DME	37
4	L2 instead of L1	62
5	L3 instead of L1	61
6	DMF as solvent	64
7	THF as solvent	0
8	Toluene as solvent	10
9	NMP (0.1 M)	57
10	NMP (1.0 M)	71
11	2.0 equiv AlMe ₃	72
12	MeZnCl, MeMgBr, MeLi instead of AlMe ₃	0
13	2a instead of 1a	74 (72) ^[c]



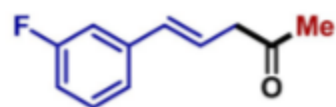
a) Carbonylative cross-coupling of allylic alcohols



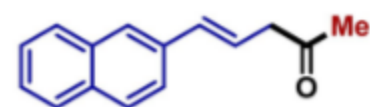
4a, 72%^[a,b]



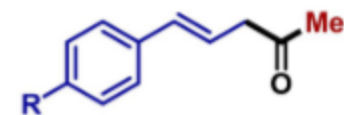
4b, 69%^[a] (64%^[b])



4c, 51%^[a] (68%^[b])



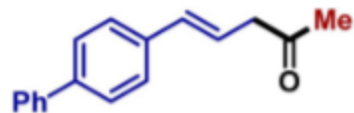
4d, 66%^[a,b]



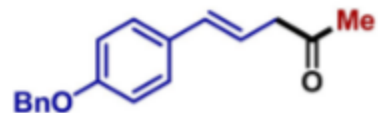
4e, R = Me, 69%^[a]

4f, R = *i*Pr, 50%^[b]

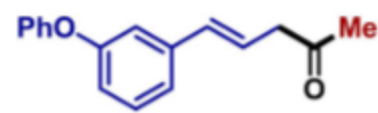
4g, R = *t*Bu, 70%^[b]



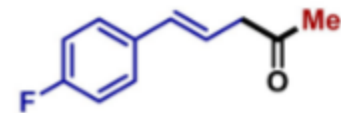
4h, 59%^[b]



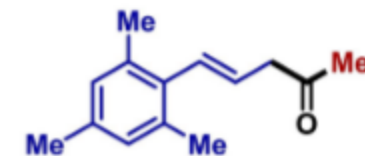
4i, 78%^[b]



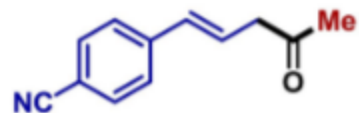
4j, 72%^[b]



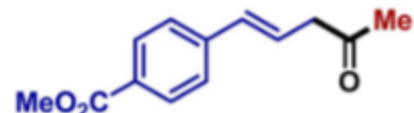
4k, 70%^[b]



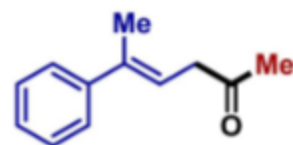
4l, 65%^[b]



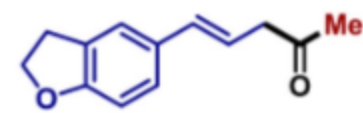
4m, 57%^[b]



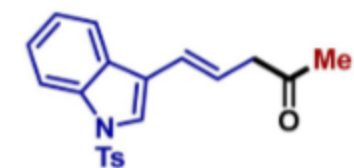
4n, 66%^[b]



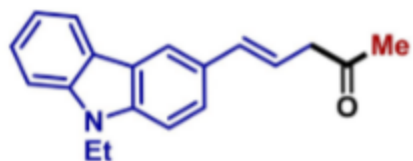
4o, 63%^[a]



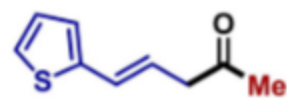
4p, 47%^[b]



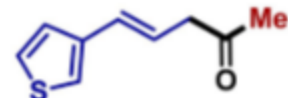
4q, 57%^[a]



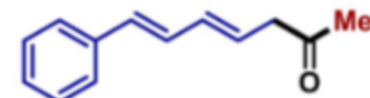
4r, 41%^[b]



4s, 43%^[b]



4t, 53%^[a]



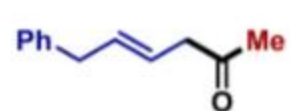
4u, 44%^[b]

E/Z = 5/1



4v, 49%^[c]

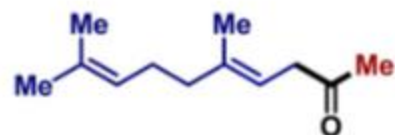
E/Z = 2.6/1



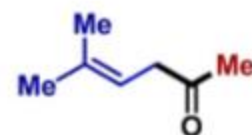
4w, 62%^[c]
E/Z = 2.8/1



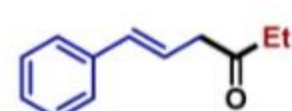
4x, 46%^[c]
E/Z = 2/1



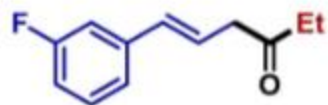
4y, 53%^[c]
E/Z = 2.1/1



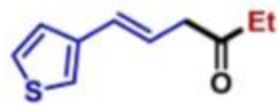
4z, 59%^[c]



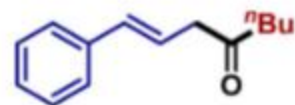
4aa, 75%^[a]



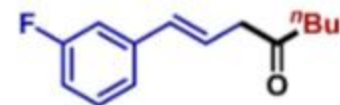
4ab, 47%^[a]



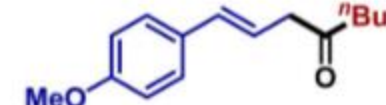
4ac, 62%^[a]



4ad, 51%^[a]

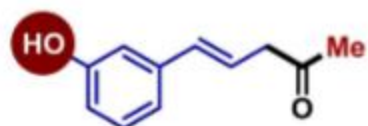


4ae, 52%^[a]

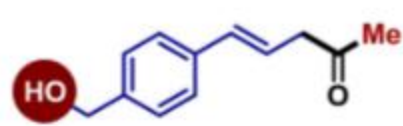


4af, 44%^[a]

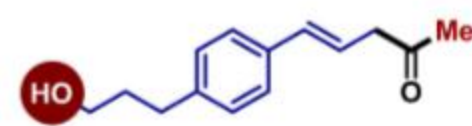
b) Chemoselective carbonylative cross-coupling of diols



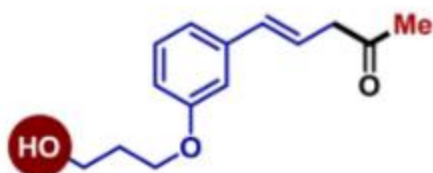
4ag, 72%^[b]



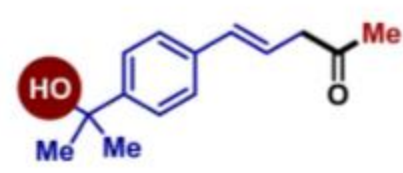
4ah, 49%^[b]



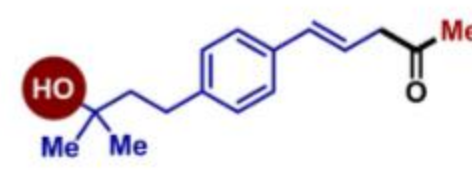
4ai, 66%^[b]



4aj, 68%^[b]

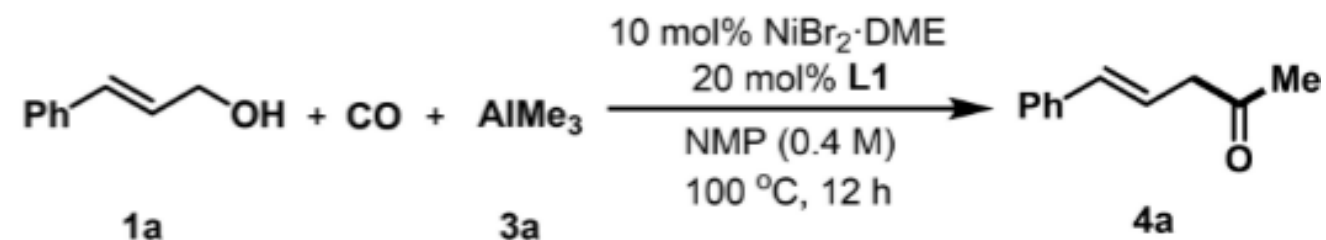


4ak, 45%^[b]

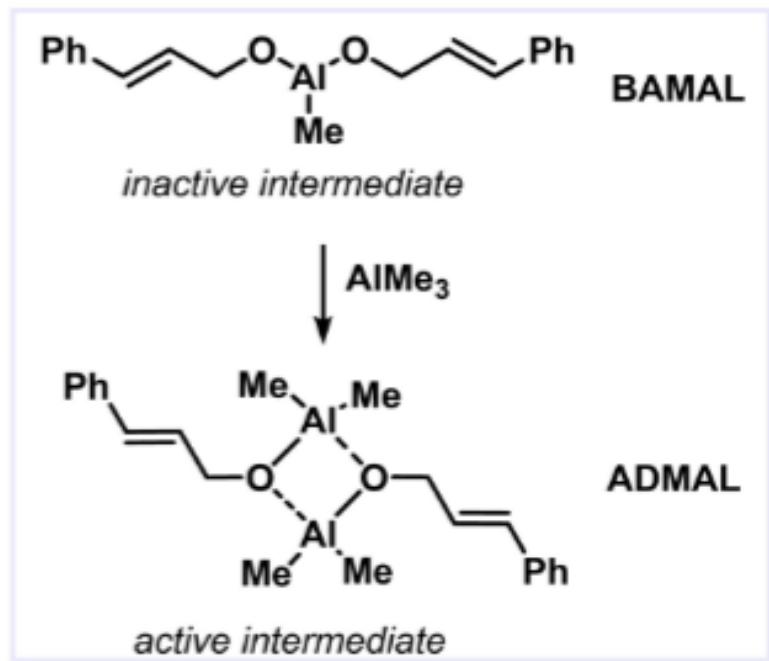


4al, 70%^[b]

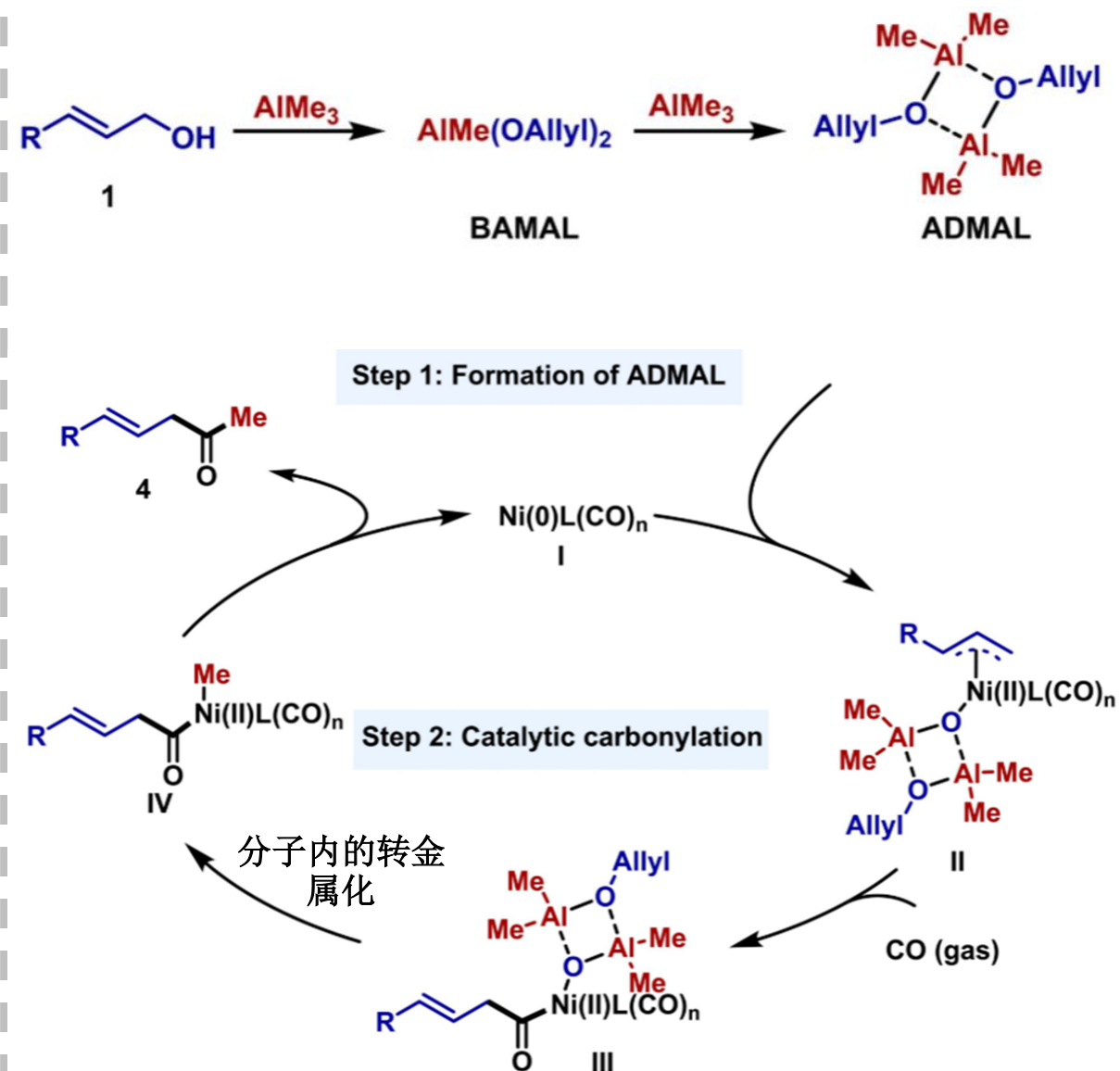
b) Investigation of the amount of AlMe_3



Equiv of 3a	Yield of 4a (%)
0.33	0
0.5	0
0.75	15
1.0	35
1.5	74

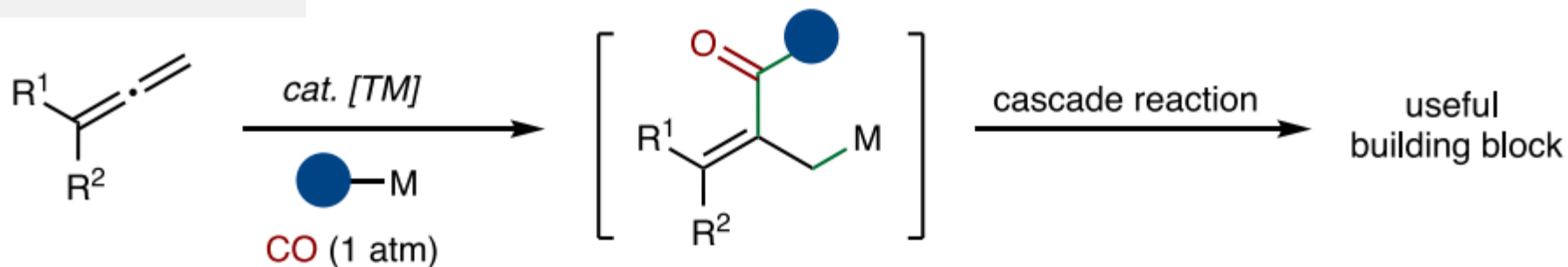


促进C-O键氧化断裂的反应中间体

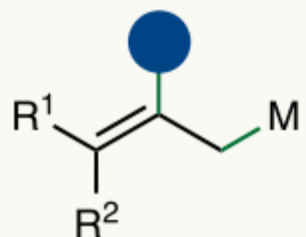


3. Catalytic Carbonylation of 1 atm CO by Nickel

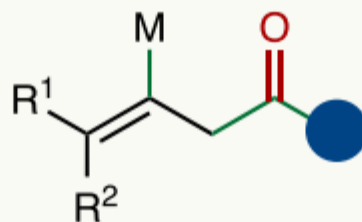
(iii) allene (elusive)



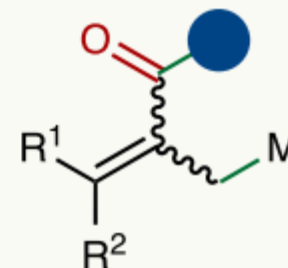
potential challenges



chemoselectivity

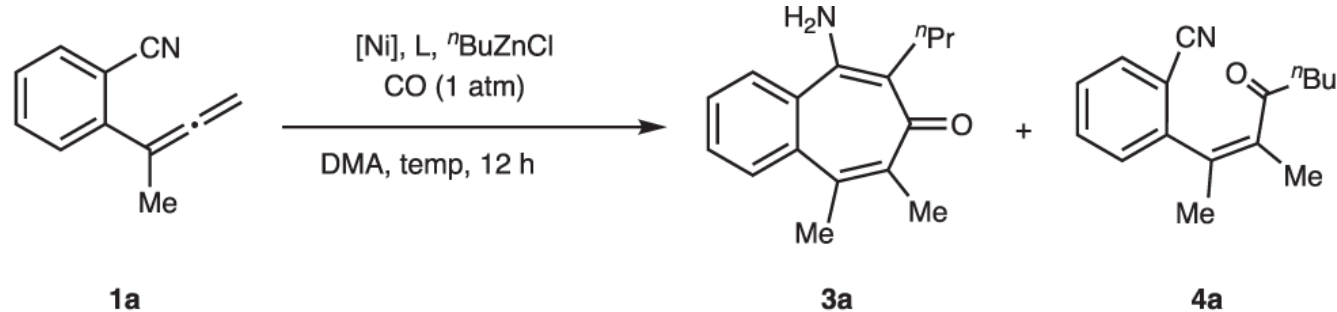


regioselectivity

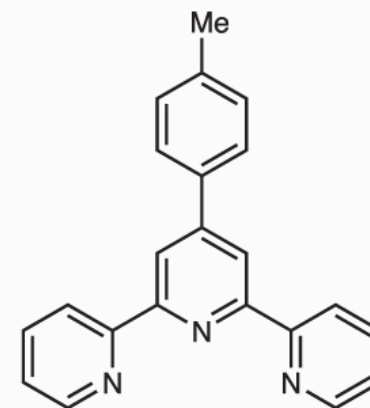


stereoselectivity

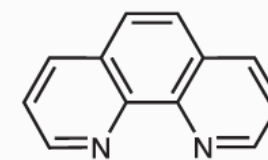
Nat. Commun. 2023, 14, 6960.



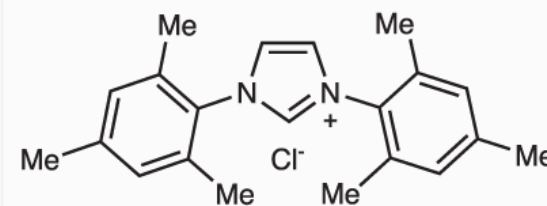
Entry	Ligand	Catalyst	T (°C)	Yield (%) ^a	
				3a	4a
[Ni](0.01 mmol, 10 mol%) and L (0.02mmol, 20 mol%)					
1	L1	Ni(acac) ₂	r.t.	33	28
2	L1	Ni(acac) ₂	40 °C	53	3
3	L1	Ni(acac) ₂	60 °C	51	2
4	L2	Ni(acac) ₂	40 °C	38	7
5	L3	Ni(acac) ₂	40 °C	57	6
6	L4	Ni(acac) ₂	40 °C	61	6
7	L5	Ni(acac) ₂	40 °C	63	7
8	L6	Ni(acac) ₂	40 °C	63	5
9	L6	NiCl ₂ •DME	40 °C	61	5
10	L6	NiBr ₂ •DME	40 °C	67	2
11 ^b	L6	NiBr ₂ •DME	40 °C	75 (71) ^c	2
[Ni] (0.0025 mmol, 2.5 mol%), L6 (0.005 mmol, 5 mol%)					



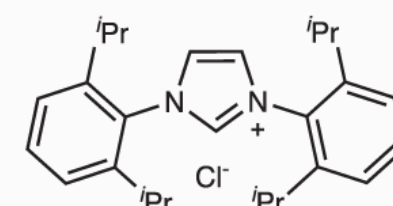
L1



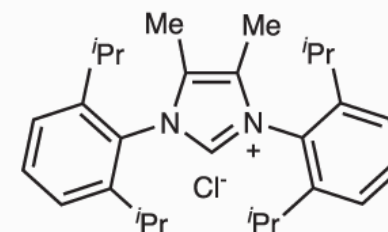
L2



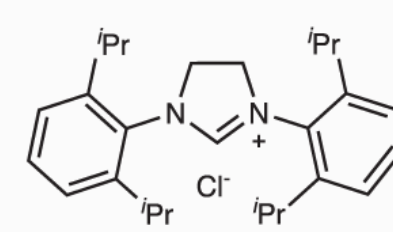
L3



L4

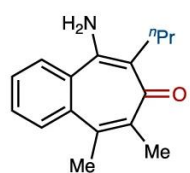
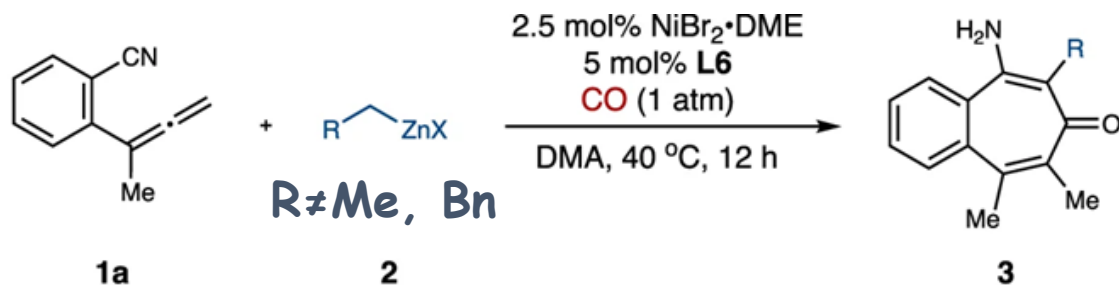


L5

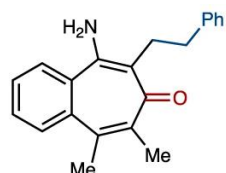


L6

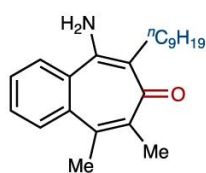
3. Catalytic Carbonylation of 1 atm CO by Nickel



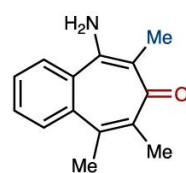
3a, 71%



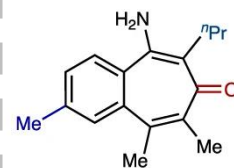
3b, 67%



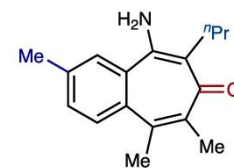
3c, 55%



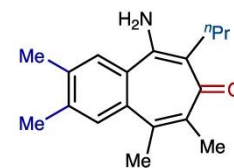
3d, 35%



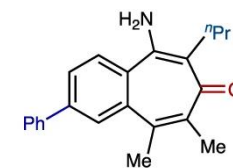
3m, 63%



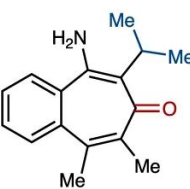
3n, 62%



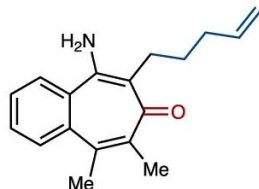
3o, 61%



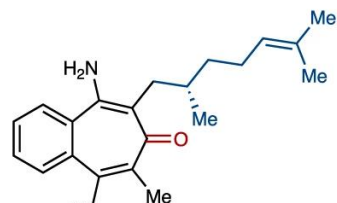
3p, 57%



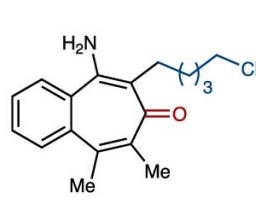
3e, 50%



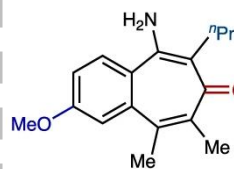
3f, 61%



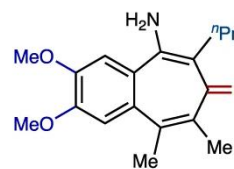
3g, 65%



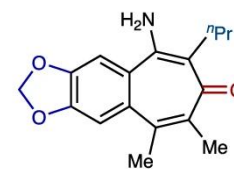
3h, 50%



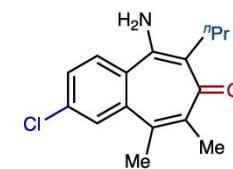
3q, 67%



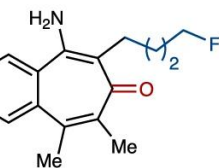
3r, 42%



3s, 63%



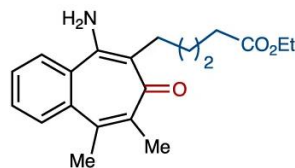
3t, 30%



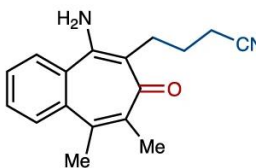
3i, 54%



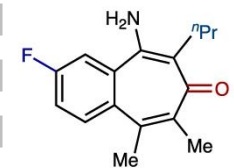
3j, 54%



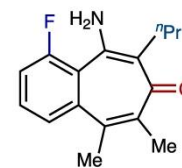
3k, 47%



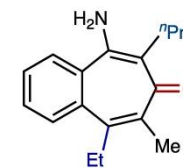
3l, 32%



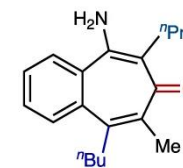
3u, 53%



3v, 32%



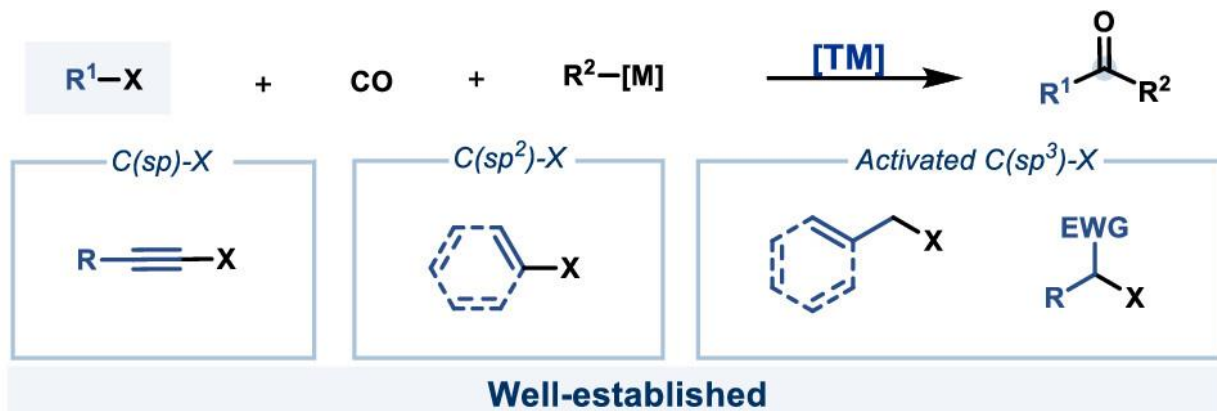
3w, 60%



3x, 51%

3. Catalytic Carbonylation of 1 atm CO by Nickel

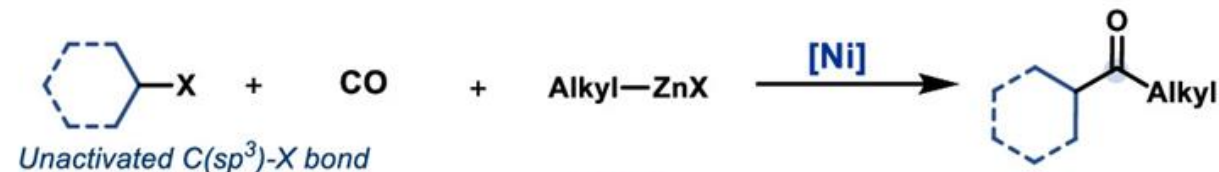
a) Transition metal-catalyzed carbonylative cross-coupling for ketone synthesis



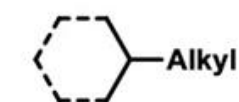
b) Transition metal-catalyzed carbonylative cross-coupling of unactivated alkyl electrophiles for dialkyl ketone synthesis



ii) Ni-catalyzed carbonylative Negishi coupling of unactivated alkyl-halides with CO



Pitfall



Negishi coupling



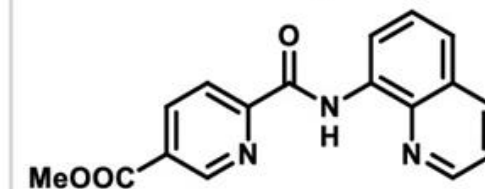
reductive dehalogenation



β -H elimination

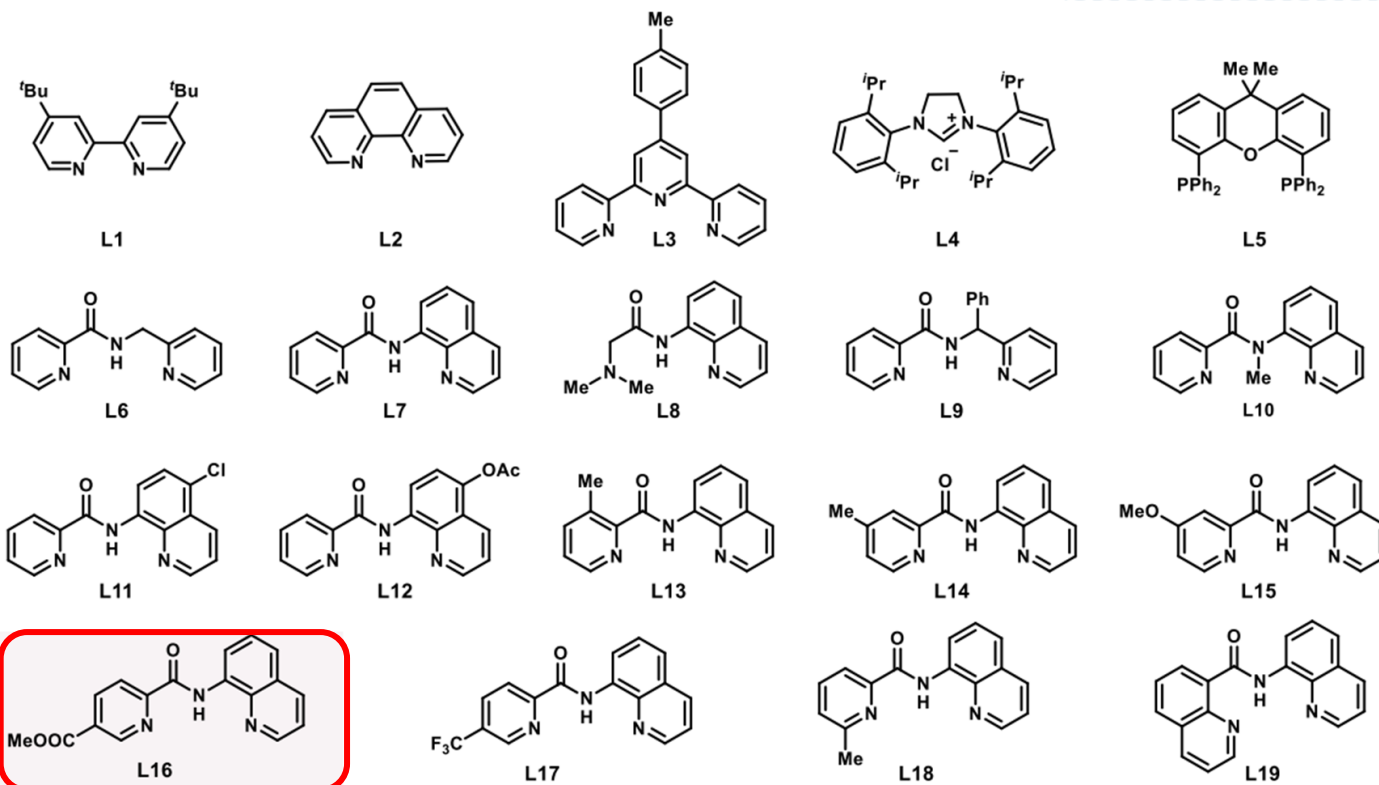
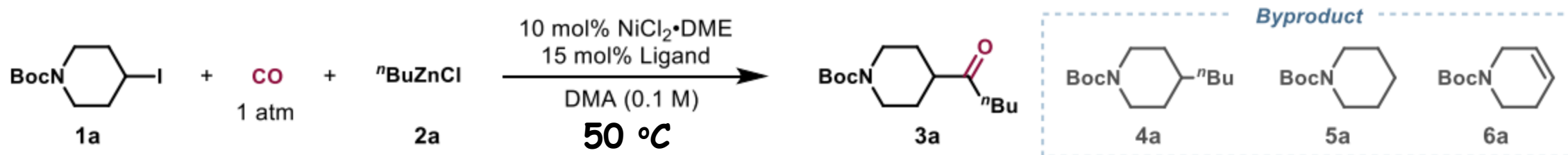
- New tridentate NNN-pincer ligand
- Mild condition with atmospheric CO gas
- Broad functionality with organozinc reagent
- Facile unsymmetric dialkyl ketone synthesis

New NNN-ligand



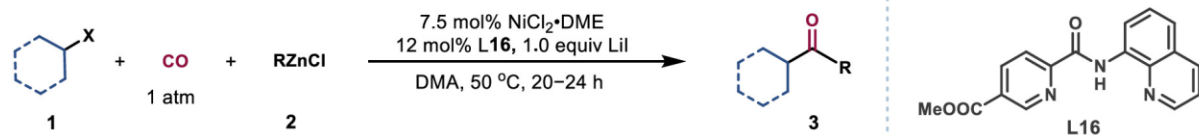
J. Am. Chem. Soc. **2024**, 146, 7971-7978.

3. Catalytic Carbonylation of 1 atm CO by Nickel

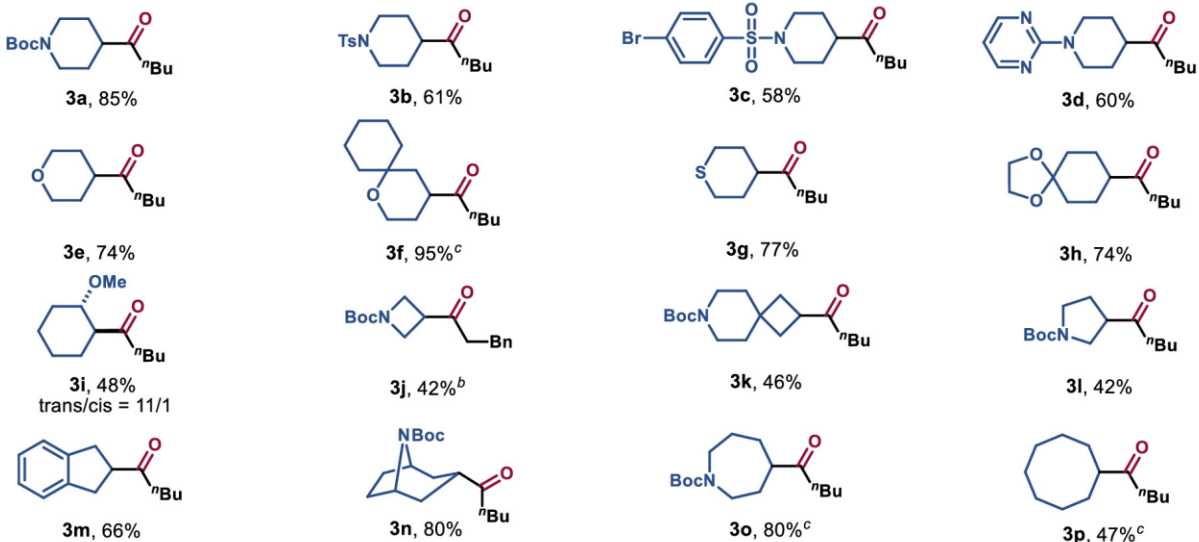


entry	ligand	conversion of 1a (%)	yield of 3a/4a/5a/6a (%) ^a
22 ^{c,d}	L16	100	90 (85) ^e /0/4/1

3. Catalytic Carbonylation of 1 atm CO by Nickel



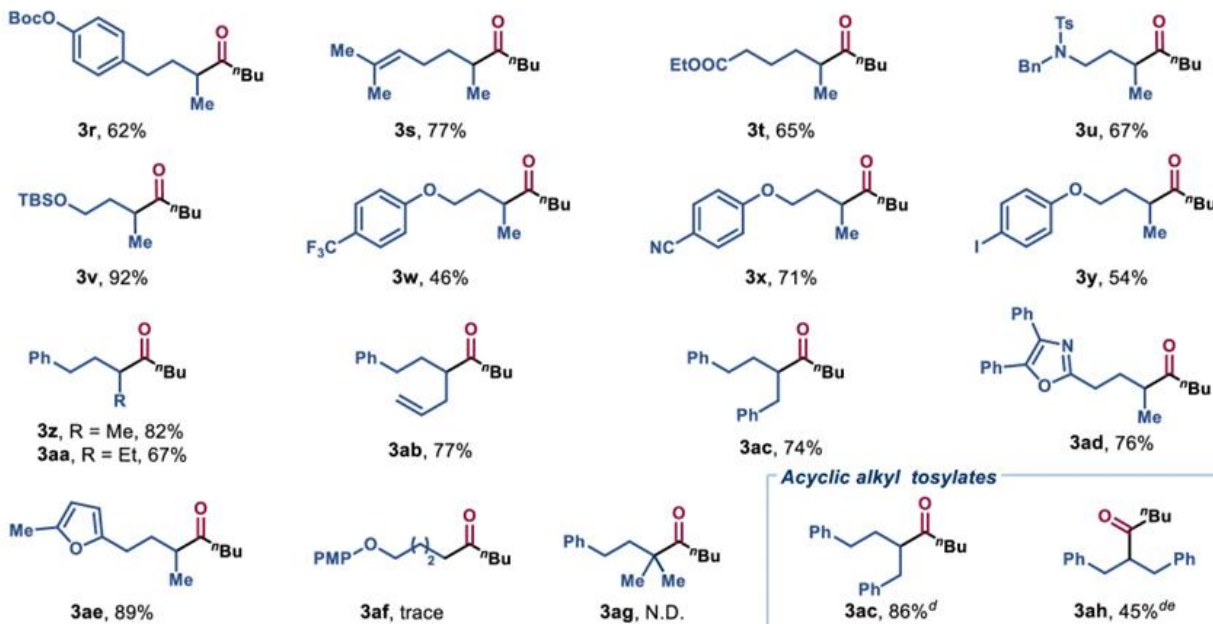
Cyclic alkyl iodides 环状烷基碘



Cyclic alkyl tosylates 环状烷基甲苯磺酸酯

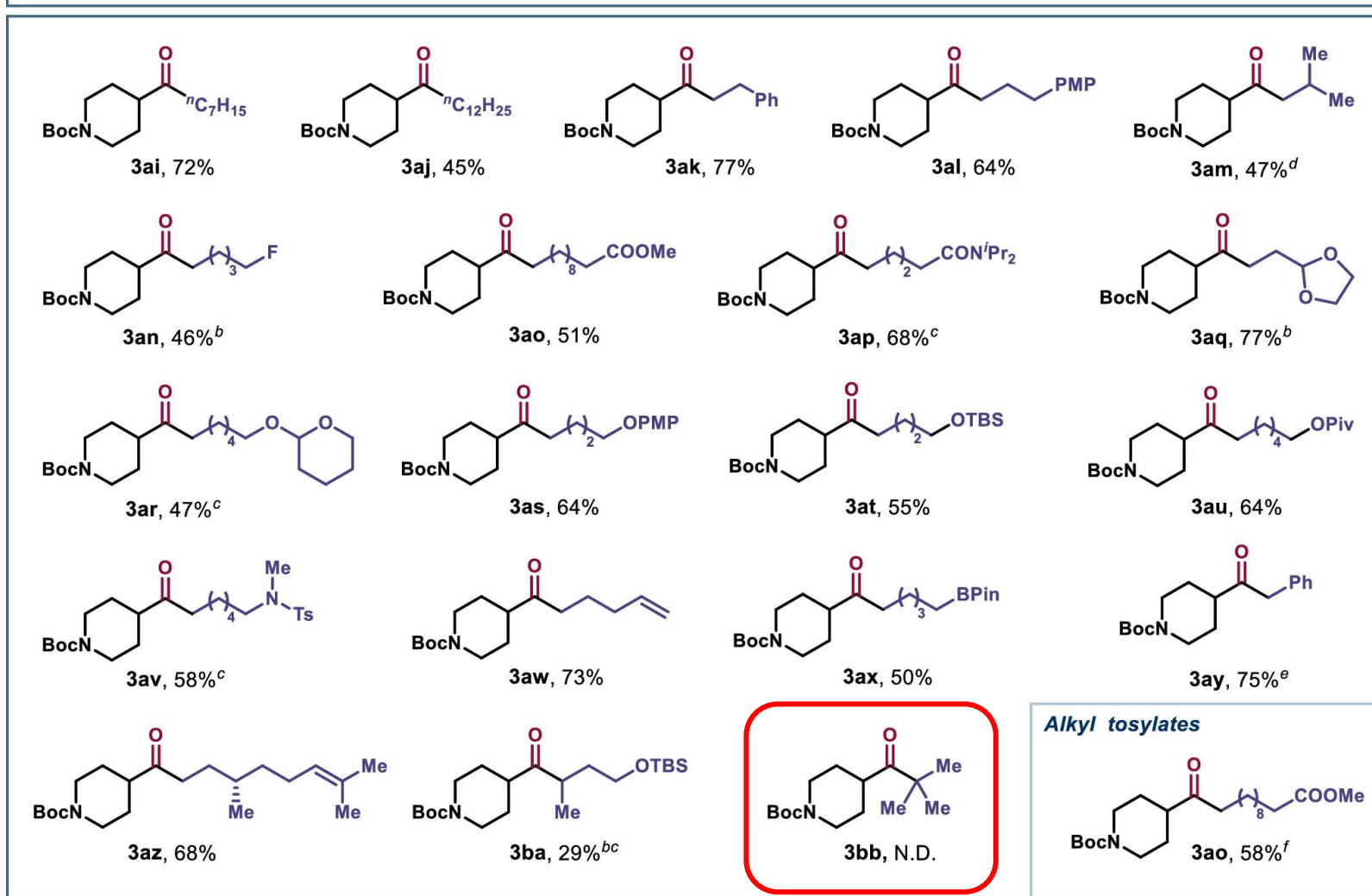
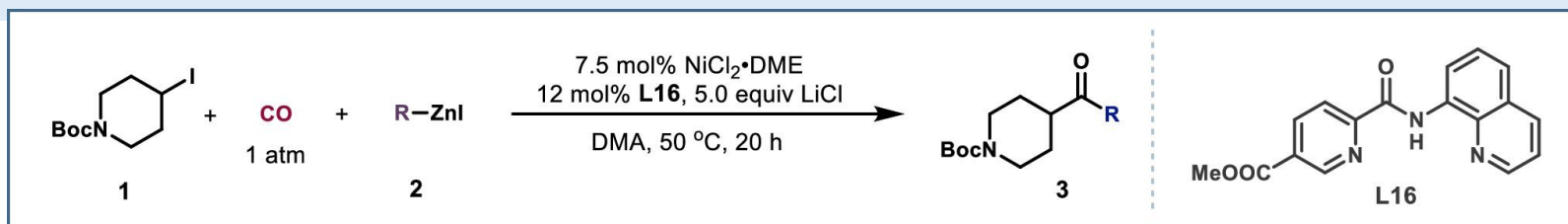


Acyclic alkyl iodides

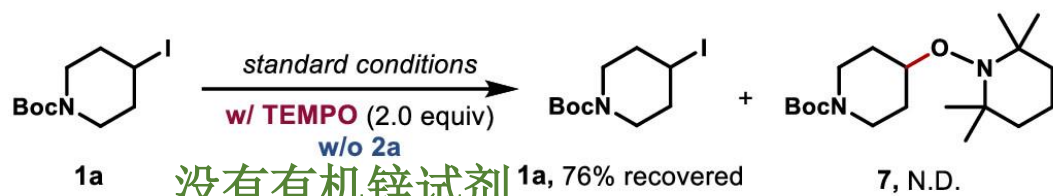
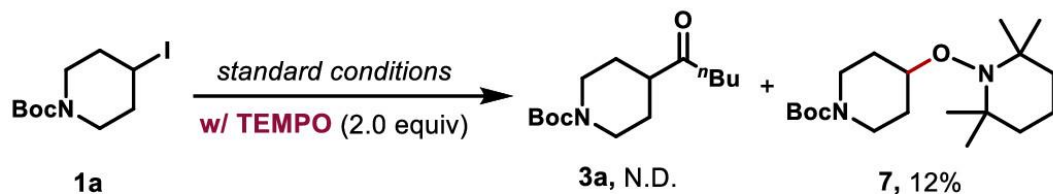


Acyclic alkyl tosylates

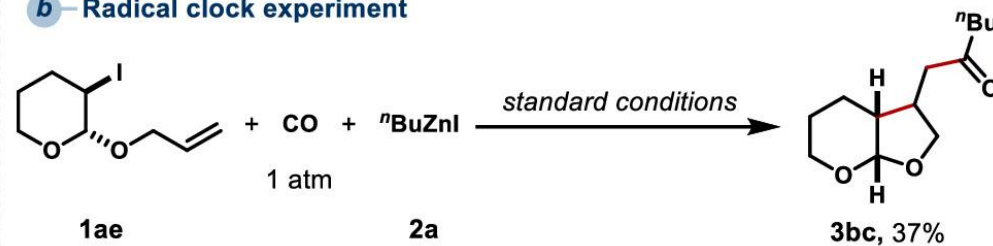
3. Catalytic Carbonylation of 1 atm CO by Nickel



a Radical trapping experiments



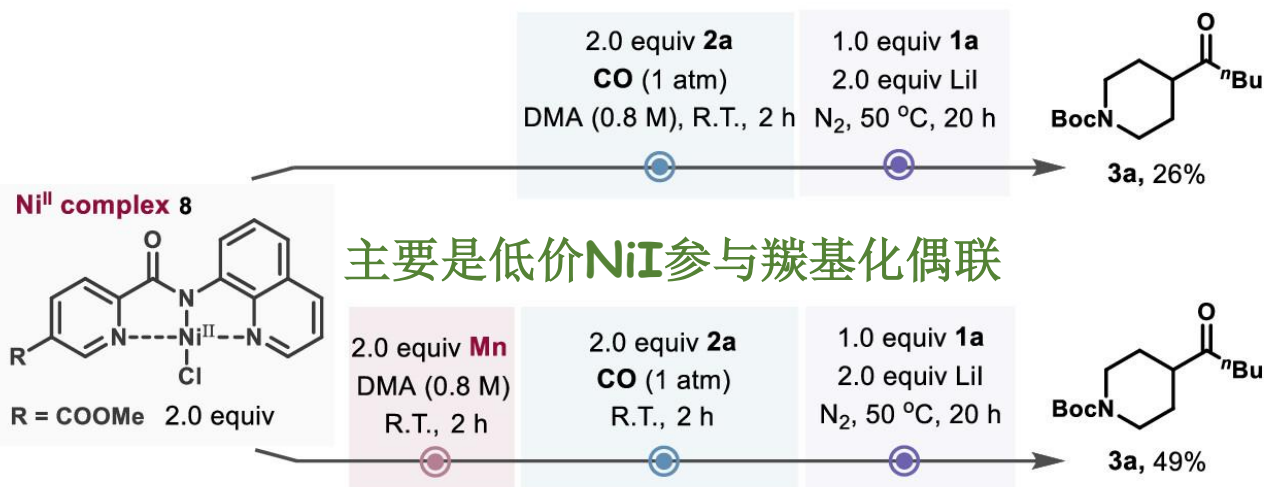
b Radical clock experiment



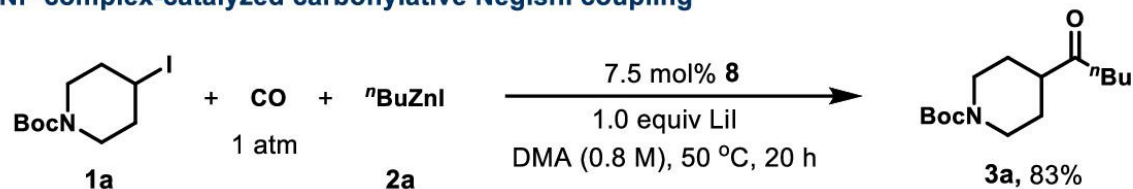
c Testing of stereospecific alkyl iodide



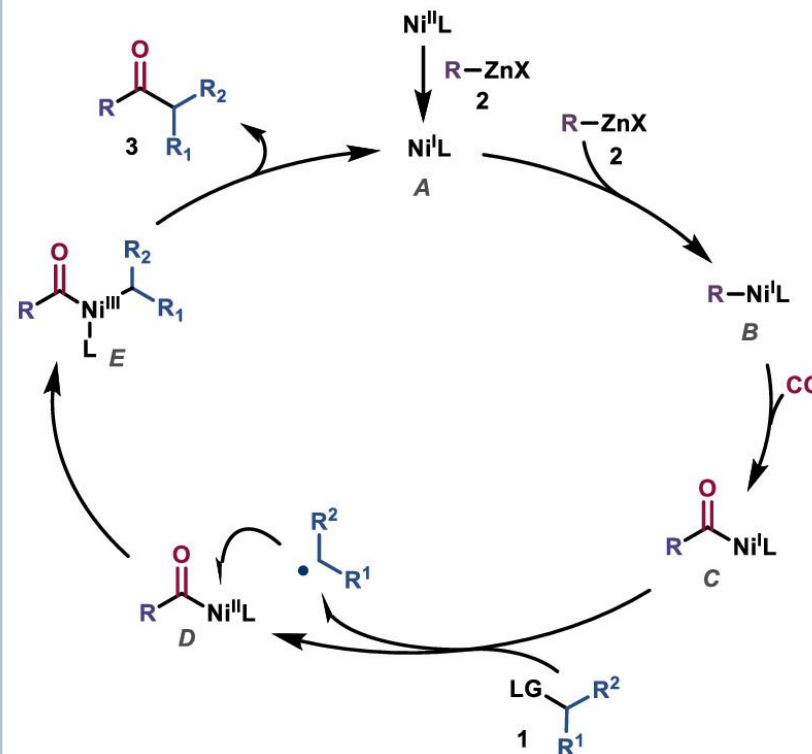
d Stoichiometric reaction



e Ni^{II} complex-catalyzed carbonylative Negishi coupling

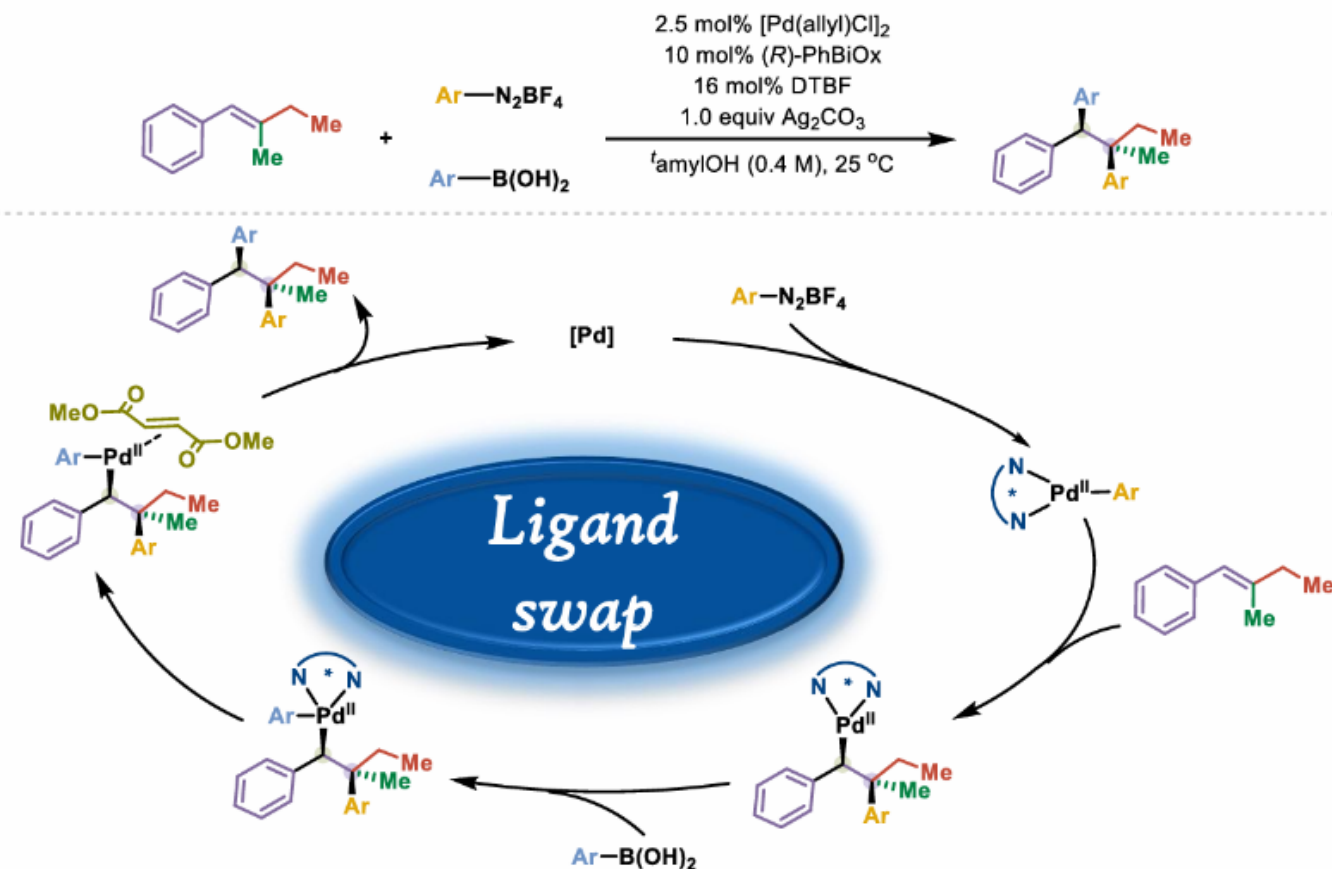


f Plausible mechanism



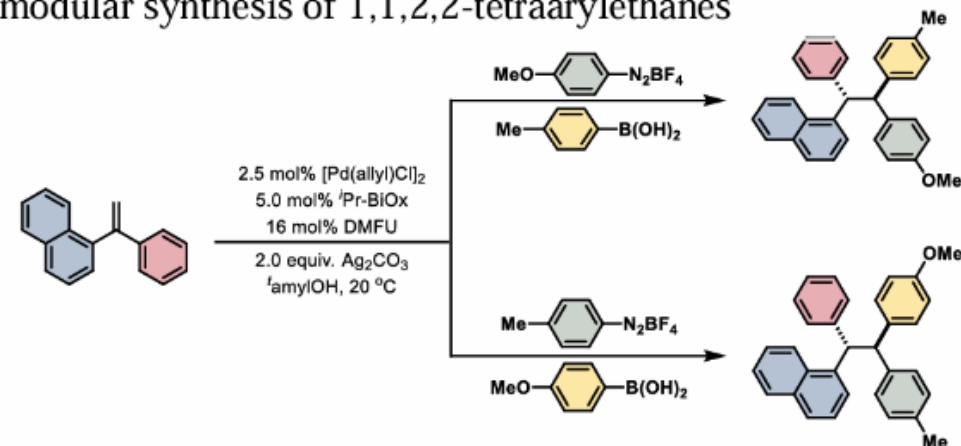
4. Catalytic Intermolecular Cross-Coupling of Olefins

Ligand-swap enabled DF cross-coupling of trisubstituted olefin

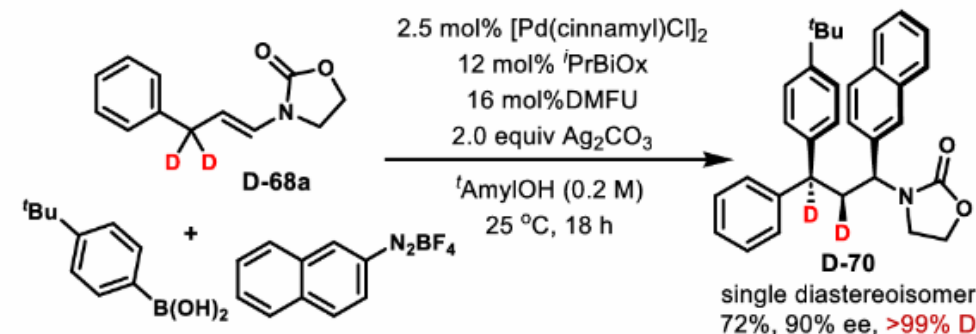


1,1-DF cross-coupling

modular synthesis of 1,1,2,2-tetraarylethanes

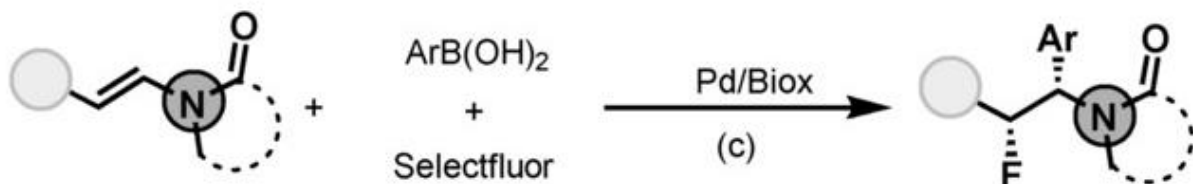


1,3-DF cross-coupling



4. Catalytic Intermolecular Cross-Coupling of Olefins

2) This work: Pd-catalyzed asymmetric arylation of internal enamides



■ Enantioselective difunctionalization of internal enamides for C-C bond formation

■ β -Fluoroamine derivatives with vicinal chiral centers up to 99% ee

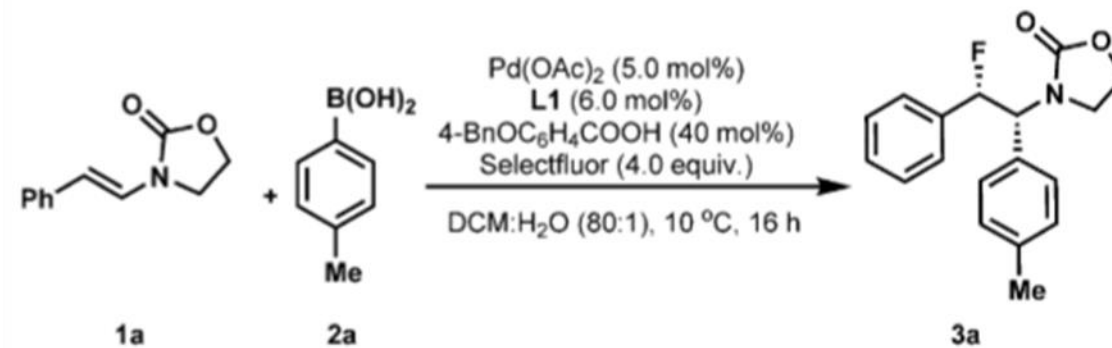
■ High regio-, diastereoselectivities enabled by removable oxazolidinone



[a] Reaction conditions: **1a** (0.2 mmol), **2a** (2.0 equiv, 0.4 mmol), Selectfluor (4.0 equiv, 0.8 mmol), Pd(OAc)₂ (5.0 mol %, 0.01 mmol), ligand (6.0 mol %, 0.012 mmol), 4-(benzyloxy)benzoic acid (40 mol %, 0.08 mmol), DCM:H₂O (v/v) = 80:1 (0.2 M), 10 °C, 16 h. [b] Yield determined by ¹H NMR using CH₂Br₂ as the internal standard. [c] Determined by chiral HPLC. [d] Yield of isolated products in parentheses. [e] DCM:H₂O (v/v) = 9:1 (0.2 M). [f] Selectfluor (2.0 equiv, 0.4 mmol). [g] Without 4-(benzyloxy)benzoic acid.

Scheme 1. Enantioselective synthesis of amine derivatives with vicinal stereocenters from alkenes.

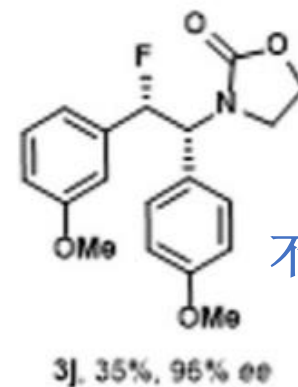
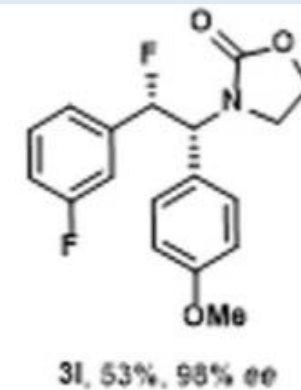
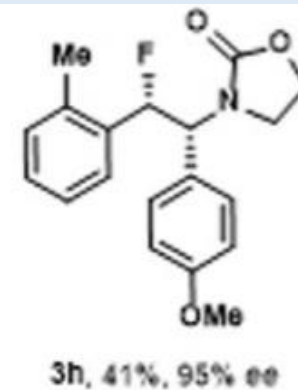
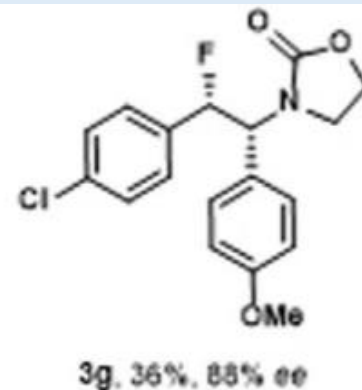
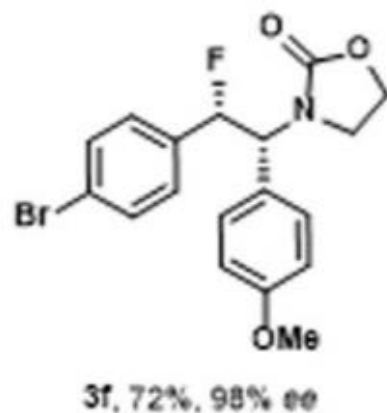
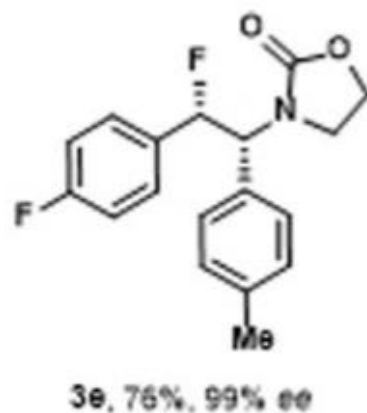
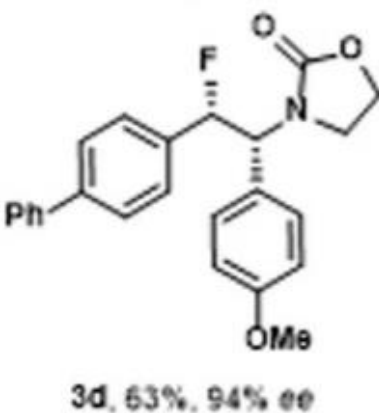
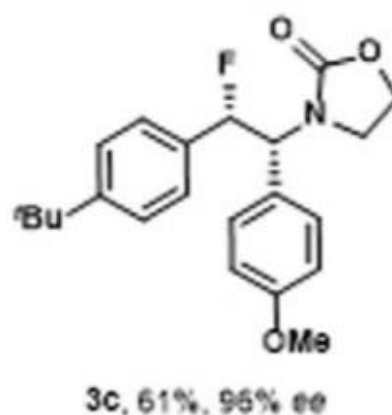
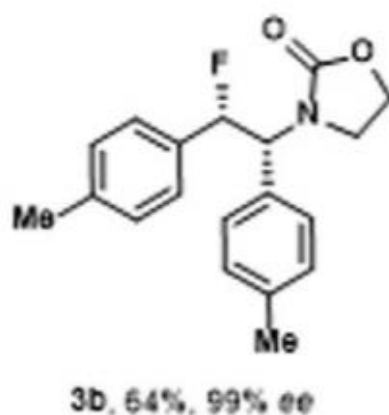
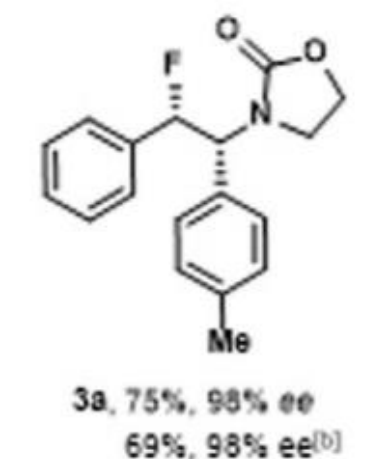
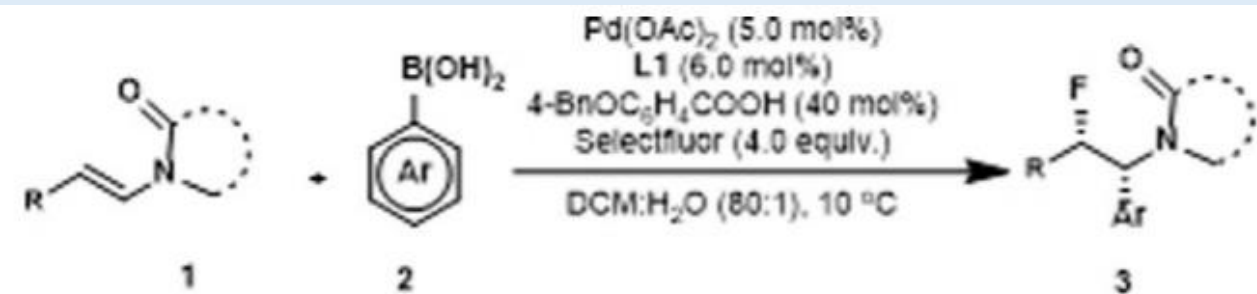
Table 1: Optimization of the reaction conditions.^[a]



Entry	Variation from standard conditions	Yield (%) ^[b]	ee (%) ^[c]
1	none	84 (75) ^[d]	98
2	DCM:H ₂ O (v/v) = 9:1 as solvent	79	98
3	no H ₂ O	nr	—
4 ^[e]	no 4-BnOC ₆ H ₄ COOH	77	98
5 ^[e,f]	25 °C instead of 10 °C	20	—
6 ^[e,g]	2.0 equiv Selectfluor	66	98
7 ^[e]	L2 instead of L1	76	98
8 ^[e]	L3 instead of L1	26	—
9 ^[e]	L4 instead of L1	71	95
10 ^[e]	L5 instead of L1	46	-38

Angew. Chem. Int. Ed. 2021, 60, 2699-2703.

4. Catalytic Intermolecular Cross-Coupling of Olefins



烯酰胺R
不能是烷基

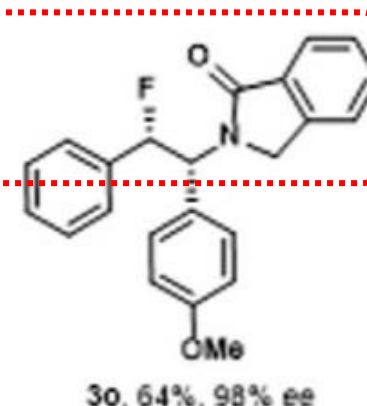
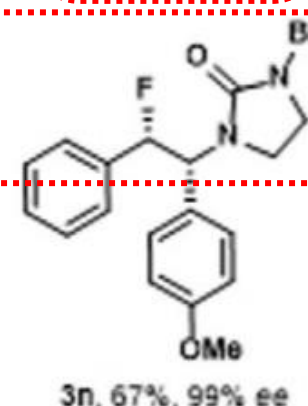
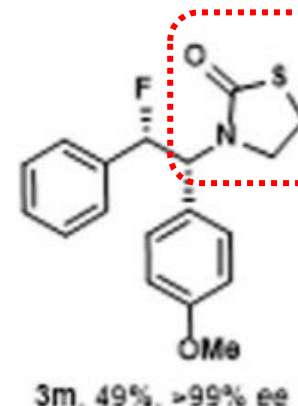
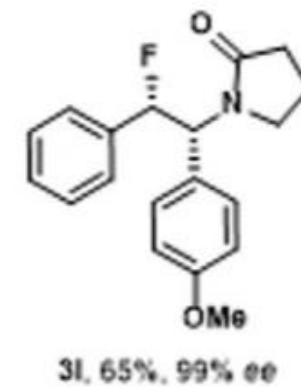
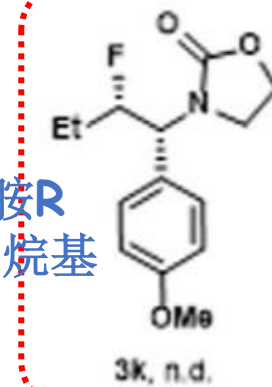
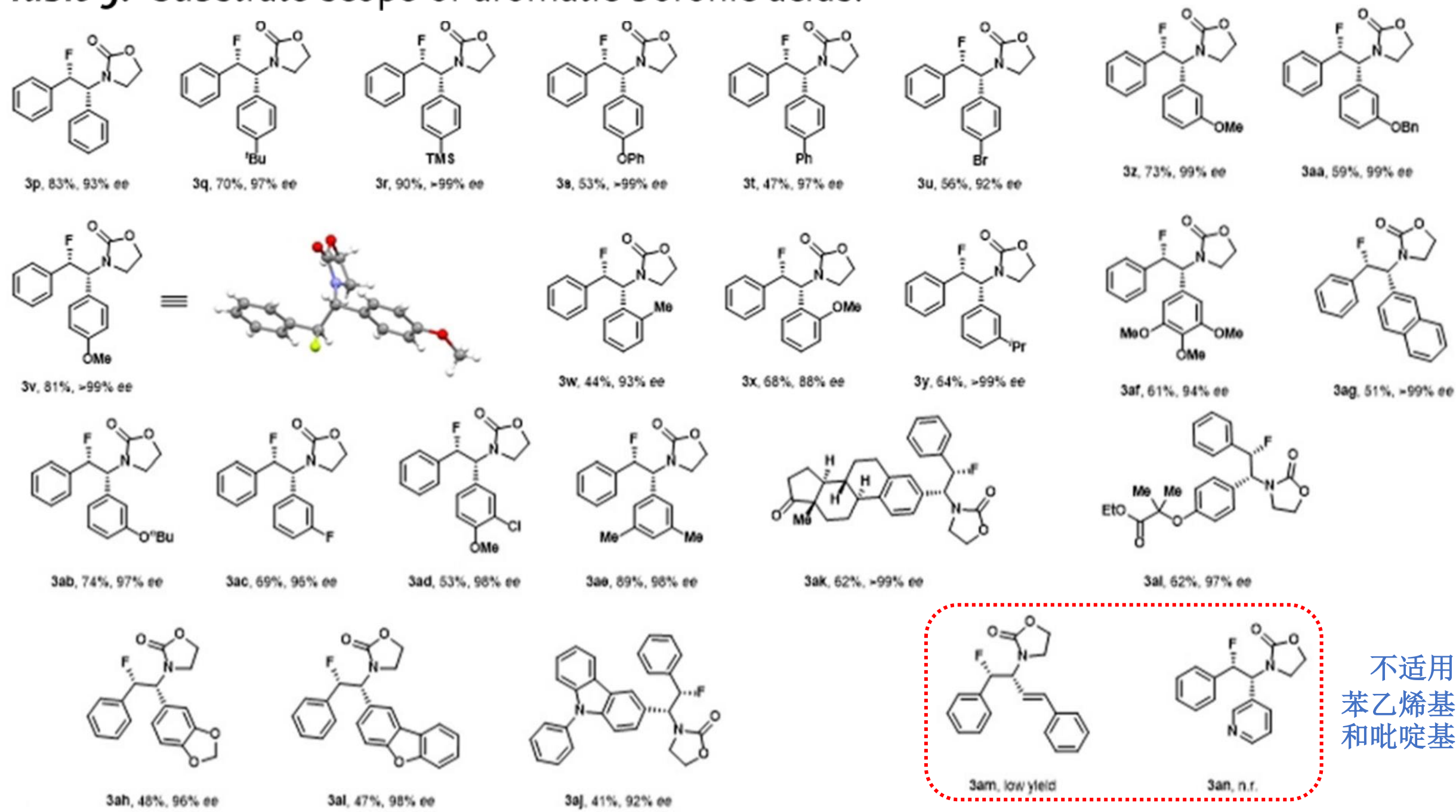


Table 3: Substrate scope of aromatic boronic acids.^[a]

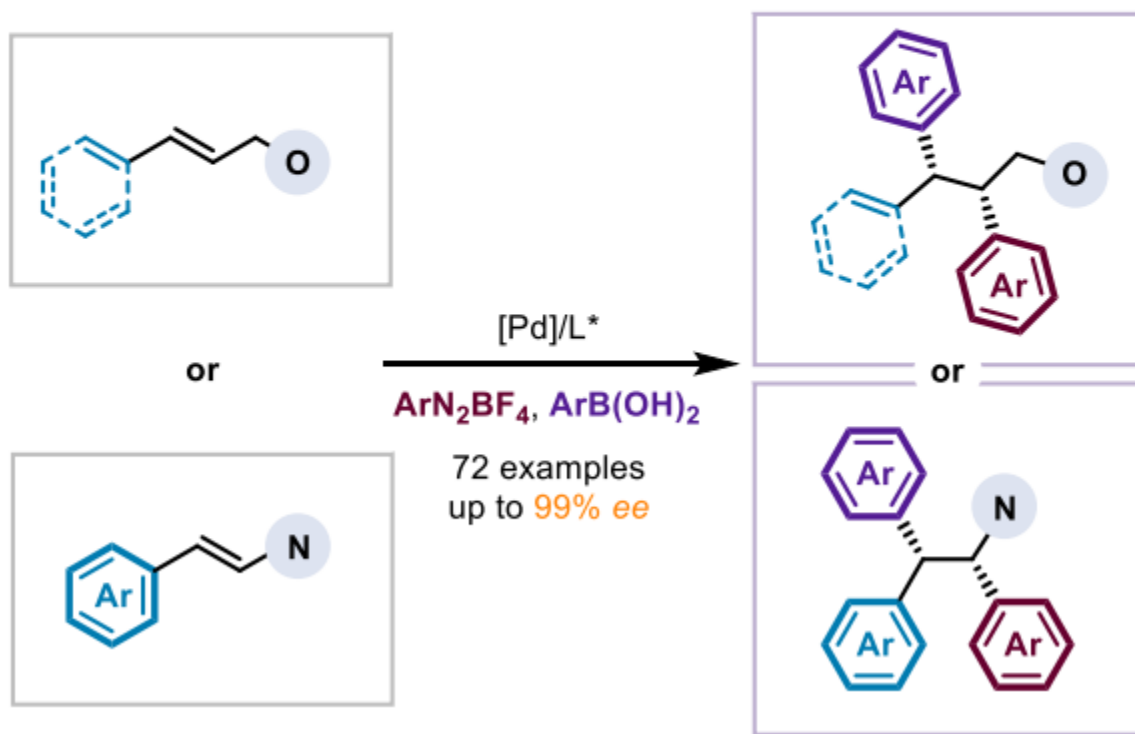


不适用于
苯乙烯基硼酸
和吡啶基硼酸

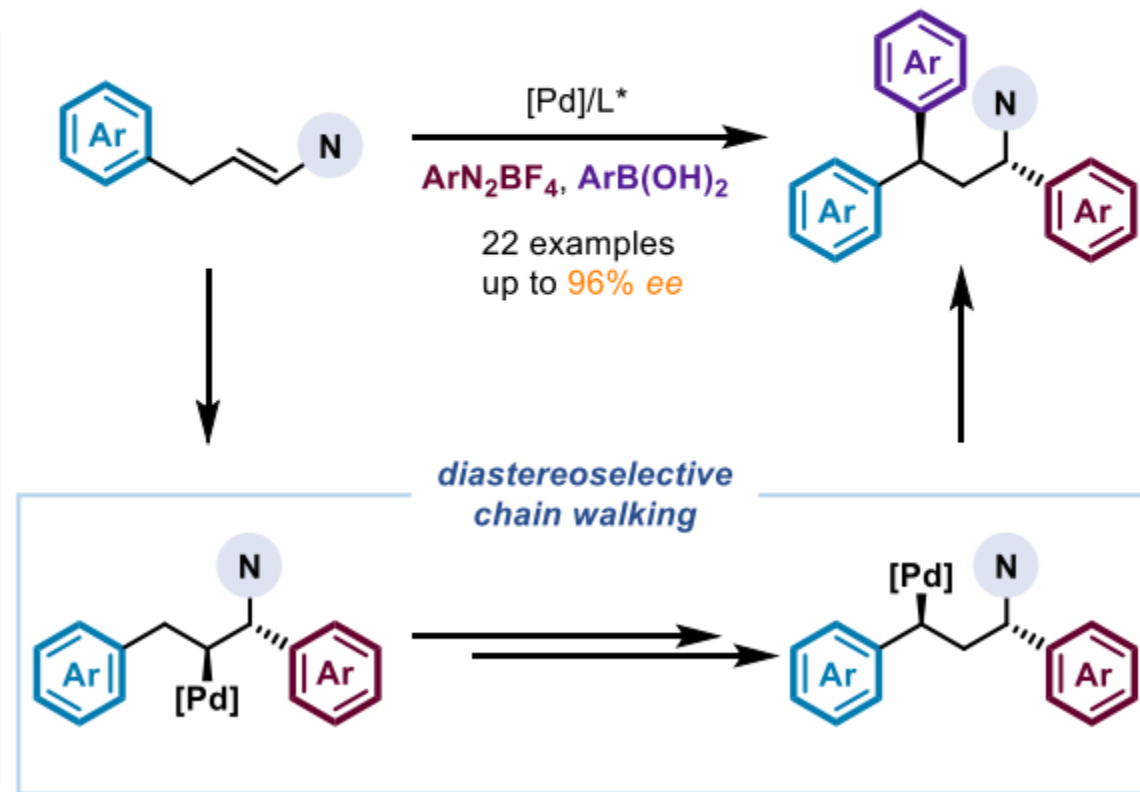
4. Catalytic Intermolecular Cross-Coupling of Olefins

c Pd-catalyzed asymmetric divergent diarylation of internal acyclic alkenes (**this study**)

i: vicinal 1,2-diarylation

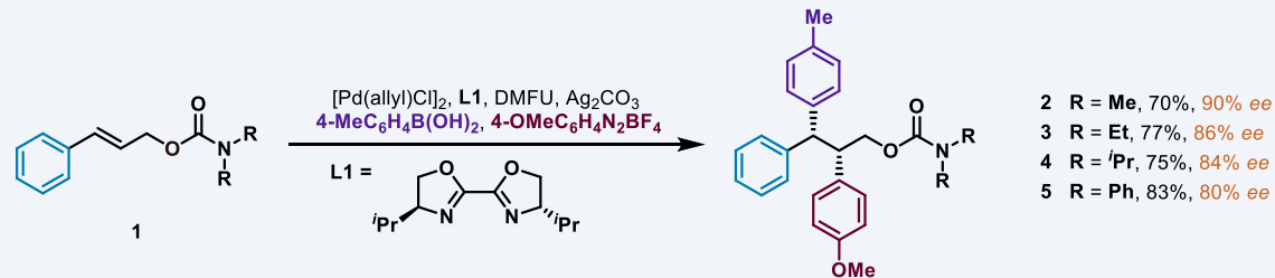


ii: migratory 1,3-diarylation



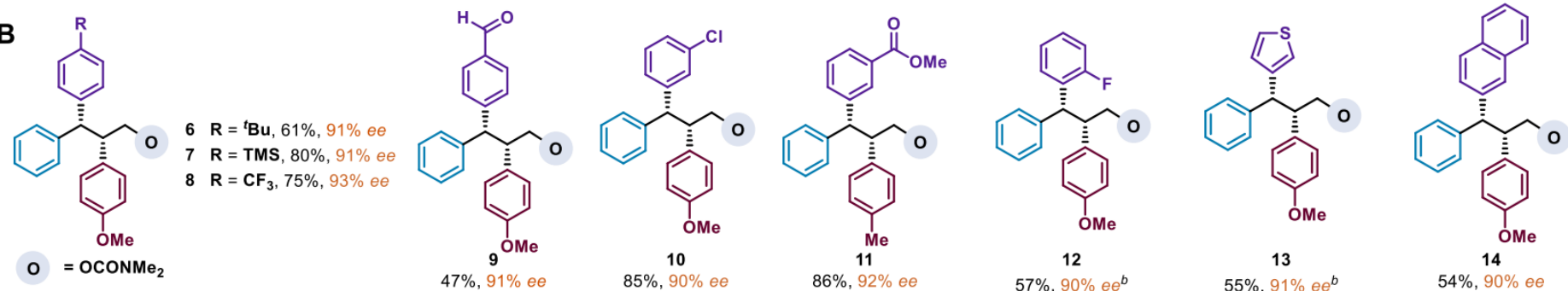
J. Am. Chem. Soc. **2022**, *144*, 8389–8398.

A

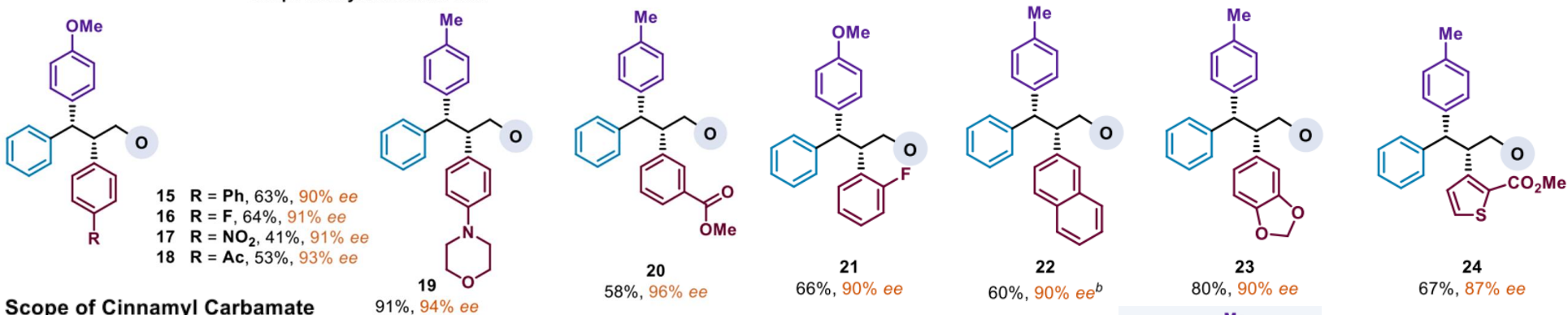


Scope of Arylboronic Acid

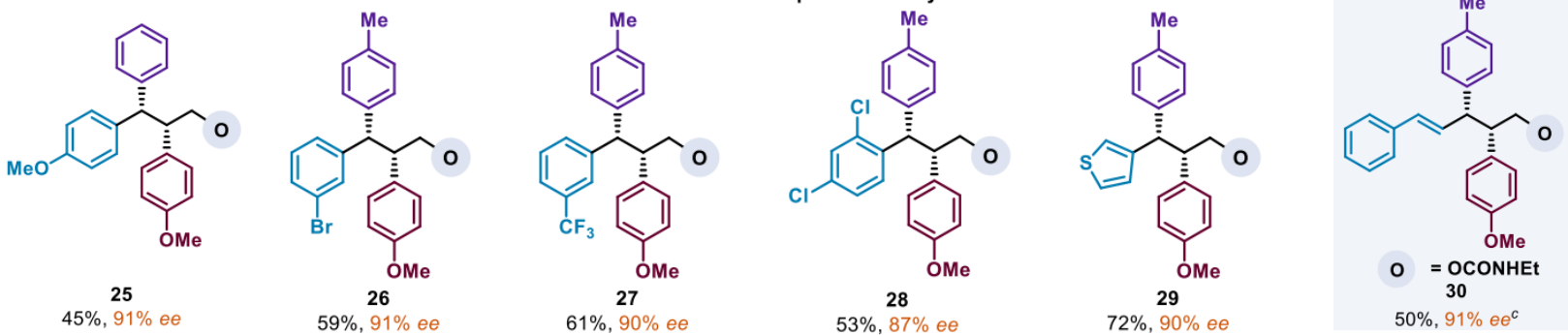
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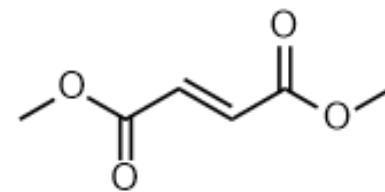
Scope of Aryldiazonium Salt



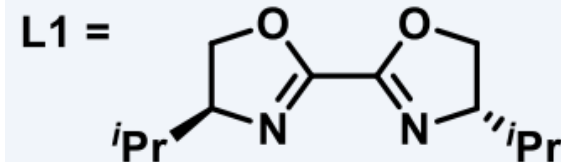
Scope of Cinnamyl Carbamate

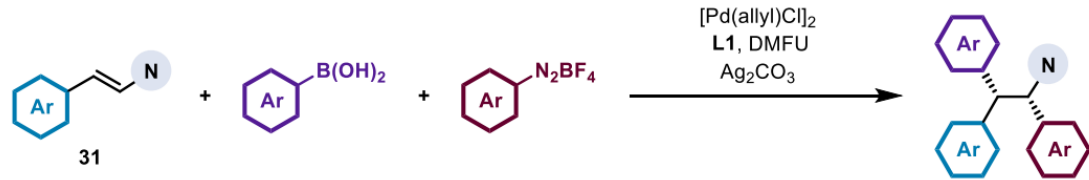


苯基取代的氨基
甲酸烯丙酯

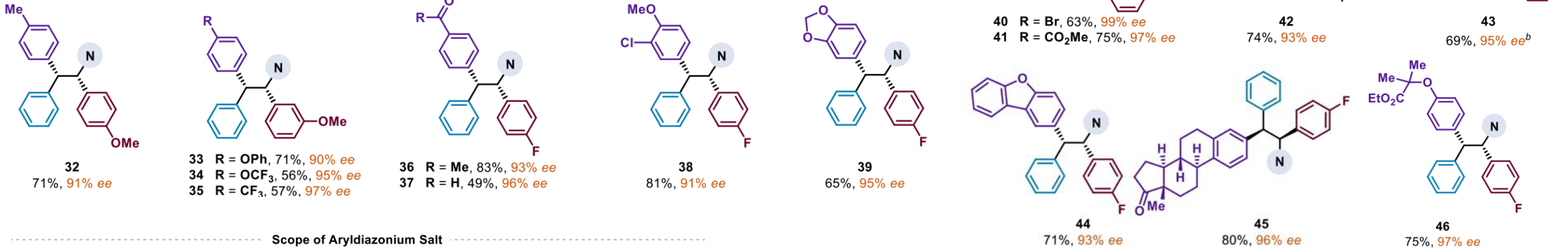
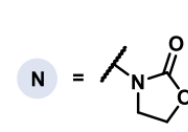


DMFU

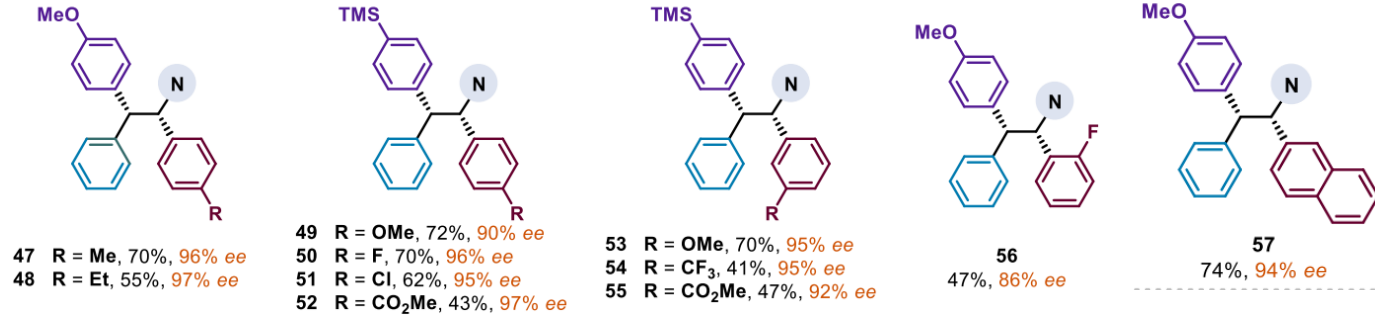




Scope of Arylboronic Acid



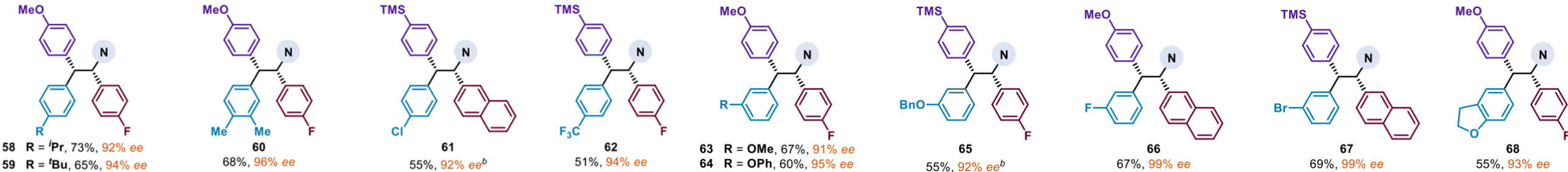
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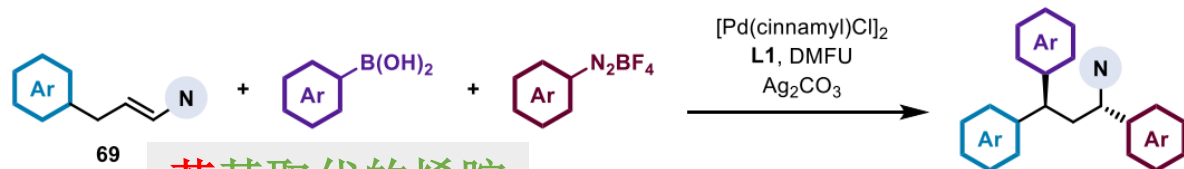


苯基取代的烯胺

[Pd(allyl)Cl]₂

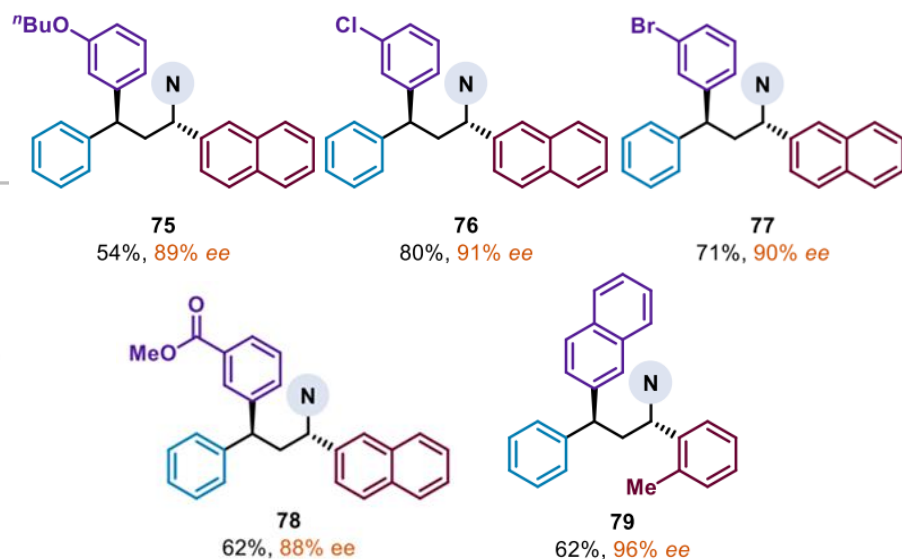
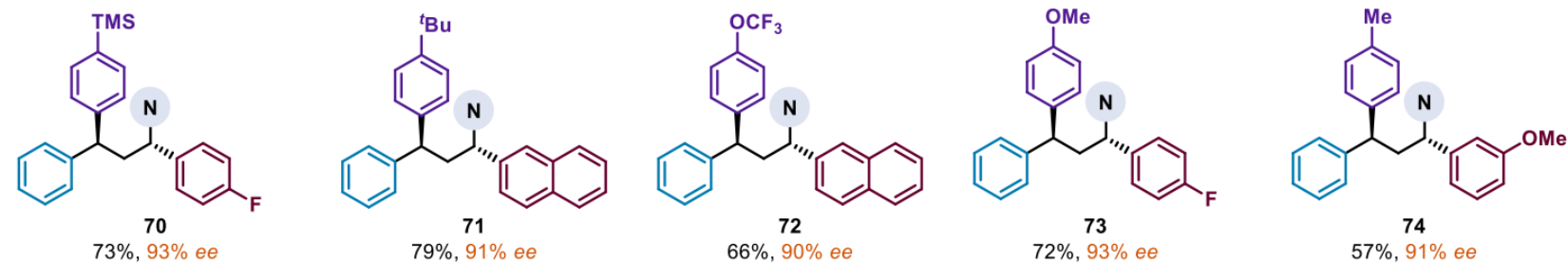
Scope of Internal Enamide



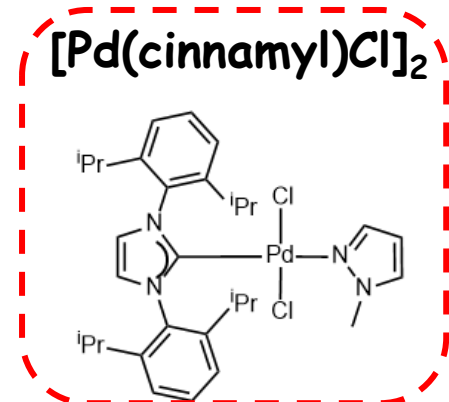
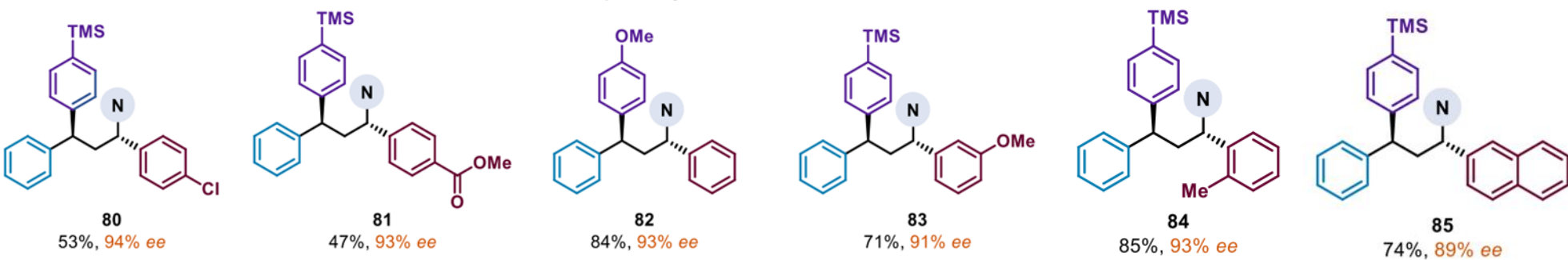


苯基取代的烯胺

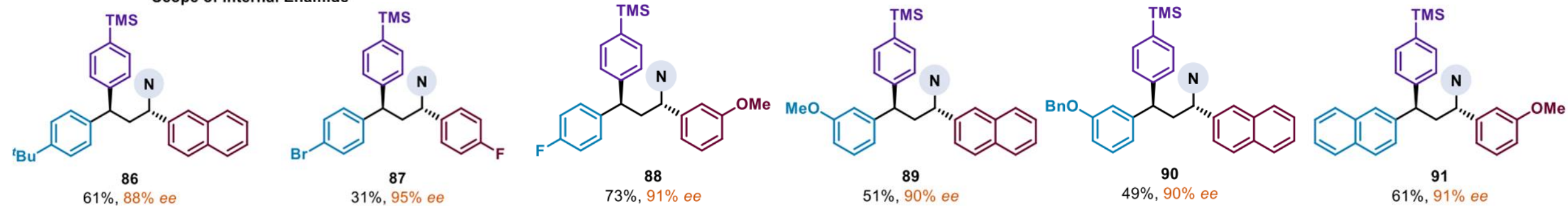
Scope of Arylboronic Acid



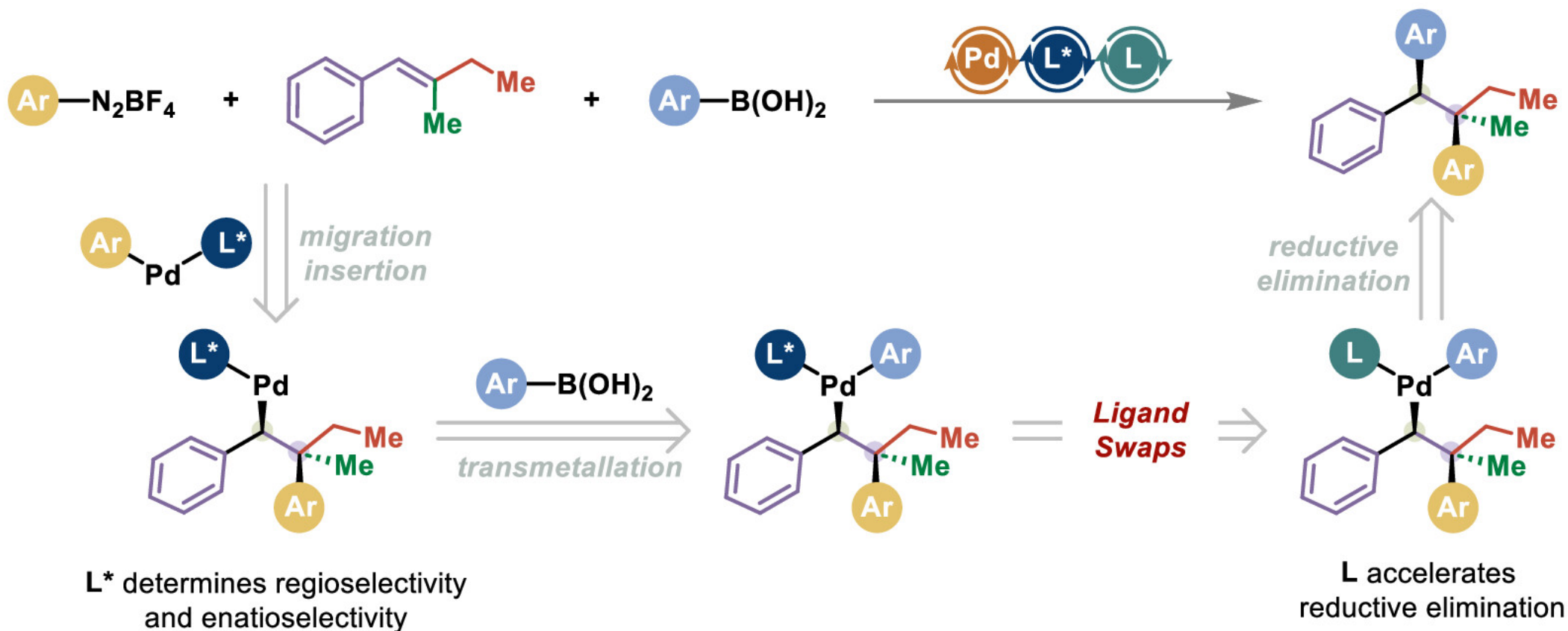
Scope of Aryldiazonium Salt



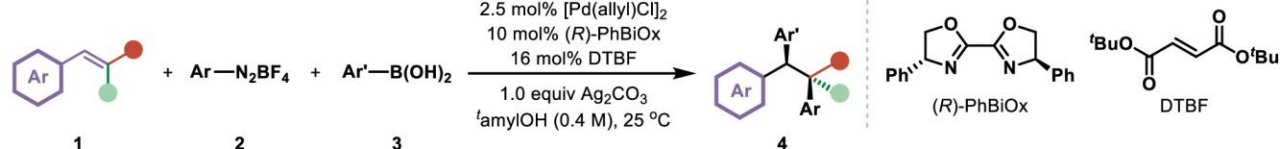
Scope of Internal Enamide



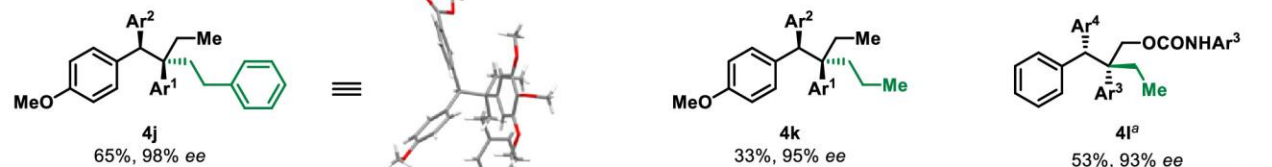
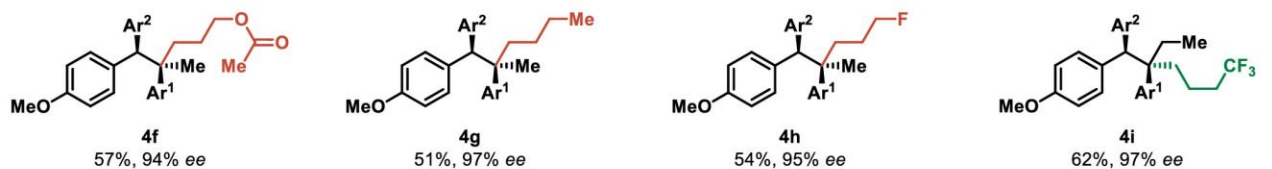
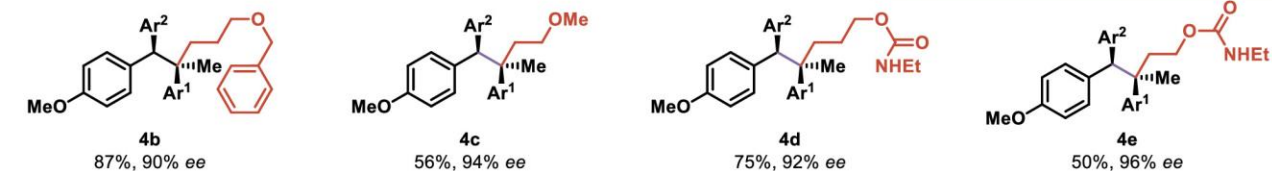
4. Catalytic Intermolecular Cross-Coupling of Olefins



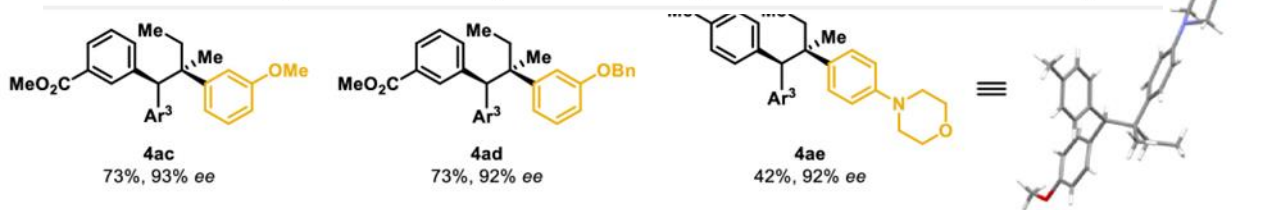
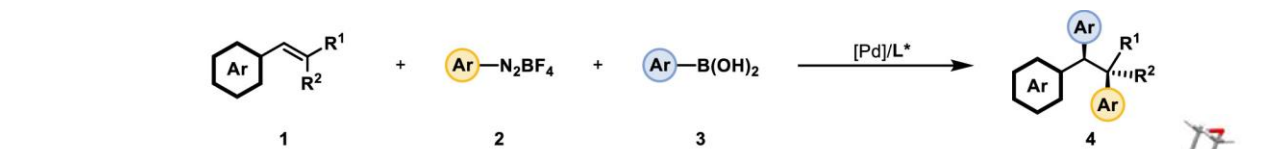
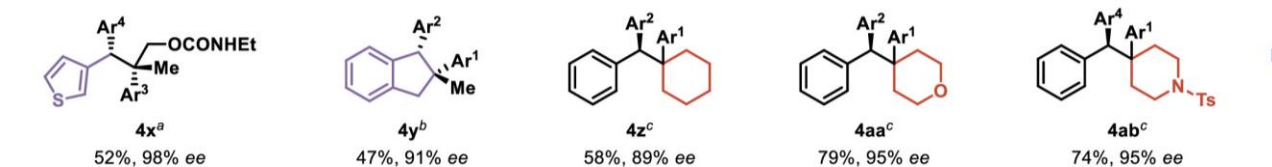
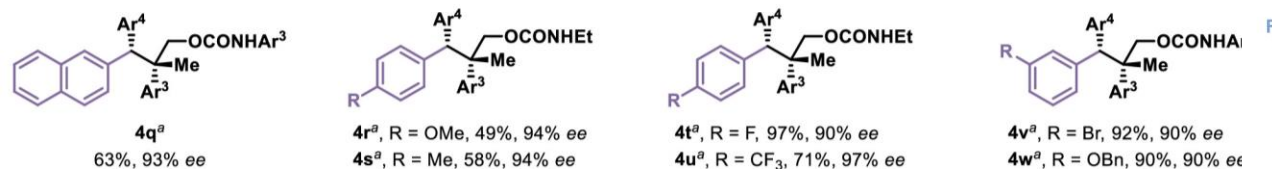
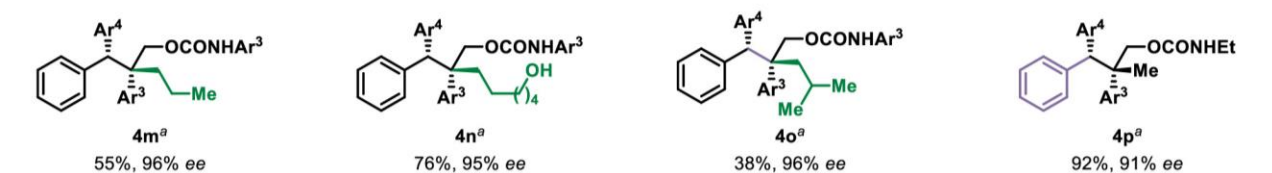
J. Am. Chem. Soc. **2024**, 146, 24, 16892-16901.



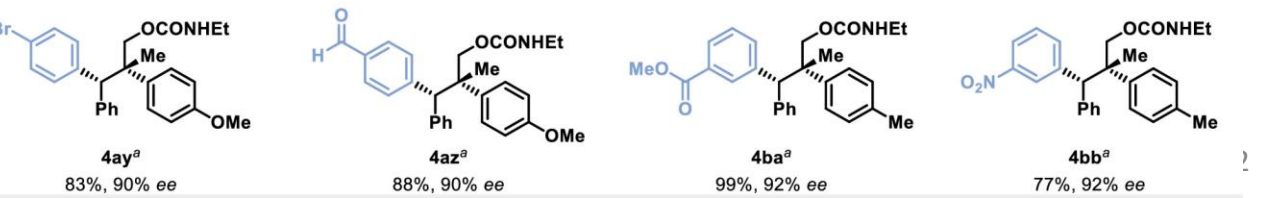
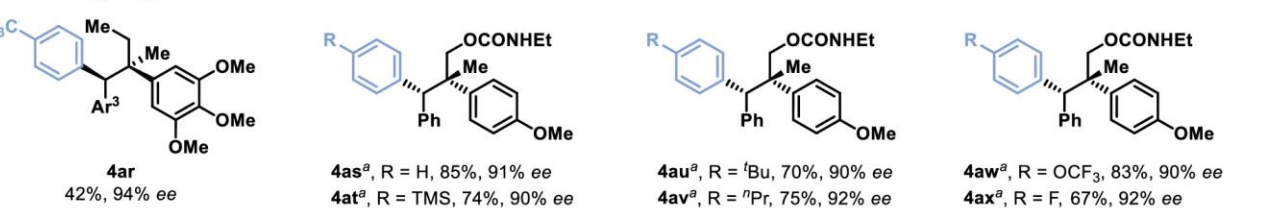
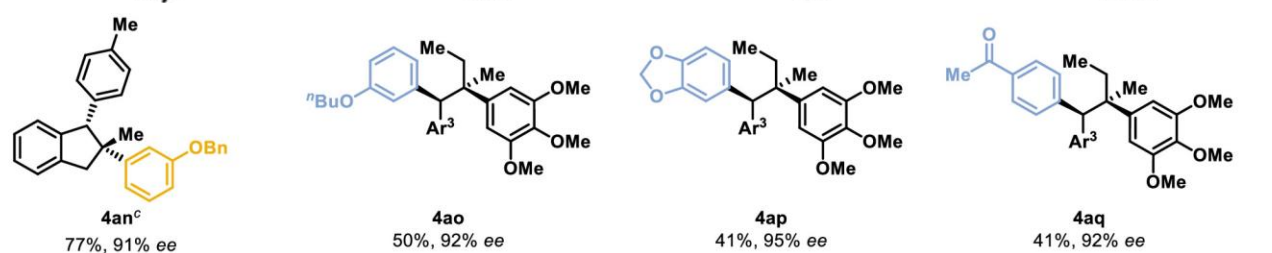
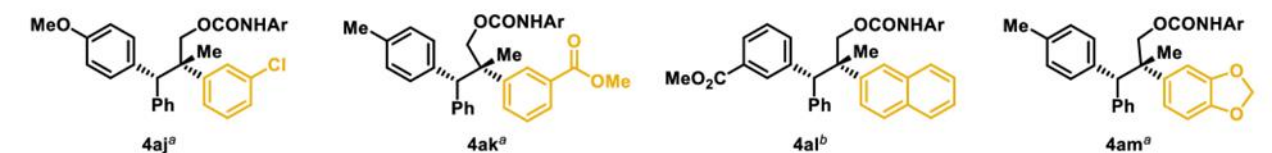
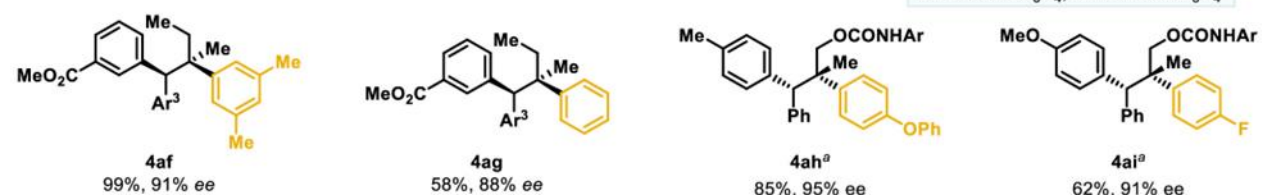
Ar¹ = 3,4,5-triMeOC₆H₂, Ar² = 3-CO₂MeC₆H₄



Ar³ = 4-MeOC₆H₄, Ar⁴ = 4-MeC₆H₄

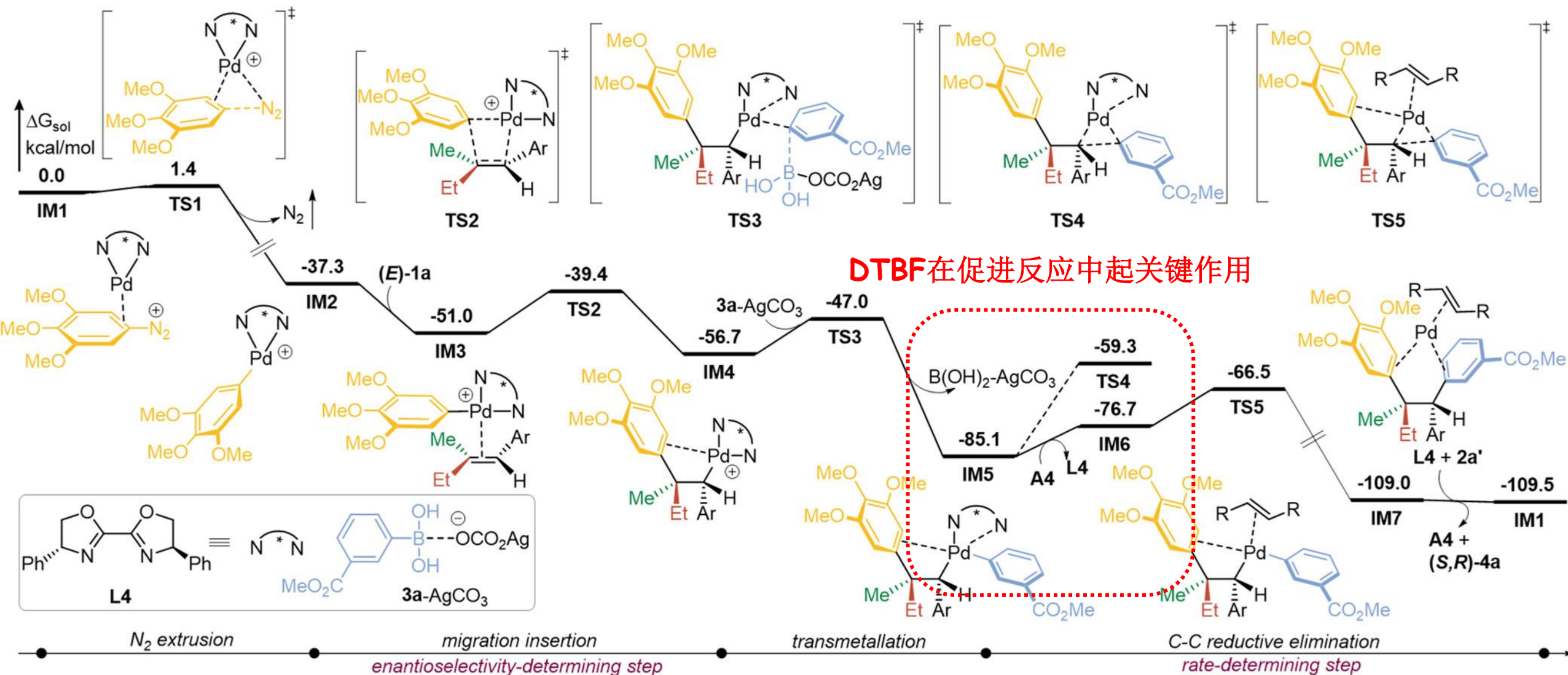


Ar = 4-OMeC₆H₄, Ar³ = 4-MeOC₆H₄



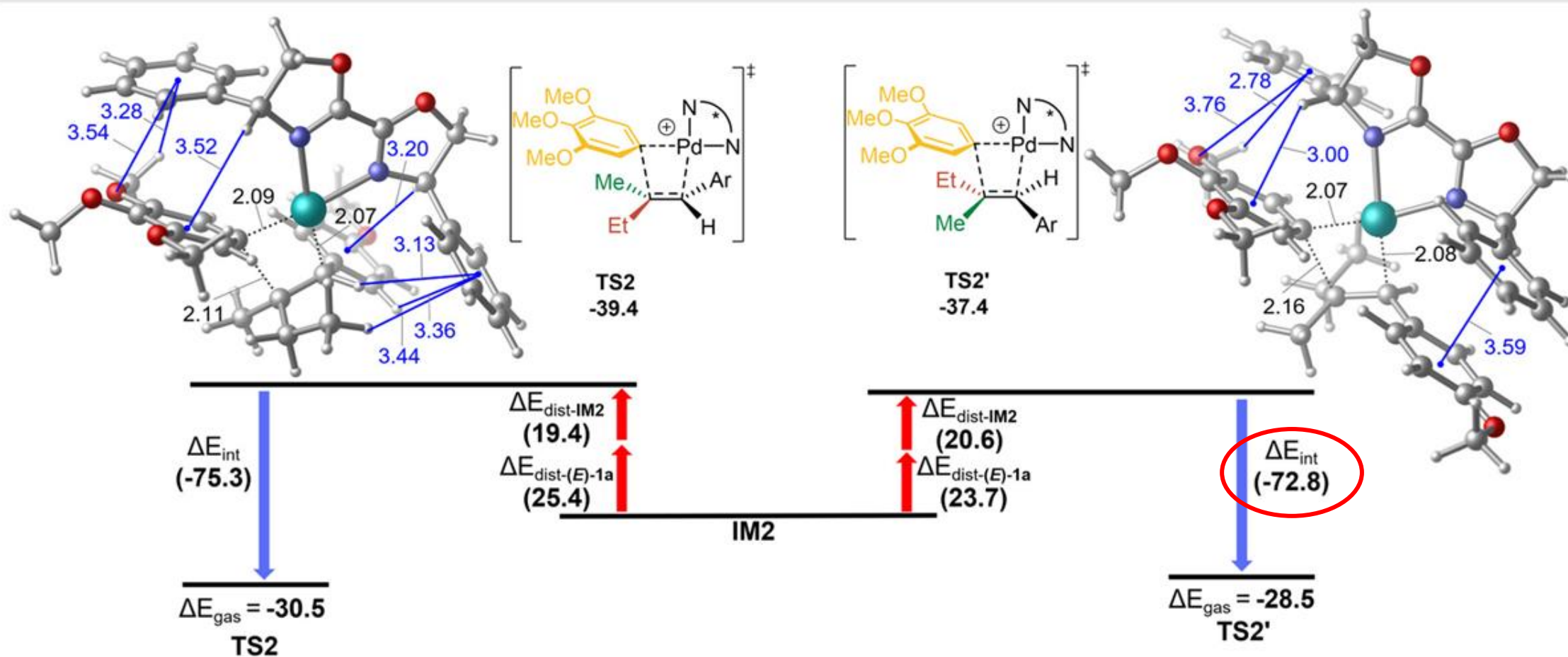
4. Catalytic Intermolecular Cross-Coupling of Olefins

a



4. Catalytic Intermolecular Cross-Coupling of Olefins

b



芳香环上的甲氧基与 (R) -PhBiOx 的苯基之间存在更有效的孤对电子 $p \cdots \pi$ 相互作用

三取代烯烃 (E) -1a 和 (R) -PhBiOx 之间的 $C-H \cdots \pi$ 相互作用



Thanks for you listening!